सीनेट की उन्नतीसवीं बैठक का कार्यवृत्त Minutes of the 29th Meeting of the Senate

5TH MAY 2009



भारतीय प्रौद्योगिकी संस्थान रूड़की रइकी – 247 667 (भारत)

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE ROORKEE- 247 667 (INDIA)



Lt Col A K Srivastava (Retd)
Registrar

भारतीय प्रौद्योगिकी संस्थान रुड़की

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No. IITR/MS/29th Senate/266/

Dated: 4th June 2009

ALL MEMBERS OF THE SENATE Indian Institute of Technology Roorkee

Subject: Minutes of the 29th Meeting of the Senate held on 5th May 2009.

Enclosed herewith please find a copy of the Minutes of the 29th Meeting of the Senate held on 5th May 2009 in the Senate Hall of the Institute, for your perusal. Your comments, if any, on the minutes may please be sent within 15 days.

Encl: As above

(A.K. Srivastava)

Lt. Col. (Retd)

Registrar & Secretary, Senate

INDIAN INSTITUTE OF TECHNOLOGY, ROORKEE ROORKEE-247 667 (INDIA)



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Minutes of the $29^{\rm th}$ Meeting of the Senate held on $5^{\rm th}$ May 2009 in the Senate Hall of the Institute.

The following were present: -

1110	ionowing were present.		
1.	Prof. S.C. Saxena	Director -C	Chairman
2.	Prof. H.K. Verma	Dy. Director	
3.	Prof. S.Y. Kulkarni	(Architecture & Plannii	ng)
4.	Prof. G.S. Randhawa	(Biotechnology)	
5.	Prof. R.P. Singh	(Biotechnology)	•
6.	Prof. Shri Chand	(Chemical Engineering)	
7.	Prof. I.D. Mall	(Chemical Engineering)	
8.	Prof. G. Bhattacharjee	(Chemistry)	
9.	Prof. R.N. Goyal	(Chemistry)	
10.	Prof. Ravi Bhushan	(Chemistry)	
11.	Prof. Kamaluddin	(Chemistry)	•
12.	Prof. V.K. Gupta	(Chemistry)	
13.	Prof. Anil Kumar	(Chemistry)	
14.	Prof. (Mrs.) Mala Nath	(Chemistry)	
15.	Prof. U.P. Singh	(Chemistry)	
16.	Prof. M.R. Maurya	(Chemistry)	
17.	Prof. G.L. Asawa	(Civil Engineering)	
18.	<u> </u>	(Civil Engineering)	
19.	•	(Civil Engineering)	
20.	Prof. V.K. Gupta	(Civil Engineering)	
21.	Prof. Deepak Kashyap	(Civil Engineering)	
22.	Prof. (Mrs) Renu Bhargava	(Civil Engineering)	
23.	Prof. U.C. Kothyari	(Civil Engineering)	
24.	Prof. P.K. Garg	(Civil Engineering)	
25.	Prof. Pradeep Bhargava	(Civil Engineering)	
26.	Prof. S.K. Ghosh	(Civil Engineering)	
27.	Prof. Mahendra Singh	(Civil Engineering)	
28.	Prof. Manoj K. Arora	(Civil Engineering)	
29.	2	(Civil Engineering)	
30.	Prof. N.K. Samadhiya	(Civil Engineering)	
31.	Prof. D.K. Paul	(Earthquake Engineering	
32.	Prof. Ashwini Kumar	(Earthquake Engineering	ıg)
33.	Prof. H. Sinvhal	(Earth Sciences)	
34.	Prof. V.N. Singh	(Earth Sciences)	
35.	Prof. A.K. Awasthi	(Earth Sciences)	
36.	Prof. H.O. Gupta	(Electrical Engineering)	
37.	Prof. Vinod Kumar	(Electrical Engineering)	
38.	Prof. Pramod Agarwal	(Electrical Engineering)	
39.	Prof. D.K. Mehra	(Electronics & Compute	
40.	Prof. S.N. Sinha	(Electronics & Compute	er Engg.)
41.	Prof. N.K. Goel	(Hydrology)	
42.	Prof. Himanshu Joshi	(Hydrology)	
43.	Prof. Pashupati Jha	(Humanities & Social S	
44.	Prof. Sukh Pal Singh	(Humanities & Social S	ciences)
45.	Prof. J.S. Upadhyay	(Paper Technology)	



46.	Prof. V.K. Nangia	(Management Studies)
47.	Prof. G.S. Srivastava	(Mathematics)
48.	Prof. S.P. Sharna	(Mathematics)
49.	Prof. T.R. Gulati	(Mathematics)
50.	Prof. (Mrs) Rama Bhargava	(Mathematics)
51.	Prof. R.C. Mittal	(Mathematics)
52.	Prof. Satish C. Sharma	(Mechanical & Industrial Engg.)
53.	Prof. Dinesh Kumar	(Mechanical & Industrial Engg.)
54.	Prof. P.K. Jain	(Mechanical & Industrial Engg.)
55 <i>.</i>	Prof. B.K. Gandhi	(Mechanical & Industrial Engg.)
56.	Prof. P.S. Mishra	(Metallurgical & Materials Engg.)
57.	Prof. Satya Prakash	(Metallurgical & Materials Engg.)
58.	Prof. R.D. Agarwal	(Metallurgical & Materials Engg.)
59.	Prof. P.K. Ghosh	(Metallurgical & Materials Engg.)
60.	Prof. (Mrs) Vijaya Agarwala	(Metallurgical & Materials Engg.)
61.	Prof. S.K. Nath	(Metallurgical & Materials Engg.)
62.	Prof. R.C. Agarwala	(Metallurgical & Materials Engg.)
63.	Prof. B.D. Indu	(Physics)
64.	Prof. Ravindra Nath	(Physics)
65.	Prof. A.K. Jain	(Physics)
66.	Prof. Rajesh Srivastava	(Physics)
67.	Prof. Vir Singh	(Physics)
68.	Prof. D. Das	(Water Resource Dev. & Management)
69.	Prof. U.C. Chaube	(Water Resource Dev. & Management)
70.	Prof. Ram Pal Singh	(Water Resource Dev. & Management)
71.	Prof. M.L. Kansal	(Water Resource Dev. & Management)
72.	Prof. Prem K.Kalra	IIT Kanpur
73.	Prof. Karmeshu	JNU, New Delhi
74.	Dr. Arun Kumar	Head, AHEC
75.	Mr. Yogendra Singh	Librarian, Central Library
76.	Dr. (Ms) Bina Gupta	Chief Warden, Kasturba Bhawan
77.	Dr. Ankush Mittal	Associate Professor, E & CE Deptt.
78.	Dr. Sandeep Singh	Associate Professor, Earth Sciences
79.	Dr. A.K. Sen	Associate Professor, Earth Sciences
80.	Lt.Col. (Retd.) A.K. Srivastava	Registrar - Secretary

The Chairman (Director) welcomed the members to the 29th Meeting of the Senate especially Prof. Prem K. Kalra, IIT Kanpur, who was attending a meeting of the Senate for the first time, and Prof. Karmeshu, JNU, New Delhi.

Before taking up the agenda, the Senate thanked the undermentioned outgoing member and recorded its appreciation for his valuable contribution in the meetings of the Senate.

Dr. G.S. Agarwal, Associate Professor Department of Mathematics & Ex-Chief Warden, Govind Bhawan



The Senate also welcomed the under- mentioned new member and hoped for her valuable contribution and active participation in its functioning:

Dr. (Ms.) Bina Gupta, Chief Warden, Kasturba Bhawan & Associate Professor, Department of Chemistry.

The Senate noted communications received from the following members for not attending the current meeting:

- 1. Prof. Sushil, I IT Delhi
- 2. Prof. S.P. Gupta, Department of Electrical Engineering
- 3. Prof. S.S. Jain, Department of Civil Engineering
- 4. Prof. A.K. Saraf, Department of Earth Sciences
- 5. Prof. R.P. Maheshwari, Department of Electrical Engineering

The Agenda was then taken up:

Item No.29.1.1: To confirm the minutes of the 28th meeting of the Senate held on 2nd March 2009.

The minutes of the 28th meeting of the Senate held on 2nd March 2009 were confirmed.

Item No.29.1.2: To receive a report on the actions taken to implement the decisions taken by the Senate in its 28th meeting held on 2nd March 2009.

The Senate noted that the required actions have been taken on the decisions taken by the Senate in its 28th meeting held on 2nd March 2009.

Item No.29.2.1: To consider the syllabi of Integrated M.Sc. Physics Courses of III year onwards.

As considered and recommended by the Board of Studies, the Senate, after minor modifications, approved the syllabi of Integrated M.Sc. Physics Courses of III year onwards. The detailed syllabi of the courses are appended at **Appendix 'A'**.

Item No.29.2.2: To consider the syllabi of B.Arch. I year courses of the structure applicable from 2009-10 session.

As considered and recommended by the Board of Studies, the Senate, after minor modifications, approved the syllabi of the following courses of B.Arch. I year for the structure applicable from 2009-10 sessions:



AR-101	Basic Design and Model Making
AR-103	Architectural Graphics-I
AR-102	Architectural Design-I
AR-104	Architectural Graphics-II
AR-106	Introduction to Building Materials and
	Construction-I
AR-108	Principles of Architectural
AR-110	Colour Fundamentals and its
	Applications
	AR-103 AR-102 AR-104 AR-106 AR-108

The syllabi of the above courses are appended at **Appendix 'B'**.

Item No.29.2.3: To consider the syllabus of a new Departmental Elective proposed by the Department of Mathematics.

As considered and recommended by the Board of Studies, the Senate decided that the Departmental Elective MA-905: Evolutionary Algorithms, as appended at **Appendix 'C'** be approved.

Item No.29.2.4: To consider the syllabus of a new Institute Elective proposed by the Department of Metallurgical and Materials Engineering.

As considered and recommended by the Board of Studies, the Senate decided that the Institute Elective IMT-01: Non Destructive Testing, as appended at **Appendix 'D'** be approved.

Item No.29.2.5: To consider the minimum CGPA required for OBC category students to be eligible for a change of branch.

As considered and recommended by the Board of Studies, the Senate decided that a minimum CGPA of 7.0 will be required for OBC category students for a change of their branch.

Item No.29.2.6: To consider the syllabi of III and IV year of B.Tech. (Metallurgical & Materials Engineering) programme.

As considered and recommended by the Board of Studies, the Senate, after minor modifications, approved the syllabi of III and IV year of B.Tech. (Metallurgical & Materials Engineering) Programme. The syllabi of all these courses are appended at **Appendix 'E'**.



Item No.29.2.7: To consider the Grievance Redressal System for Students.

As considered and recommended by the Board of Studies, the Senate decided that the under-mentioned Standing Committee be constituted to address the grievances of any student.

1. Dean, Academic Studies

Chairman

- 2. Coordinator, I year
- 3. Associate Dean, Discipline
- 4. One Senate Nominee
- 5. Chairman, DAC (Concerned Department)

Item No.29.2.8: To consider the syllabi of new Institute Electives proposed by the Department of Electronics & Computer Engineering.

The Senate decided that the new institute Electives proposed by the Department of Electronics & Computer Engineering be referred in to the Department for consideration and placed in the next meeting of the Senate.

Item No.29.2.9: To consider the syllabi of new Institute Electives proposed by the Alternate Hydro Energy Centre.

As considered and recommended by the Board of Studies, the Senate decided that the syllabi of the following new Institute Electives be approved:

IAH-01: Small Hydropower Development

IAH-02: Renewable Energy Sources Development Technology

The syllabi of these Institute Electives are appended at **Appendix 'F'**.

Item No.29.2.10: To consider the Academic Calendar for the Autumn Semester 2009-2010 Session.

The Senate decided that the Academic Calendar for the Autumn Semester 2009-2010 Session as appended at **Appendix 'G'** be approved.



Item No.29.2.11: To consider the panel of Senate's Nominees on the Selection Committee for academic positions.

After due deliberation and certain some modifications, the Senate decided that the panel of the Senate's Nominees on the Selection Committees of the academic positions in respect of each of the following Departments/ Centres be approved (Appendix 'H'):

- 1. Alternate Hydro Energy Centre
- 2. Architecture & Planning
- 3. Biotechnology
- 4. Chemical Engineering
- 5. Chemistry
- 6. Civil Engineering
- 7. Paper Technology
- 8. Earth Sciences
- 9. Earthquake Engineering
- 10. Electrical Engineering
- 11. Electronics & Computer Engg.
- 12. Humanities & Social Sciences
- 13. Hydrology
- 14. Management Studies
- 15. Mathematics
- 16. Mechanical & Industrial Engineering
- 17. Metallurgical & Materials Engg.
- 18. Physics
- 19. Water Resources Development & Management

ItemNo.29.2.12: To consider the issue related to the curriculum structure of Post B.Sc. programmes.

The Senate decided that the recommendations of the committee in respect of curriculum structure of Post B.Sc. Programme as appended at **Appendix 'I'** be approved.

Item No. 29.3.1: To report that the Director has approved the acceptance of a proposal for institution of the "Dwarka Dass Balwant Kaur Thapar Cash Prize" of Rs.5,000/- p.a. for student of M.Tech. II year (AHES), having second highest CGPA, admitted on self finance basis.

Senate noted that the Director, on behalf of the Senate has approved the proposal of Prof. O.D. Thapar for institution of the "Dwarka Dass Balwant Kaur Thapar Cash Prize" of Rs.5,000/-. The donors have contributed a sum of Rs.1,25,000/- for this purpose.

The meeting ended with a vote of thanks to the Chair.



PROGRAM CODE 30 DEPARTMENT PH YEAR III

7

Integrated M.Sc. (Physics) Physics Department

AUTUMN SEMESTER

	Teaching Scheme (Hrs./Week)		е	Dur	am. ation rs.)	Relative Weightage (%)								
S. NO.	SUBJECT	COURSE TITLE	SUBJECT AREA	CREDITS	ב	F	ል	Theory	Practical	CWS	PRS	MTE	ETE	PRE
1.	PH-301	Plasma Physics	DCC	3	3	0	0	3	0	15		35	50	T
2.	PH-303	Quantum Physics	DCC	4	3	1	0	3		25		25	50	-
3.	PH-305	Properties of Matter and Acoustics	DCC	4	3_	1	0	3	-	25	-	25	50	
4.	PH-307	Atomic Physics	DCC	4	3	1	0	3	-	25		25	50	-
7	PH-IE1	Institute Elective I	HSS MC	3	2	1	0	2	-	25	-	25	50	-
		Sub Total		18	14	4	0							

			SPRING SEMESTE	R i					•					
1	PH-302	Laboratory – I	DC	3	0	0_	6		6	: -	50	-	-	50
2.	PH-304	Elements of Condensed Matter Physics	,DC	4	3	1	0	3	-	25	-	25	50	-
3.	PH-306	Elements of Classical Mechanics	DC	4	_3	1	0	3	-	-25	-	25	50	-
4.	PH-308	Nuclear Physics and its Applications	DC	. 4	3	1	0	3	-	25	-	25	50	-
5.	PH-IE2	Institute Elective II	HSS MC	3	2	1	0	2	-	25	-	25	50	
			-	18	11	4	6						· · · · · ·	C.
		Sub Total												

PROGRAM CODE 30 DEPARTMENT PH

YEAR

PH IV Integrated M.Sc. (Physics)
Physics Department

AUTUMN SEMESTER

			γ	Teaching Scheme (Hrs./Week)		Dur	am. ation rs.)	F	lelative	Weigl	htage (%	6)		
S. NO.	SUBJECT	COURSE TITLE	SUBJECT	CREDITS	J	F	d.	Theory	Practical	CWS	PRS	MTE	ETE	PRE
	PH-501	Semiconductor Devices	Dec					·		·	·			<u> </u>
$\frac{1}{2}$.	PH-503	Quantum Mechanics – I	DCC	3 4	3	0	0	3	0	15	0	35	50	0
3.	PH-505	Mathematical Physics	DCC DCC	3	3		0	3	0	15	0	. 35	50	0
4.	PH-507	Classical Electrodynamics	DCC	4	3	0	0	3	0	15	0_	35	50	0_
-1. – 5.	PH-509	Classical Mechanics	DCC	3	3	0	0	3	0	25	0	25	50	0
6.	PH-511	Computational Physics	DCC	3	1	0	4	0	2	15 15	0 15	35	50	0
do	112-311	Sub Total		20	16	0	4			1 -3	13	30	0	40
		SPRING SEN	IESTE	R										
1.	PH-502	Laboratory - Il	DCC	3	0	0	6	0	4	0	50	0	0	50
2.	PH-504	Condensed Matter Physics	DCC	3	3	0	0	3	0	15	0	35	50	0
3.	PH-506	Statistical Mechanics	DCC	3	3	0	()	3	0	15	0	35 .	50	0
4.	PH-508	Quantum Mechanics - II	DCC	3	3	0	0	3	U	15	0	35	50	0
5.	PH-510	Nuclear and Particle Physics	DCC	2	2	0	Ü	2	0	15	0	35	50	0
6.	PH-512	Physics of Earth's Atmosphere	DCC	2	2	0	0	2	0	15	0	3.5	50	0
7.	PH-514	Molecular Spectroscopy and Lasers	DCC	2	2	0	0	2	0	15	0	35	50	0
		Sub Total	-	18	15	0	. 6	 						<u></u>

200

PROGRAM CODE 30 DEPARTMENT PH YEAR V Integrated M.Sc. (Physics)
Physics Department

AUTUMN SEMESTER

				Teaching Scheme (Hrs./Week)		Dur	am. ation rs.)	Relative Weightage			ntage (%	6)		
S. NO.	SUBJECT	COURSE TITLE	SUBJECT AREA	CREDITS	٦	<u> </u>	ď	Theory	Practical	CWS	PRS	MTE	ETE	PRE
1	PH-ELE1	Departmental Elective – Group A	DEC	4] 3	1 1	0	3	0	25	1 0	1 25	50	
2.	PH-ELE2	Departmental Elective – Group B	DEC	3	10	0	6	0	1 3	0	50	25	50 0	50
3.	PH-ELE3	Departmental Elective - Group C	DEC	3	3	0	0	3	0	15	0	35	50	- 30
4.	PH-ELE4	Departmental Elective - Group C	DEC	3	3	0	0	3	0	15	0	35	50	0
5.	PH-ELE5	Departmental Elective - Group C	DEC	3	3	0	0	3	0	15	0	35	50	0
6.	PH-IE3	Institute Elective - III	ESEC	4	3	1	0	: 3	0	25	0	25	50	0
9		Sub Total		20	15	2	6							
		SPRING	SEMESTE	R			-							
1	PH- ELE6	Departmental Elective – Group D	DEC	4	3	1	0	3	0	25.	0	25	50	()
2.	PH-608	Seminar	DCC	4	0	0	0	0	0	0	0	0	0	0
3.	PH-610	Dissertation	DCC	14	0	0	0	0	0	0	0	0	0	()
				22	3	1	0							
		Sub Total	•				1	1	1.	}				

Group A	
PH-601	Advanced Condensed Matter Physics
PH-603	Advanced Atmospheric Physics
PH-605	Advanced Laser Physics
PH-607	Advanced Nuclear Physics
	•
Group B	
PH- 609	Experiments in Condensed Matter Physics
PH- 611	Experiments in Atmospheric Physics
PH- 613	Experiments in Laser Physics
PH- 615	Experiments in Nuclear Physics
Group C	
РН- 617	Advanced Characterization Techniques
PH- 619	Introduction to Quantum Field Theory
PH- 621	Fiber and Nonlinear Optics
PH- 623	General Relativity
PH- 625	Particle Physics
PH- 627	Quantum Theory of Solids
PH- 629	Weather Forecasting
Group D	
- A .	
PH- 602	Nuclear Astrophysics
PH- 604	Physics of Nanosystems
PH- 606	Superfluidity and Superconductivity

NAME OF DEPTT./CENTRE:

DEPARTMENT OF PHYSICS

1. Subject Code: PH-301

Course Title: Plasma Physics

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.):

Theory

3

Practical^{*}

4. Relative Weightage: CWS

15

PRS MTE

PRE

0

5. Credits:

3

6. Semester: Autumn

7. Subject Area: DCC

8. Pre-requisite:

PH-202M

9. Objective:

To familiarize students with the basic principles of plasma physics

with application to other areas.

S. No.	Contents	Contact Hours
1.	Introduction to Plasma Physics: Plasma definition; Debye shielding; Plasma parameters; Criteria for Plasma	6
2.	Motion of Particles in the Presence of Electric and Magnetic Fields: Motion of charge particle in uniform and non-uniform E and B fields; Time varying E and B fields.	6
3.	Plasma as Fluids: Relation of Plasma Physics with ordinary Electromagnetics; The Fluid Equation of Motion; Fluid drifts perpendicular and parallel to B; The Plasma Approximation	8
4.	Waves in Plasma: Representation of waves; Plasma oscillations; Electron plasma waves, ion waves; Validity of plasma approximation; Comparison of ion and electron waves, Electrostatic electron oscillations perpendicular to E and B; Electrostatic ion waves perpendicular to B	10
5.	Diffusion and Resistivity: Diffusion and mobility in weakly ionized gases; Decay of a Plasma by Diffusion; Steady state solution, recombination; Diffusion across a magnetic field; Collision in a fully ionized plasma; The single fluid MHD equation; Diffusion in fully ionized plasma	12
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Chen F F, "Introduction to Plasma Physics", Plenum Press New York	1990
2.	Davidson R C, "Physics of Non-Neutral Plasmas", Allied Publishers Pvt. Ltd	2001
3.	Eliezer S and Eliger Y, "The Fourth State of Matter: An Introduction to Plasma Science", CRC Press	2001
4.	Paul M. B., "Fundamentals of Plasma Physics", Cambridge University Press	2004

NAME OF DEPTT./CENTRE: **DEPARTMENT OF PHYSICS** 1. Subject Code: PH-303 Course Title: Quantum Physics 2. Contact Hours: L: 3 T: 1 P: 0 3. Examination Duration (Hrs.): Theory Practical 3 4. Relative Weightage: CWS PRS MTE ETE PRE 50 5. Credits: 6. Semester: Autumn 7. Subject Area: DCC

8. Pre-requisite:

PH-101

9. Objective:

To familiarize students with the basic principles of quantum

mechanics and applying these to single-particle systems.

S. No.	Contents	Contact Hours
1.	Wave Packets and Uncertainty Principle: Plane waves; Superposition of plane waves; Wave packets; Fourier analysis; Group velocity; Propagation of wave packets; Wave packet broadening; Gaussian wave packet.	8
2.	Schrödinger Equation: The wave equation and the interpretation of ψ; Operators and expectation values of dynamical variables; Commutators and operator algebra; Stationery states; Dirac notations.	10
3.	Problems in One-dimension: Potential step, rectangular potential barrier, symmetries and invariance properties, reflection and transmission coefficients, potential well, Kroning-Penny Model.	12
4.	Harmonic Oscillator: Energy eigen values and eigen functions of a 1-D harmonic oscillator; Matrix formulation of oscillator problem, N-Harmonic oscillators in contact with a heat bath of temperature T; Boltzmann factor, average energy of a harmonic oscillator at temperature T; Bose-Einstein and Fermi-Dirac distribution.	12
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
. 1.	Gariorowicz S, "Quantum Physics", 3rd Ed, John Wiley & Sons	2006
2.	Mathews P.M and Venkatesan K., "A Text Book of Quantum Mechanics", Tata McGraw Hill	2000
3.	Beiser A, "Concepts of Modern Physics", McGraw Hill International	2004
4.	Ghatak A and Lokanathan S, "Quantum Mechanics", Mcmillan India Ltd.	2004

NAME OF DEPTT./CENTRE:	DEPARTMEN	T OF PHYSICS	en e
1. Subject Code: PH-305	Course Title: Pro	operties of Matter a	nd Acoustics
2. Contact Hours: L: 3	T: 1	P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical	0
4. Relative Weightage: CWS 25	PRS 0 MT	E 25 ETE 50	PRE 0
5. Credits: 4 6. Sem	ester: Autumn	7. Subject Area: D	cc

8. Pre-requisite:

PH-101

9. Objective:

To familiarize students with fundamentals of properties of matter,

waves and acoustics.

S. No.	Contents	Contact Hours
1.	Elasticity: Hooke's Law Stress; Strain Diagram; Elastic moduli; Relation between elastic constants; Poisson's Ratio; Work done in stretching and twisting a wire; Twisting couple on a cylinder; Rigidity modulus by static torsion; Torsional pendulum; Rigidity modulus and moment of inertia.	8
2.	Bending of Beams: Cantilever, Cantilever oscillations; Expression for time period; Experiments to find Young's modulus; Non uniform bending.	6
3.	Fluids : Surface Tension; Excess of pressure over curved surfaces - Application to spherical and cylindrical drops and bubbles - Variation of Surface tension with temperature - Jaegar's method.	6
4.	Viscosity: Co-efficient of viscosity; Rate of flow of liquid through a capillary tube; Poiseuilles' formula; Experiment to determine co-efficient of viscosity of a liquid; Variation of viscosity of a liquid with temperature; Applications of viscosity.	6
5.	Waves and Oscillations: Simple harmonic motion; Free, Damped; Forced vibrations and Resonance; Fourier's Theorem - Application to saw tooth wave and square wave, Intensity & loudness of sound; Decibels - Intensity levels; Noise pollution.	8
6.	Ultrasonics: Production of ultrasonic waves; Piezoelectric crystal method; Magnetostriction, Application of ultrasonics to science, industry and medicine.	8
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Feynman R P, Leighton R B and Sands M, "The Feynman	2005
2.	Lectures on Physics", Vols. I, Narosa Chakrabarthy P K, "Mechanics and General Properties of	2001
٠. ا	Matter", Allied Publishers Pvt. Ltd	2001
3.	Flowers B H and Mendoza E, "Properties of Matter", Wiley Publisher	1991
4.	Bajaj N K, "The Physics of Waves and Oscillations", Tata MC Graw Hill	1988
5.	Ingard K U, "Fundamentals of Waves and Oscillations", Cambridge Univ. Press	1988

NAME OF DEPTT./CEN	TRE: DEPARTM	TENT OF F	PHYSICS		
1. Subject Code: PH-30	7 Course Title:	Atomic Ph	ysics		-5
2. Contact Hours: L:	3 T: 1		P: 0		
3. Examination Duration	(Hrs.): Theory	3 F	Practical	0	
4. Relative Weightage: 0	CWS 25 PRS 0	MTE 25	ETE 50	PRE	0
5. Credits: 4	6. Semester: Autum	n 7. S	ubject Area: I	occ	
8. Pre-requisite: PH	-101				
9. Objective: To	introduce basic princip	les of spectro	scopy to the	students	i
40 Details of Course				·	

Sl.No	Contents	Contact Hour
1.	Basic Principles of Spectroscopy; The Optical Spectrum of the Hydrogen Atom; Bohr's Postulates; Motion of the Nucleus; Spectra of Hydrogen-like Atom; Excitation of Quantum Jumps by Collisions; Sommerfeld's model; Lifting of Orbital Degeneracy by the Relativistic Mass Change.	9
2.	Magnetic Moment of the Orbital Motion; Precession and Orientation in a Magnetic Field; Stern and Gerlach experiment; Calculation of Spin-Orbit Splitting in the Bohr Model; Level Scheme of the Alkali Atoms; Fine Structure in the Hydrogen Atoms;	9
3.	The Lamb Shift. Directional Quantisation in a Magnetic Field; Electron Spin Resonance; The Zeeman Effect Experiments; Explanation of the Zeeman Effect from the Standpoint of Classical Electron Theory; Description of the Ordinary Zeeman Effect by the Vector Model; The Anomalous Zeeman Effect; Magnetic Moments with Spin-Orbit Coupling; The Paschen-Back Effect.	8
4.	The Spectrum of the Helium Atoms; Electron Repulsion and the Pauli Principle; Angular Momentum Coupling, Coupling Mechanism; LS Coupling (Russell-Saunders Coupling); jj Coupling, Magnetic Moments of Many-Electron Atoms; Multiple Excitations.	8
5.	X-Radiation from Outer Shells; X-Ray Bremsstrahlung Spectra; Emission Line Spectra: Characteristic Radiation; Absorption Spectra; The Auger Effect; Nuclear Spin; Hyperfine Structure	8
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1.	Hanken H and Wolf H C, "The Physics of Atoms and Quanta", 6th Ed., Springer	2007
2.	Laud B B, "Non-linear Optics", Wiley Eastern Ltd.	1992
3.	Beiser A, "Concept of Modern Physics", 6th Ed., Tata McGraw Hill	2002

NAME OF DEPTT/CENTRE:	DEPARTMENT	OF PHYSICS	, et et
1. Subject Code: PH-302	Course Title: Labor	ratory -l	·
2. Contact Hours: L: 0	T: 0	P: 6	 12 ¹ Fe 1)
3. Examination Duration (Hrs.):	Theory 0	Practical 6	
4. Relative Weightage: CWS 0	PRS 50 MTE	0 ETE 0 PR	E 50
5. Credits: 3 6. Sem	ester: Autumn	7. Subject Area: DCC	
8. Pre-requisite: PH-101	•		

9. Objective: To familiarize students with the basic experiments in properties of matter and acoustics.

S. No.	Particulars	Contact Hours
1.	To determine the Young's modulus of steel, aluminum and brass by method of flexure	
2.	Determination of shear modulus of steel, copper and brass	
3.	Determination of surface tension of olive oil at different temperatures	14
4.	To measure the dynamic viscosity of water and methanol at different temperatures	14 x 6
5.	To study free oscillations and forced oscillations under damped ands un-damped conditions	
6.	Determination of wavelength and frequency of sound wave by Quincke's method	
7.	Determination of phase and group velocity of ultrasonic wave in different liquids	

8.	Study of stationary ultrasonic wave and determination of its	N
	wavelength	ė as
9.	Study of interference of ultrasonic wave with Michelson's interferometer	
10.	Determination of the value of g (acceleration due to gravity) using Kater's type compound pendulum	·
11.	Coefficient of viscosity of water b rotating disc/cylinder method	
12.	Determination of surface tension of a liquid at different temperatures by Jaeger's method	
13.	Coefficient of viscosity of water by Poiseuilles's method	1 a
14.	Determination of modulus of elasticity of different materials	
	Total	84

S. No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1.	Chattopadhayay D and Rakshit P.C., "An advanced course in Practical Physics" 7 th Ed., New Central Book agency (P) Ltd.	2005
2.	G. L. Squires "Practical Physics" 4th Ed., Cambridge University Press	2001

NAME OF DEPTT./CENTI	E: DEPARTMENT OF PHYSICS		
1. Subject Code: PH-304	Course Title: Elements of	f Condensed Matter Physics	
2. Contact Hours: L: 3	T: 1	P: 0	
3. Examination Duration (H	rs.): Theory 3	Practical 0	
4. Relative Weightage: CW	VS 25 PRS 0 MTE	25 ETE 50 PRE 0	
5. Credits: 4	6. Semester: Spring	7. Subject Area: DCC	

8. Pre-requisite: PH-202M

9. Objective:

To familiarize students with bonding, mechanical properties, crystal structure, lattice vibrations, defects in solids and theory of ferromagnetism.

S. No.	Contents	Contact Hours
1.	Bonding and Mechanical Properties: Covalent bonding, ionic	6
	bonding, metallic bonding, hydrogen bonding and Van der waals	1
	bonding. Elastic constants and elastic waves.	<u> </u>
2.	Crystal Structure: Point symmetry, translational symmetry, two-	9
	and three- dimensional lattices, simple crystal structures, Miller	
	indices, diffraction from periodic structures, reciprocal lattice,	}
	Brillouin zones.	
3.	Lattice Vibrations: One dimensional lattices (monoatomic and	7
	diatomic), quantization of elastic waves, phonon momentum, density	
[of modes.	
4.	Electrons in Solids: Free electron gas in metals, periodic potential	8
}	and Bloch's theorem and Kronig-Penney model.	ŀ
5.	Defects in Solids: Lattice vacancies, diffusion, colour centers and	4
	elementary idea of dislocation.	
6.	Magnetism: Langevin theory of dia- and para- magnetism, quantum	8
	theory of dia- and para- magnetism, magnetic ordering, Weiss	
	molecular field theory of ferromagnetism and Neel theory of anti-	
ļ	ferromagnetism.	
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Kittel C., "Introduction to Solid State Physics", 8 th Ed., Wiley Eastern Ltd	2004
2.	Ashcroft N W and Mermin N D, "Solid State Physics", 2 nd Ed. Holt-Saunders	2000
3.	Hook JR and Hall HE, "Solid State Physics", John Wiley	2001

NAME OF DEPTT./CENTRE: DEPARTMENT OF PHYSICS

1. Subject Code: PH-306 Course Title: Elements of Classical Mechanics

2. Contact Hours: L: 3 T: 1 P: 0

3. Examination Duration (Hrs.): Theory 3 Practical 0

4. Relative Weightage: CWS 25 PRS 0 MTE 25 ETE 50 PRE 0

5. Credits: 4 6. Semester: Spring 7. Subject Area: DCC

8. Pre-requisite: None

9. Objective: To familiarize students with the basics of classical mechanics

S. No.	Contents	Contact Hours
1.	Central Force: Equations of motion, equivalent one body problem,	10
	orbits, Virial theorem, Kepler's problem, scattering theory, centre of	
	mass and laboratory frames of reference	
2.	Rigid Body Motion: Orthogonal transformation, transformation matrix, Euler angles, Cayley-Klein parameters, Euler's theorem, Finite & infinitesimal rotations; Rotating frames of reference, Coriolis' force; Angular momentum and kinetic energy, dyadic & tensors; Moment of inertia, principal axis transformation, Euler equation of motion and its solutions, tops, precession, satellite orbits.	15
3.	Small Oscillations : Eigenvalue problem, normal coordinates, frequencies of vibrations, forced vibrations, examples.	. 4
4.	Introduction to Constrained Motions: Principle of virtual work, generalized coordinates, introduction to Lagrange's equation of motion, generalized momenta, cyclic coordinates, Legendre's dual transformation, Hamilton's function and Hamilton's equation of motion; Configuration space, phase space and state space.	13
	- Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Goldstein H, "Classical Mechanics", Narosa	2001
2.	Rana N.C. and Joag P.S, "Classical Mechanics", Tata McGraw Hill	1994
3.	Gupta K.C., "Classical Mechanics of Particles and Rigid Bodies", Wiley Eastern	2001

NAME OF DEPTT./CENTRE:

DEPARTMENT OF PHYSICS

1. Subject Code: PH-308

Course Title: Nuclear Physics and its Applications

2. Contact Hours:

L: 3

T: 1

3

P: 0_f

3. Examination Duration (Hrs.):

Theory

Practical

4. Relative Weightage: CWS 25

PRS

MTE 25 ETE 50

PRE

0

5. Credits:

6. Semester: Spring

7. Subject Area: DCC

8. Pre-requisite: None

9. Objective: To familiarize students with the basic concepts of nuclear physics and its industrial, analytical, medicinal and energy applications.

S. No.	Contents	Contact Hours
1.	Nuclear shape, size, radii, matter/charge distributions; Nuclear force; Concept of isospin; Charge independence of nuclear forces in the light of isospin. Mass defect and binding energy; Liquid drop model; Semi empirical mass formula; Evidence of shell structure; Shell model with harmonic oscillator and spin-orbit potential and its predictions.	9
2.	α -decay, its properties, range, range-energy relationship, Geiger-Nuttal law, theory of α -decay, β -decay and its classifications (only basics), γ -decay: range, properties, pair production, energy spectra and nuclear energy levels.	8
3.	Nuclear reaction, Kinematics, Direct nuclear reaction, Compound nuclear reaction, Nuclear fission and fusion.	7
4.	Gas, Scintillation and Semiconductor detectors. Neutron detectors, Accelerators: Cyclotron and Linac.	9
5.	Industrial, analytical and medicinal applications; Power from fission, Nuclear reactors; Source of stellar energy	9
	Total	42

S. No.	Name of Authors/Books/Publishers	Year of Publication/Reprint
1.	Lilley J S, "Nuclear Physics", John Wiley & Sons	2001
2.	Ghoshal S.N., "Nuclear Physics", S. Chand & Comp. Ltd	2000
3.	Povh B, Rith K, Scholz C and Zetsch F, "Particles and Nuclei", 2 nd Ed. Springer	1999
4.	Heyde K, "From Nucleons to the Atomic Nucleus", Springer	1998

DEPARTMENT OF PHYSICS NAME OF DEPTT./CENTRE: 1. Subject Code: PH-501 Course Title: Semiconductor Devices 2. Contact Hours: P: 0 L: 3 T: 0 3. Examination Duration (Hrs.): Theory Practical 3 0 4. Relative Weightage: CWS MTE 50 PRE PRS 35 ETE 5. Credits: 7. Subject Area: DCC 6. Semester: Autumn

8. Pre-requisite:

EC-102 or equivalent

9. Objective:

To introduce the physics of semiconductors, p-n junction, bipolar

junction transistors, FET and MOSFET.

S. No.	Contents	Contact Hours
1.	Semiconductors: Energy bands, direct and indirect semiconductors, charge carriers, mobility, drift of carriers in field, Diamond and Zinc-Blende structure, bonds and bands in semiconductors, intrinsic and extrinsic semiconductors, law of mass action, Hall effect and cyclotron resonance in semiconductors.	12
2.	Optical Injection: Carrier life time, direct and indirect recombination of electron and holes, steady state carrier generation, diffusion and drift of carriers, the continuity equation, steady state carrier injection, The Haynes-Shockley experiment.	8
3.	Junctions: Metal-Semiconductor contact: under equilibrium, and non-equilibrium conditions, the junction diode theory, tunnel diode, photodiode, LED, solar cell, Hetro-junctions and Laser diode.	10
4.	Bipolar Junction Transistors: Charge transport and amplification, minority carrier distribution and terminal currents switching behaviour in bipolar transistor.	4
5.	FET and MOSFET: Ideal MOS capacitor, effect of work function and interface charge on threshold voltage.	6
6.	Gunn Diode: Transferred electron mechanism and drift of space charge domain.	2
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Streetman B G and Banerjee S "Solid State Electronic Devices", 6 th Ed. Prentice Hall	2005
2.	Sze S M, "Semiconductor Devices Physics and Technology" 2 nd Ed. John Wiley & Sons	2003
3.	Tyagi M S, "Semiconductor Materials and Devices", John Wiley & Sons	2000

NAME OF DEPTT/CENTRE:	VIRE: DEPARTMENT OF PHYSICS		
1. Subject Code: PH-503	Course Title: Quan	tum Mechanics-I	
2. Contact Hours: L: 3	T: 1,	. P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical 0	
4. Relative Weightage: CWS 25	PRS 0 MTE	25 ETE 50 PRE (0
5. Credits: 4 6. Sem	ester: Autumn	7. Subject Area: DCC	-

8. Pre-requisite: PH-303 or equivalent

9. Objective: To apply quantum mechanics to the dynamics of single particle in one-, two-and three- dimensional potential fields.

S. No.	Contents	Contact Hours
1.	Introduction: Postulates of Quantum Mechanics and meaning of measurement, Operators and their expectation values, Schrodinger	6
	equation, Particle in a box, Orthogonality of eigen functions, Dirac rotations, Hilbert space.	
2.	Matrix Formulation: Matrix formulation of 1-dimensional harmonic oscillator problem; creation and annihilation operators; Equation of motion and classical correspondence, Heisenberg equation of motion, Schrodinger, Heisenberg and Interaction picture, Motion in a one-dimensional periodic potential, Kroning-penny model.	8
3.	Motion in a Central Potential: Angular momentum operator, expressions of L ² and L _z , eigen values and eigen functions of L ² and L _z , hydrogen atom, solution of radial equation, energy eigen values, eigen functions of H atom, orthogonality of eigen functions, rigid rotator, matrix representation L ² , L _x , L _y , L _z , generalized angular momentum, generator of rotation and their commutation relations, spin – ½ matrices, coupling of angular momenta, Clebsch-Gordon Coefficients.	10

4.	Scattering Theory: Scattering amplitude, differential and total	10
	cross-section, scattering by a central potential, method of partial waves, phase-shift analysis, optical theorem, scattering by a square-well potential, integral equation, the Born approximation.	
5.	Approximate Methods: WKB approximation, WKB expansion, connecting formulas, variational principle and its application to Helium atom and hydrogen molecule	8
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Schiff L.I., "Quantum Mechanics", 3 rd Ed, McGraw Hill Book Co.	1990
2.	Merzbacher E, "Quantum Mechanics", 2 nd Ed., John Wiley & Sons	1996
3.	Gasiorowicz S, "Quantum Physics", John Wiley	2000
4.	Mathews P. M. and Venkatesan K, "A Text Book of Quantum Mechanics", Tata McGraw Hill	2000

NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS	
1. Subject Code: PH-505	Course Title: Math	ematical Physics	•
2. Contact Hours: L: 3	T: 0	P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical	0
4. Relative Weightage: CWS 15	PRS 0 MTE	35 ETE 50	PRE 0
5. Credits: 3 6. Sem	ester: Autumn	7. Subject Area: DO	ÇC

8. Pre-requisite: MA-101 or equivalent, MA-202M or equivalent

9. Objective: To familiarize the students with the standard techniques in modern mathematical physics

S. No.	Contents	Contact Hours
1.	Complex variables and applications, analytic functions, contour integration, residue calculus, conformal mapping and its applications. Fourier and Laplace transforms, evaluation of integral transforms and their inverses using contour integrals.	6
2.	Special equations of Mathematical Physics; Legendre and associated Legendre equations; Hermite equation; Laguerre and associated Laguerre equations; Bessel's equation; Hypergeometric equation; Beta and gamma functions.	8.
3.	Green's functions and solutions to inhomogeneous differential equations and applications.	8
4.	Covariant and Contravariant tensors, covariant derivatives, affine connections Christoffel symbols, Curvature tensor.	6
5.	Classification and examples of (finite) groups, homomorphisms, isomorphisms, representation theory for finite groups, reducible and irreducible representations, Schur's Lemma and orthogonality theorem,	. 8
6.	Characters; Lie Groups and Lie algebra; Vector Spaces; Hilbert Space and operators	6
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Arfken G. B. and Weber H. J., "Mathematical Methods for Physicists", 5 th Ed. Academic Press.	2005
2.	Whittaker E.T.and Watson E.W., "A Course of Modern Analysis", Cambridge University Press	2008
3.	Hammermesh M., "Group Theory and Applications to Physical Problems", Dover publications, NY.	1989
4.	Akhiezer N. I. and Glazman I. M., "Theory of Linear Operator in Hilbert Space", Dover Publications	1993

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NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS	
1. Subject Code: PH-507	Course Title: Clas	sical Electrodynamics	
2. Contact Hours: L: 3	T: 1	P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical 0	v s
4. Relative Weightage: CWS 25	PRS 0 MTE	25 ETE 50 PRE	0
5. Credits: 4 6. Sem	ester: Autumn	7. Subject Area: DCC	
0 D	* . 1 . 4		

8. Pre-requisite: PH-202M or equivalent

9. Objective: To emphasize electric and magnetic phenomena and introduce the covariant formulation of Maxwell's theory of electromagnetism

S. No.	Contents	Contact Hours
1.	Maxwell's Equation: Maxwell's equation, vector and scalar potentials, Gauge transformation, Poynting theorem., plane electromagnetic waves, waves in non-conducting and conducting medium; Linear and Circular polarization, reflection and refraction.	12
2.	Covariant Formulation of Vacuum Electrodynamics: Space-Time symmetry of the field equations; Covariant formulation; Four-vector potential; Electromagnetic field tensor and its invariants; Lorentz-Force equation in a covariant form.	12
3.	Radiation from Accelerated Charges: Retarded potentials; Lienard-Wiechert potentials; Fields produced by a charge in uniform and arbitrary motion, radiated power; Angular and frequency distribution of radiation, radiation from charged particle with colinear velocity and acceleration; Synchrotron radiation; Thomson scattering; Cherenkov radiation.	14
4.	Multipole Fields: Inhomogeneous wave equation, multipole expansion of electromagnetic fields, angular distribution, multipole moments.	. 4
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Jakson J D, "Classical Electrodynamics", John Wiley	2002
2.	Griffiths D J, "Introduction to Electrodynamics", Prentice Hall	1999
3.	Capri A.Z. and Panat P.V., "Introduction to Electrodynamics" Narosa Publication House	2002
4.	Franklin J., "Classical Electromagnetism", Pearson Education	2007

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NAME OF DEPTT/CENTRE:	DEPARTMENT	OF PHYSICS	
I. Subject Code: PH-509	Course Title: Class	sical Mechanics	٠.
2. Contact Hours: L: 3	T: 0		. 11. fal
3. Examination Duration (Hrs.):	Theory 3	Practical 0	
4. Relative Weightage: CWS 15	PRS 0 MTE	35 ETE 50 PRE	0
5. Credits: 3 6. Sem	ester: Autumn	7. Subject Area: DCC	

- 8. Pre-requisite: PH-306 or equivalent
- 9. Objective: To familiarize students with the various methods of solving problems in classical mechanics using the techniques of Lagrange, Hamilton, Hamilton-Jacobi and Poisson Brackets.

S. No.	Contents	Contact Hours
1.	Lagrange's Equation: Constraints; D'Alembert's principle and Lagrange's equation of motion, dissipation function, Hamilton's principle, calculus of variations, nonholonomic systems, conservation laws, relativistic and covariant formulation.	10
2.	Hamilton's Equations: Hamilton's equation of motion, cyclic coordinates, Routh's procedure, relativistic formation, variational principle, principle of least action.	. 8
3.	Canonical Transformations: Equations of canonical transformations and examples, sympletic approach, Poisson brackets and equation of motion, conservation laws, angular momentum, symmetry groups & Louville's theorem.	8
4.	Hamilton-Jacobi Theory: Hamilton-Jacobi equation's of motion, harmonic oscillations, separation of variables, action-angle variables, Kepler problem, geometrical optics and wave mechanics.	8
5.	Canonical Perturbation Theory: Time-dependent perturbation, examples, time-independent theory in first order and higher orders, applications to celestial and space mechanics, Adiabatic invariants.	8
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Goldstein H, "Classical Mechanics", Narosa	2001
2.	Rana W.C. and Jog P.S, "Classical Mechanics", Tata McGraw Hill	1991
3.	Gupta K.C., "Classical Mechanics of particles and Rigid Bodies", Wiley Eastern	2001

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NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS	:
1. Subject Code: PH-511	Course Title: Com	putational Physics	
2. Contact Hours: L: 2		P: 2	
3. Examination Duration (Hrs.):	Theory 2	Practical 2	
4. Relative Weightage: CWS	10 PRS 15 MTE	15 ETE 30 PRE	30
5. Credits: 3 6.	Semester: Autumn	7. Subject Area: DCC	
8. Pre-requisite: MA-202	M or equivalent		

9. Objective: To provide the knowledge of computation with a suitable mathematical software and its applications to solve the problems of Physics.

S. No.	Particulars Particulars	Contact Hours
1.	Introduction to mathematical software: Need and advantages of numerical computation in physics, programming in a suitable mathematical software (Matlab/Mathematica/Scilab/ Octave), input/output, interactive input, loading and saving data, loops branches and control flow. Matrices and Vectors, Matrix and array operations, eigenvalues and eigen vectors.	8
2.	Sub programs: Advantages of modular programming, built-in functions, scripts, functions, sharing of variables between modules.	3
3.	Graphics: 2D plots, style options, axis control, overlay plots, subplot, histogram, 3D plots, mesh and surface plots, contour plots.	4
4.	Numerical computation: Computer programs for: solving linear system of simultaneous equations, nonlinear algebraic equation, roots of polynomials, curve fitting, polynomial curve fitting, least square curve fitting, interpolation, data analysis and statistics, numerical integration, Monte-Carlo simulation, ordinary differential equation, first order and second order ODEs, event location.	13
	Total	28

List o	f Experiments	
1. 2. 3. 4. 5. 6. 7. 8.	Black body radiation (computation and graphical representation) Reflection and transmission of an electromagnetic wave Statistical distributions at different temperatures Binding energy curve for nuclei using liquid drop model Eigen-value problem: 1-D square potential well Eigen-values and wave-functions of a simple harmonic oscillator Monte-Carlo simulation Linear/Projectile motion (simulation and solutions)	<u>*</u>
	Total	28

S.No.	Name of Authors / Books / Publisher	Year of Publication/ Reprint
1.	Pratap R, "Getting started with MATLAB 7", Oxford Univ. Press	2006
2.	Gilat A, "Matlab: An introduction with applications", Wiley	2008
3.	Eaton J W, Batchman D and Hauberg S "GNU Octave Manual Version 3", Network Theory Ltd.	2008
4.	Campbell S, Chancelier J P and Nikoukhah R, "Modeling and simulation in Scilab", Springer	2005
5.	Wolfram S, "The Mathematica Book," 5th Ed., Wolfram Media	2003
6.	Gerald C F and Wheatley P O, "Applied Numerical Analysis", 7th Ed, Addison Wesley	2003

NAME OF DEPTT,/CENTRE:	DEPARTMENT	OF PHYSICS
1. Subject Code: PH-502	Course Title; Labor	ratory-II
2. Contact Hours: L: 0	T: 0	P; 6
3. Examination Duration (Hrs.):	Theory 0	Practical 4
4. Relative Weightage: CWS 0	PRS 50 MTE	0 ETE 0 PRE 50
5. Credits: 3 6. Sem	ester: Spring	7. Subject Area: DCC

8. Pre-requisite: I

PH-304, PH-308 or equivalent

9. Objective: To familiarize with the basic experiments in Solid State Physics, Nuclear Physics, Laser Physics and Atmospheric Physics.

S. No.	Contents	Contact Hours
1	Study of Hall effect and to determine the Hall coefficient	
2	To measure resistivity of semiconductor by Four Probe method and determination of band gap.	
3	To determine reverse saturation current, material constant and band gap of PN Junction	}
4	To ascertain of the Random nature of nuclear radiation	
5	To study G.M. tube characteristics and to calculate the dead time,	14 x 6
·6	To determine the relative beta counting of two strong β-sources of nuclear radiation and to determine the absorption coefficients,	
7	To determine the distribution of the size of Aerosol.	
8	To measure the attenuation of laser radiation in varying atmospheric condition.	
9	The measurement of precipitation rate of water using rain gauge.	
10	To determine the numerical aperture of a given multimode fiber using the far field measurements.	

11	To measure the spot size and the angle of divergence of a laser beam, to produce the elliptically and circularly polarized light from	
12	an unpolarized laser beam and study their angular intensity profiles. Design of counter using JK flip flop and a relaxation oscillator with given frequency and duty cycle	
13.	Design a Schmitt trigger with given UTP LTP and hysteresis	
14.	To design a binary/BCD up-down counter using IC 74190/74191	:
 	Total	84

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Nakra B.C. &. Chaudhery K.K, "Instrumentation Measurements & Analysis", Tata McGraw Hill	2002
2,	Sayer M. & Mansingh A., "Measurement, Instrumentation & Experiment Design in Physics and Engineering", Prentice Hall India	2000
3.	Melissinos A.C. and Napolitano J, "Experiments in Modern Physics", Academic Press	2000
4.	W.R. Runyan, "Semiconductor Measurements and Instrumentation", McGraw Hill	2002

NAME OF DEPTT/CENTRE:	DEPARTMENT	OF PHYSICS	
1. Subject Code: PH-504	Course Title: Conc	lensed Matter Physics	
2. Contact Hours: L: 3	Ť: 0	P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical 0	
4. Relative Weightage: CWS 15	PRS 0 MTE	35 ETE 50 PRE	0
5. Credits: 3 6. Sem	ester: Spring	7. Subject Area: DCC	1 .
8. Pre-requisite: PH-304 or equ	uivalent		

9. Objective: To familiarize with the structural and electronic properties of crystalline and non-crystalline materials and their dynamical properties.

S. No.	Contents	Contact Hours
1.	Crystalline Materials: Scattering of x-ray, neutrons and electrons from solids; Atomic scattering factor; Lattice planes and Miller indices.	6
2.	Lattice Dynamics: Harmonic and adiabatic approximations; Lattice vibrations of three dimensional crystals; Periodic boundary conditions; Normal modes. Quantization of lattice vibrations; Lattice heat capacity (Einstein and Debye theories) anharmonicity of thermal expansion.	9
3.	Electronic Energy Bands: Resume of free-electron model; Fermi energy; Fermi surface and electronic heat capacity, electrical and thermal conductivity, nearly free electron model; Periodic potential and Bloch theorem, extended and reduced zone scheme, tight binding model.	9
4.	Superconductivity: Experimental evidence (Meissner effect, heat capacity, energy gap, microwave properties and isotope effect), Thermodynamics of superconductors; London equation; Elementary BCS theory.	9

5.	Non-crystalline Materials: Non-crystalline solids - diffraction	9
	pattern and radial distribution function, Elementary idea of glass	
	transition, Quasi crystals, Liquid crystals - idea of orientational	.
	order and Landau theory of isotropic-nematic phase transition,	
	Physics of Polymers.	
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Taylor P. L. and Heinonen O., "A Quantum Approach to Condensed Matter Physics", Cambridge University Press	2004
2.	Ashcroft N W and Mermin N D, "Solid State Physics", Holt-Saunders	2000
3.	Chaikin P M and Lubensky T C, "Principles of Condensed Matter Physics", Cambridge University Press	2000
4.	Hamley I. W., "An Introduction to Soft Matter: Polymers, Colloids, Amphiphiles and Liquids" John Wiley	2000

NAME OF DEPTT/CENTRE:	DEPARTM	ENI OF PHIS	0103
1. Subject Code: PH-506	Course Title:	Statistical Mec	nanics
2. Contact Hours: L: 3	T: 0	P:	0
	e to the contract of the		
3. Examination Duration (Hrs.):	Theory	3 Praction	cal 0
4. Relative Weightage: CWS 15	PRS 0	MTE 35 ETE	50 PRE 0
5. Credits: 3 6. Sem	nester: Spring	7. Subject	Area: DCC
8. Pre-requisite: PH-503 & PH	I-509 or equiv	alent	
•			and the second s

To understand the macroscopic behaviour of the classical and

10. Details of Course:

9. Objective:

S.No.	Contents	Contact Hours
1.	Classical Statistical Mechanics: Macro and microstates, connection between statistics and thermodynamics, phase space; Liouville's Theorem. Microcanonical, canonical and grand canonical ensembles; Energy and Density fluctuations; equivalence of various ensembles. Equipartition and virial theorem, partition function; Derivation of thermodynamic properties; some examples including (i) classical ideal gas (ii) system of classical harmonic oscillators, (iii) system of magnetic dipoles in magnetic field.	10
2.	Quantum Statistical Mechanics: Quantum mechanical ensembles theory, the density matrix and partition function with examples including (i) an electron in a magnetic field (ii) a free particle in a box (iii) a linear harmonic oscillator. Symmetric and Antisymmetric Wavefunctions. Microcanonical ensemble of ideal Bose, Fermi and Boltzmann gases, derivation of Bose, Fermi and Boltzmann statistics; Grand Partition function of ideal Bose and Fermi gases; Statistics of the occupation.	12
3.	Ideal Bose and Fermi Systems: Thermodynamic behaviour of an ideal Bose gas; Bose condensation; Liquid Helium; Blackbody radiation and Planck's law of radiation; Thermodynamic behaviour of an ideal Fermi gas; Electrons in metals, specific heat and Pauli susceptibility of electron gas.	10

quantum thermodynamic systems.

4.	Phase Transitions and Critical Phenomenon: Order parameter, 1st and Ilnd order phase transitions. Ising model in zeroth and first approximation. Critical exponents, thermodynamic inequalities, Landau theory of phase transitions.	10
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication/
1.	Patharia R K "Statistical Mechanics" (2 nd Ed.), Pergaman press Huang K "Statistical Mechanics" (2 nd Ed., 2 nd reprint), John Wiley & Sons	Reprint 2001 2003
3.	Landau L.D. and Lifshitz E M "Statistical Mechanics", Butteworth-Heinemaun	1998
4.	McQuarrie D A "Statistical Mechanics", Harper & Row	2003

NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS	•
 Subject Code: PH-508 Contact Hours: L: 3 	Course Title: Quar	ntum Mechanics P: 0	-1 1
3. Examination Duration (Hrs.):	Theory 3	Practical	0
4. Relative Weightage: CWS 15	PRS 0 MTE	35 ETE 50	PRE 0
5. Credits: 3 6. Sem	ester: Spring	7. Subject Area:	DCC
8. Pre-requisite: PH-503 or equ	uivalent		

9. Objective:

To introduce various time-independent and time-dependent perturbation methods and relativistic quantum mechanics.

S. No.	Contents	Contact Hours
14.	Time-independent Perturbation Theory: Non-degenerates and degenerate perturbation theory, its application to Stark effect, Zeeman effect, spin-orbit coupling fine structure and to anharmonic oscillator.	10
2.	Time-dependent Perturbation Theory: Transition probability, harmonic perturbation, Fermi-golden rule, semi-classical theory of radiation, spontaneous and stimulated emission cross-section.	10
3.	Identical Particles: Indistinguishability, permutation symmetry, two-particle system; Helium atoms, simple idea of Hartee self-consistent field, Hartee-Fock theory.	10
4.	Relativistic Quantum Mechanics: Klein-Gordon equation and its applications, Dirac theory of electron, spin of the electron, solution of Dirac equation for free particles, hole (positron)-Dirac equation for Hydrogen atom.	12
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Schiff L I, "Quantum Mechanics", 3 rd Ed, McGraw Hill Book Co.	1990
2.	Merzbacher E, "Quantum Mechanics", 2 nd Ed., John Wiley & Sons	1997
3.	Bjorken J D and Drell S D, "Relativistic Quantum Mechanics", McGraw Hill Book Co.	1998

NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS	•
1. Subject Code: PH-510	Course Title: Nuc	ear and Particle	Physics
2. Contact Hours: L: 2	T: 0	P: 0	
3. Examination Duration (Hrs.):	Theory 2	Practical	0
4. Relative Weightage: CWS 15	PRS 0 MTE	35 ETE 50	PRE 0
5. Credits: 2 6. Sem	ester: Spring	7. Subject Area: I	DCC
8. Pre-requisite: PH-308 or eq	uivalent		

9. Objective: To introduce the foundations of nuclear and particle physics.

S. No.	Contents	Contact Hours
1.	Binding Energy and Mass Formula; Nature of nuclear force; Two nucleon problem: Ground state of deuteron, its magnetic moment and quadrupole moment; Tensor nature of nuclear force; Gamma decay selection rules; Rotational spectra in deformed nuclei; Nuclear reaction mechanisms; Compound nuclear reaction	8
2.	Concept of isospin, Charge independence of nuclear force in the light of isospin; mirror nuclei; estimate of decay rates from isospin conservation.	4
3.	Strangeness, Lepton and other quantum numbers, conservation of these quantum numbers related to strong and weak reactions, Strong interaction, salient features of Quantum chromodynamics, structure of nucleon, quark model, concept of colour.	5
4.	SU(2) and SU(3) of isospin symmetry and its generators, preliminary idea of Lie algebra, SU(3) flavour symmetry and construction of meson octet, Baryon octet & decuplet and their wave functions.	5
5.	β -decay and its classifications, neutrino hypothesis, energy spectrum of β -decay, Fermi theory of β -decay, concept of parity, helicity, non-conservation of parity in β -decay and its experimental verifications, Standard model of particle physics.	6
	Total	28

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Perkins, D.H., "Introduction to High Energy Physics", Addison-Wesley Publishing company.	2004
2.	Griffith D, "Introduction of Elementary Particles", John Wiley	2005
3.	Burcham W E and Jobes M, "Nuclear & Particle Physics", Addison-Wesley	2002
4.	Ghoshal, S.N., "Nuclear Physics", S. Chand and Company	2000

38

NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS	
1. Subject Code: PH-512	Course Title: Phys	ics of Earth's At	mosphere
2. Contact Hours: L: 2	T: 0	P: 0	e de la companya de l
3. Examination Duration (Hrs.):	Theory 2	Practical	0
4. Relative Weightage: CWS	15 PRS 0 MTE	35 ETE 50	PRE 0
5. Credits: 2 6. Se	emester: Spring	7. Subject Area: [occ

8. Pre-requisite:

None

9. Objective: To introduce the basics of atmospheric physics.

S. No.	Contents		
1.	Atmospheric Evolution: Solar radiation, present atmospheric constituents, evolution of the atmosphere, formation of ozone.	6	
2.	Lower Atmosphere: Variation of temperature, density, ionization and pressure with altitude, hydrostatic equation, green house effect, lapse rate and stability criteria, cloud formation and precipitation.	8	
3.	Upper Atmosphere: Chapman theory of layer production, formation of ionosphere, photochemistry of the thermosphere, electron, ion and neutral temperatures in the thermosphere, airglow and auroral emissions.	8	
4.	Weather: weather and climate, weather modification, artificial rain making, cloud suppression, storms.	6	
	Total	28	

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Seeds M.A., "Solar System", Brooks/Cole Thomson Learning	2007
2.	Houghton J.T. "Physics of Atmosphere", Cambridge Univ. Press	2002
3.	Rogers R R, "A Short Course in Cloud Physics", Pergamon Press	1989

NAME OF DEPIT./CENTRE:	DEPARTMEN	I OF PHYSICS	
1. Subject Code: PH-514	Course Title: Mo	lecular Spectroscopy a	nd Lasers
2. Contact Hours: L: 2	T: 0	P: 0	-
3. Examination Duration (Hrs.):	Theory 2	Practical	0
4. Relative Weightage: CWS 15	PRS 0 MT	E 35 ETE 50	PRE 0
5. Credits: 2 6. Sem	ester: Spring	7. Subject Area: [occ
8. Pre-requisite: PH-307 or eq	uivalent		
9. Objective: To introduce molecula	r spectroscopy and	the basics of lasers.	

S.No.	Contents	Contact Hours
1.	Types of molecular energy states and molecular spectra, pure rotational spectra; Vibration-Rotational spectra, Raman Spectra; Electronic spectra,	7
2.	Frank-Condon principle; Isotope effect on electronic spectra; Fluorescence and Phosphorecence; Classification of molecular electronic states. Idea of Line broadening mechanisms	7
3.	Lasers-Physical principles; Threshold condition; Generation of population Inversion; Optical Resonator; Gaussian beam; Single mode lasers; He-Ne, CO ₂ and Semiconductor Lasers.	7
4.	Laser absorption Spectroscopy; Saturated absorption spectroscopy; Doppler- free two photon spectroscopy; Level crossing spectroscopy. Laser cooling and trapping of atoms	7
	Total	28

S.No.	Name of Authors/ Books/Publishers	Year of Publication /Reprint
1.	Hanken H and Wolf H C, "The Physics of Atoms and Quanta", (6th edition), Springer	2007
2.	Herzberg G, "Molecular Spectra and Molecular Structure of Diatomic Molecules", Van Nostrand Reinhold	1989
3.	Bransden B H, Joachain C J, "Physics of Atoms and Molecules" (2nd Edition) Pearson Education	2003
4.	Metcalf H J, "Laser Cooling and Trapping", Peter van der Straten, Springer	2001

NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS
1. Subject Code: PH-601	Course Title: Adv	anced Condensed Matter Physics
2. Contact Hours: L: 3	T: 1	P: 0
3. Examination Duration (Hrs.):	Theory 3	Practical 0
4. Relative Weightage: CWS 2	5 PRS 0 MTE	25 ETE 50 PRE 0
5. Credits: 4 6. Sep	mester: Autumn	7. Subject Area: DEC
8. Pre-requisite: PH-504		

9. Objective: To introduce the general aspects of phase transition, electronic transport phenomena, superconductivity, dielectric, optical and magnetic properties of solids.

S. No.	Contents	Contact Hours
1.	Surface and Interfaces: Work function and contact potential; Thermoionic emission; Low-energy electron diffraction; Electronic surface levels; Super lattices; Quantum wells; Quantum wires, Quantum dots and carbon Nanotubes.	9
2.	Magnetism: Magnetic properties of insulators, Langevin diamagnetism and Van Vleck paramagnetism, Curie paramagnets and Curie-Weiss ferromagnets, Neel Antiferromagnets, Heisenberg model; Spin Waves, Ising model; Elements of magnetic properties of metals, Landau diamagnetism, Pauli paramagnetism, Stoner ferromagnetism; Magnetic resonance; NMR and EPR.	9
3.	Transport Properties: Boltzmann equation; Relaxation time approximation; General transport coefficients; Electronic conduction in metals; Thermoelectric effects; Transport phenomena in magnetic field; Magnetoresistance; Hall effect and Quantum Hall effect.	8
4.	Phase Transitions: Order parameter; Critical points; First and second order phase transitions; Mean field theory; Properties near critical point; Landau theory; Bragg-Williams theory; Liquid-gas	8

	transition and Isotropic-mematic transition.	
5.	Superconductivity: Cooper pairing and BCS theory; Ginzburg- Landau theory; Flux quantization; Supercurrent tunneling; DC and AC Josephson effects; High-Tc superconductors.	8
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Kittel C, "Introduction to Solid State Physics", 6 th Ed. Wiley eastern Ltd	2004
2.	Ashcroft N W and Mermin N D, "Solid State Physics", Holt-Saunders	2000
3.	Chaikin P M and Lubensky T C, "Principles of Condensed Matter Physics", Cambridge University Press	1995
4.	Harrison P, "Quantum Wells, Wires and Dots", Wiley & Sons Ltd.	2005

NAME OF DEPTT./CENTRE:	DEPARTMENT OF PHYSICS		
1. Subject Code: PH-603	Course Title: Adva	nced Atmospheric Physics	
2. Contact Hours: L: 3	T: 1	P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical 0	
4. Relative Weightage: CWS 25	PRS 0 MTE	25 ETE 50 PRE 0	
5. Credits: 4 6. Seme	ester: Autumn	7. Subject Area: DEC	
8. Pre-requisite: PH-512			

9. Objective: To provide the knowledge of advances in atmospheric physics.

S. No.	Contents	Contact Hours
1.	Atmospheric Dynamics: Apparent forces, effective gravity, coriolis	12
	force, pressure gradient force, gradient wind, thermal wind,	
	continuity equation, atmospheric waves	
2.	Cloud Physics: Aerosols, CCN, IN, cloud formation, cloud	10
<u> </u>	electrification, lightning, sprites, global electric circuits.	
3.	Ionosphere: Formation of Ionosphere, ionospheric exploration,	10
] [langmuir probe, temperature measurements, airglow and aurora,	}
	radio wave propagation in the ionosphere.	
4.	Magnetosphere: Solar wind and its interaction with the	10
	magnetosphere, frozen-in magnetic field radio wave propagation	}
}	through the magnetosphere, whistlers, satellite observations of	
)	various plasma domains and plasma instabilities.	1
	Total	42

11. Suggested Books:

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Vallace J and Hobbs, P V, "Atmospheric Science", Academic Press	1997
2.	Rees M H, "Physics & Chemistry of Upper Atmosphere", Cambridge Univ. Press	1989
3.	Ratcliffe J A, "An Introduction to the Ionosphere & Magnetosphere, Cambridge Univ. Press	1972
4.	Smithson P, "Fundamentals of Physical Environment", Ken Addison and Attrinson,	2008
5.	Rogers R R, "A short course in Cloud Physics", Pergamon Press	1989

NAME OF DEPTT./CENTRE:	DEPARTME	INT OF PHYSICS	3
1. Subject Code: PH-605	Course Title:	Advanced Laser Ph	ysics
2. Contact Hours: L: 3	T: 1	P: 0	
3. Examination Duration (Hrs.):	Theory 3	Practical	0
4. Relative Weightage: CWS 25	PRS 0	ATE 25 ETE 50	PRE 0
5. Credits: 4 6. Sen	nester: Autumn	7. Subject Area	: DEC

8. Pre-requisite:

PH-514

9. Objective: To introduce the concept of laser physics and its applications.

S. No.	Contents	Contact Hours
1.	Quantum theory for the evaluation of the transition rates and Einstein's coefficients, interaction of matter with radiation having broad spectrum, interaction of near monochromatic radiation with an atom having broad frequency response.	6
2.	Line broadening mechanisms, homogeneous and inhomogeneous broadening, natural collision and Doppler broadening mechanisms and line shape functions.	4
3.	Laser rate equations, the three levels and four levels system, variation of power around threshold, optimum output coupling, quality factor, the ultimate line width of the laser.	5
4.	Optical resonators, modes of a rectangular cavity and open planar resonators, confocal resonator system, modes of a confocal resonator using Huygen's principle, planar resonators, Fox and Li theory.	6
5.	Pulsed lasers, Q-switching techniques, active and passive shutters, mode-locking, various techniques for mode-locking of a laser.	5
6.	Mechanism and applications of Ar-ion, CO ₂ , Nd:YAG, Ti:Sapphire, Dye, Excimer and free electron lasers.	5
7.	Semiconductor lasers, p-n junction diode lasers, hetrojunction lasers.	5
8.	Modulation techniques for laser light, electro-optic and acousto-optic modulation, electro-optic effect in KDP crystal, longitudinal and transverse modes, acousto-optic effect, Raman-Nath and Bragg diffraction, small and large angle Bragg diffraction.	6
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Laud B B, "Lasers and Nonlinear Optics", Wiley Eastern Ltd.	1992
2.	Ghatak A K and Thyagarajan K., "Optical Electronics", Cambridge University Press	2003
3.	Yariv A, "Quantum Electronics", John Wiley & Sons	1989
4.	Thyagarajan K. and Ghatak A. "Lasers: Theory and Applications", Macmillan	1997
5.	Yariv A, "optical Electronics", Oxford University Press	1997

NAME OF DEPTT./CENTRE:	DEPARTME	ENT OF PHYSICS	
1. Subject Code: PH-607	Course Title:	Advanced Nuclear Physics	
2. Contact Hours: L: 3	T: 1	P: 0	
3. Examination Duration (Hrs.): Th	neory 3	Practical 0	. • •
4. Relative Weightage: CWS 25	PRS 0 N	MITE 25 ETE 50 PRE 0	
5. Credits: 4 6. Sen	nester: Autumn	7. Subject Area: DEC	
8. Pre-requisite: PH 510 or equivale	ent		

9. Objective: To introduce the advanced concepts of nuclear physics.

S. No.	Contents	Contact Hours
1.	Yukawa theory of nuclear forces, Deuteron problem and tensor forces, n-p, p-p scattering and partial wave theory, effective range theory.	6
2.	Shell Model and its predictions: magnetic moments of nuclei and Schmidt lines, quadrupole moments; Even-even, odd-even, odd-odd nuclei, pairing interaction; Many-body basis states, Hartree-Fock single-particle Hamiltonian, selection of shell model space and effective Hamiltonian.	8
3.	Deformed nuclei and their shapes; Colletive model Hamiltonian, vibrational and rotational spectra, Nilsson model. High spin phenomena (back bending), superdeformation, octopole deformation Giant dipole resonances.	7
4.	Kinematics of nuclear reaction, reciprocity theorem, compound nuclear reaction, direct reaction and derivation of the crossections in these processes; Statistical theory of nuclear reaction and concept of nuclear temperature and entropy	7
5.	Shape-elastic, compound elastic scattering and dispersion relations, Electromagnetic transitions in nuclei, multipole expansion of the electromagnetic field; Transition probability in semiclassical treatment, Weisskopf estimate.	7
6.	Angular correlation studies; Lifetime measurements; Detection of gamma rays; Hp-Ge and other detectors; Gamma arrays.	7
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Roy R R and Nigam B P, "Nuclear Physics", John Wiley	2002
2.	Srivastava B B, "Fundamentals of Nuclear Physics", Rastogi Publications	2006
3.	Eisenberg J M and Greiner W, "Nuclear Theory", Vols. 1, North Holland	2002
4.	Eisenberg J M and Greiner W, "Nuclear Theory", Vols. 2, North Holland	2002
5.	Eisenberg J M and Greiner W, "Nuclear Theory", Vols. 3, North Holland	2002

NAME OF DEPTT/CENTRE:	DEPARTMENT OF PHYSICS
1. Subj ect Code: PH-609	Course Title: Experiments in Condensed Matter Physics
2. Contact Hours: L: 0	T: 0 P: 6
3. Exarmination Duration (Hrs.):	Theory 0 Practical 6
4. Relative Weightage: CWS 0	PRS 50 MTE 0 ETE 0 PRE 50
5. Credits: 3 6. Sem	nester: Autumn 7. Subject Area: DEC
8. Pre-requisite: PH-502	

9. Objective: To familiarize the students with the advanced experiments in Condensed Matter Physics.

S. No.	Contents	Contact Hours
1	Study of variation of resistivity of metal and highly resistive materials with temperature by Four Probe Technique.	
2	Mapping and analysis of the resistivity of large samples (thin films, superconductors etc.) by Four probe Technique.	·
3	To study the temperature dependence of Hall coefficient of N and P type semiconductors	
4	 (a) To measure the dielectric constant and Curic temperature of given ferroelectric samples. (b) To measure the coercive field (E_c), Remanent Polarization (P_r), Curie Temperature (T_c) and Spontaneous Polarization (P_s) of Barium Titanate (BaTiO₃). 	14 x 6
5	Thermoluminescence in alkali halides crystals. (a) To produce F centers in the crystal exposing to X-ray /UV source. (b) To determine activation energy of the F-centers from initial rise method.	
6	Verification of Bragg's law and determination of wavelength/energy spectrum of X-rays.	
7	Study of Solar Cell characteristics and to determine (i) Open circuit voltage V_{oc} (ii) Short circuit current I_{sc} , (iii)Efficiency η , (iv) Fill factor, (v) Spectral characteristics and (vi) Chopper characteristics.	
8	To measure the magnetoresistance of semiconductor and analyze the plots of $\Delta R/R$ and log-log plot of $\Delta R/R$ Vs magnetic field.	

	Total	84	
14	To study C-V characteristics of various solid state devices & materials. (like p-n junctions and ferroelectric capacitors		
13_	To determine the value of Lande's 'g' factor using ESR spectrometer.		
12	To measure critical current density of given superconductor and study its field dependence.		
11	To determine transition temperature of given superconducting material and study Meissner effect.	the state of the late was	
10	To study the temperature dependence of Laser diode characteristics		
· 9	To determine the coercivity, saturation magnetization and retentivity of ferromagnetic samples using Magnetic Hysteresis Loop Tracer		

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Melissinos A.C. and Napolitano J, "Experiments in Modern	2003
	Physics", Academic Press	
2.	S.M. Sze, "Semiconductor devices Physics & Tech.", Wiley	2002

NAME OF DEPTT./CENTRE:	DEPARTME	NT OF PHYSICS
1. Subject Code: PH-611 2. Contact Hours: L: 0	Course Title: T: 0	Experiments in Atmospheric Physics P: 6
3. Examination Duration (Hrs.):	Theory	0 Practical 6
4. Relative Weightage: CWS 0	PRS 50	MTE 0 ETE 0 PRE 50
5. Credits: 3 6. Sem	ester: Autumr	7. Subject Area: DEC

8. Pre-requisite: PH-502

9. Objective of Course: The lab work aims to familiarizing students with the basic experiments in Atmospheric Physics.

Sl.No	Contents	Contact Hours
1	To measure fair weather electric field and do atmospheric electric field simulation	
2	To measure the concentration of salts in the ground water and rain water using Flame Photometer	
3	To measure the rain water precipitation rate and to find rain drop size distribution using Rain Gauge:	14 x 6
4	To measure attenuation coefficient of a gas for a given wave length of electromagnetic radiation.	
5	To measure the size distribution of aerosol particles.	
6	To measure solar constant using Solarimeter and study the diuranal variation of solar flux in the visible spectrum.	
7	To measure the diuranal variation of sound noise: A case study.	Ī
8	To study and analysis of VLF generated by lightning.	
9	Study and assessment of ambient air quality using spectrophotometer.	
10	To analyze Ionosonds data and obtain electron density is the ionosphere.	
	Total	84

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	McCartney E J, "Optics of the Atmosphere", Wiley	1976
2.	Hulst H C, "Light Scattering by Small Particle", Courier Dover	1964
Ĺ <u> </u>	Pub	

NAME OF DEPTT./CENTRE:	DEPARTM	ENT OF PHYSICS	
1. Subject Code: PH-613	Course Title:	Experiments in Laser	Physics
2. Contact Hours: L: 0	T: 0	P: 6	
3. Examination Duration (Hrs.):	Theory	0 Practical	6
4. Relative Weightage: CWS 0	PRS 50	MTE 0 ETE 0	PRE 50
5. Credits: 3 6. Sem	ester: Autumr	7. Subject Area: I	DEC
8. Pre-requisite: PH-502		·	

9. Objective: The lab work aims to familiarize the students with the advanced experiments in Laser Physics Lab.

	ails of Course:	
S. No.	Contents	Contact Hours
1.	To determine the mode field diameter (MFD) of the fundamental	
	mode of a given single-mode fiber using the far field technique.	
2.	To measure the near field intensity profile of a multimode fiber and thereby its refractive index profile.	·
3.	To measure the propagation constants of a given optical waveguide	
	using the prism coupling technique.	
4.	To study electrical and optical characteristics of LED and LD.	
5.	To measure power loss at a splice between two multimode fibers and]
	study the variation of splice loss with transverse, longitudinal and angular offsets.	
6.	To study bend-induced loss in a single mode fiber.	14 x 6
7.	To study faraday effect and to measure the angle of rotation as a	14 X U
/•	function of mean flux density at different wavelengths thereby	
	evaluate Verdet's constant as a function of wavelength.	
8.	To study Kerr effect and to determine Kerr constant of a given	
0,	material.	
9.	To study fiber grating based pressure sensor.	
10.	To construct EDF ring laser and characterize it in terms of slope	
	efficiency, lasing threshold and intra-cavity loss.	
11.	To record and reconstruct holograms.	
12.	To characterizes a WDM based optical communication system in	
:	terms of insertion/return loss, isolation/extinction ratio, narrowband	
	wavelength response of WDM components and chromatic dispersion.	
13.	To construct and characterize a diode pumped Nd:YVO ₄ /Nd:YAG	
	laser and to do second harmonic generation.	

14.	To study the acousto-optic effect and determine the velocity of acoustic waves in a given medium using a laser beam	
	Total	84

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Ghatak and Shenoy, "Fiber Optics through experiments", Viva Books	1994
2.	Laud B B, "Lasers and Nonlinear Optics", Wiley Eastern Ltd.	1992
3.	Ghatak A.K., Pal, B.P., Shenoy M. R. and Khijwania S.K, "Fiber Optics through Experiment", Viva Books	2009
4.	Ghatak A. K. and Thyagrajan K., "Optical Electronics", Cambridge University Press	2003

NAME OF DEPTI/CENTRE:	DEPARTMENT	OF FRISICS	
1. Subject Code: PH-615	Course Title: Expe	eriments in Nuclear Ph	ysics
2. Contact Hours: L: 0	T: 0	P: 6	
3. Examination Duration (Hrs.):	Theory 0	Practical 6	
4. Relative Weightage: CWS	PRS 50 MTE	0 ETE 0 PRE	50
5. Credits: 3 6. Ser	mester: Autumn	7. Subject Area: DEC	
8. Pre-requisite: PH-502			
9. Objective: The lab work aims	to familiarizing stud	lents with the advanced	experiments
in Nuclear Physics.			
10. Details of Course:			

	Doutionless	Control II
S. No.	Particulars Particulars	Contact Hours
1.	To do the energy analysis of an Unknown Gamma Source by Gamma Ray	
	Spectroscopy using NaI(Tl) - Single Channel Analyzer	
	(i) Energy Calibration	
	(ii) Energy Analysis of an Unknown Gamma Source.	
[(iii) Energy Resolution.	
2.	To do Spectrum Analysis of 60Co and 137Cs by Gamma Ray Spectroscopy	
•	using NaI(TI) - Multi Channel Analyzer and study the Energy resolution	14 6
	dependence on detector size.	14 x 6
3.	To find the Mass Absorption Coefficient of lead for 662 KeV gamma ray	
4.	Alpha Spectroscopy with surface barrier detectors	
٦,	(i) Alpha spectrum and energy calibration.	
	(ii) Energy determination of an Unknown alpha source of alpha	ļ
	particles.	
5.	Spectrum expansion with Multi-channel Analyzer and decay ratios of	
	²⁴¹ Am.	-
6.	Beta spectroscopy	
	(i) Calibration with a pulser	-
_	(ii) Beta end point determination for ²⁰⁴ Tl	
	(iii) Conversion electron ratio.	
7.	Compton Scattering	
	(i) Simple Compton Scattering (Energy Determination)	
	(ii) Simple Compton Scattering (Cross-section Determination)	

8.	To study Rutherford Scattering of alpha particles from thin gold foil and Al	
_	foil.	
9		
	To determine Half-Lives of Radioactive sources prepared by neutron activation – In and Ag isotopes	
10.		<u>'</u>
	To study Gamma-gamma coincidence by	
	(i) Overlap coincidence method – ²² Na	
	(ii) Time to pulse height converter method – ²² Na	
	Total	84

S. No.	Name of Books / Authors	Year of Publication
1.	Leo W R, "Techniques for Nuclear & Particle Physics Experiments", Narosa	2000
2.	Kapoor S S and Ramamurthy V "Nuclear Radiation Detectors", New Age Publishers	1986
3.	ORTEC Lab Manual, "Experiments in Nuclear Science", ORTEC	1992

NAME OF DEPTT./CENTRE:	DEPARTM	ENT OF PHYSICS	
1. Subject Code: PH-617	Course Title:	Advanced Character Techniques	rization
2. Contact Hours: L: 3	T: 0	P: 0	
3. Examination Duration (Hrs.):	Theory	3 Practical	
4. Relative Weightage: CWS 15	PRS 0	MTE 35 ETE 50	PRE 0
5. Credits: 3 6. Sem	nester: Autumr	7. Subject Area: D	DEC

8. Pre-requisite:

PH-201M, PH-202M or equivalent

9. Objective:

To introduce the various methods of characterization of materials for their structural, electrical, magnetic and optical properties.

S.No.	Contents	Contact Hours
1	Crystal Structure Determination: Brief description of Crystal Lattices; X-ray diffractometer; Determination of Crystal Structure using X-ray diffraction	12
2	Electron Microscopes: Brief description of different microscopes like TEM, SEM, AFM; Different modes of operation of microscopes, sample preparation, Interpretation of electron diffraction and determination of Crystal Structure; Morphology of the Crystals.	11
, 3	Thermal Analysis: Thermogravimetric analysis, Differential thermal analysis and Differential scanning calorimetry and methodology; Determination of phase transitions using these methods.	0.5
4	Electrical and Magnetic Property: Measurement of Electrical conductivity in different materials, e.g. insulators, metals and semiconductors. Using Four Probe and Hall Effect method. Vibrating Sample Magnetometer (VSM), Superconducting Quantum interference Devices (SQUID)	8
5	Optical Characterization: Optical characterization of materials using Photoluminescence and UV-visible spectroscopy.	03

6	Chemical Analysis: Brief description to X-ray fluorescence, Atomic absorption and electronic s	pin	03
	resonance spectroscopy.		
	То	tal	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication /Reprint
1.	Culity B D, "Elements of X-ray Diffraction", Addison-Wesley.	2001
2.	Grundy P J and Jones G A, "Electron Microscopy in the Study of Materials", Edward Arnold	1976
3.	Egerton R F, "Physical Principles of Electron Microscopy", Springer	2008
4.	Willard, Merritt, Dean and Settle, "Instrumental Methods of Analysis", CBS publications	1991
5.	Fultz B and Howe J M, "Transmission Electron Microscopy and Diffractometry of Materials", Springer.	2007

NAME OF DEPTT/CENTRE:	DEPARTMENT OF	F PHYSICS
1. Subject Code: PH-619	Course Title: Introduc	ction to Quantum Field Theory
2. Contact Hours: L: 3	T: 0	P: 0
3. Examination Duration (Hrs.):	Theory 3	Practical 0
4. Relative Weightage: CWS 15	PRS 0 MTE 3	35 ETE 50 PRE 0
5. Credits: 3 6. Sem	ester : Autumn 7	7. Subject Area : DEC
8. Pre-requisite: PH-503, PH-5	508	

10. Details of Course:

mechanics.

9. Objective:

S.No.	Contents	Contact Hours
1.	Basics: Action principle; Euler-Lagrange equations of motion, second quantization; Symmetry (space-time and internal) Conserved Nöther charges.	4
2.	Tensors: Definitions of contravariant, covariant and mixed tensors, need to use tensors in relativistic quantum mechanics.	2
3.	Spin-0 (Klein Gordon Field Theory): Real scalar field theory and its canonical quantization; Normal Ordering; Charged scalar field theory and its canonical quantization, conserved Nöther current and charge, Propagator (also as vacuum expectation value of a time-ordered product), interpretation of negative-energy solutions as anti-matter; Recasting Klein-Gordon equation as a Schrödinger equation, Zitterbewegung.	7
4.	Spin-1/2 (Dirac Field Theory): Dirac Lagrangian for spinor fields, Feynman Gamma matrices and related identities; Covariance of the Dirac equation; Canonical quantization of the spinor fields, positive- and negative-energy spinors, positive- and negative-energy projectors, Lorentz transformations to boost from rest frame to lab frame; Propagator (also as vacuum expectation value of a time-ordered product), Discrete symmetries: Charge conjugation, Parity and Time reversal symmetries.	9
4.	Spin-1 (Gauge Field Theory): Covariant formulation of Maxwell's equations, (transverse) canonical quantization of the gauge field (in the Coulomb gauge),	5
1.	Scattering: LSZ reduction (for bosons and fermions), Wick's theorem, S-matrix, cross sections.	6

To familiarize students with applications of relativistic quantum

2.	Quantum Electrodynamics: Quantization of abelian gauge theories with fermions;	. 9
	Feynman Rules; Compton effect; Moller Scattering, radiative corrections; Anomalous	
<u> </u>	Magnetic Moment; Infrared Divergence; Lamb shift.	
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1.	Michio K, Quantum Field Theory: A Modern Introduction, Oxford University Press.	1993
2.	Claude I and Jean B. Z., "Quantum Field Theory, McGraw Hill College Div.	2006
3.	Lewis H R, "Quantum Field Theory", Cambridge University Press	2001
4.	Michael E. P, "An Introduction to Quantum Field Theory, Perseus Books Publishing	2002
5.	Lahiri A, Pal P B., A First Book of Quantum Field Theory, Narosa Publishing House	2005

NAME OF DEPTT./CENTRE:		DEPARTMENT OF PHYSICS				
1. Subject Code: P	H-621	Course Title:	Fiber an	ıd Nonlin	ear Op	tics
2. Contact Hours:	L: 3	T: 0		P: 0		•
••						er hande feet ongelee
3. Examination Dura	ation (Hrs.):	Theory	3 F	Practical	0	eli kun esir legi. Linda ini Linda eli
4. Relative Weightag	ge: CWS 1	PRS 0	MTE 35	ETE 50	PRE	0
5. Credits: 3	6. Se	mester: Autumi	n 7. Si	ubject Area: I	DEC	
8. Pre-requisite:	Nil				•	
9. Objective:		ce applications		nonlinear o	ptics, op	tical fiber

S.No.	Contents	Contact Hours
1.	Optical Waveguides: Rectangular waveguides: optical waveguides, planar mirror waveguides, electromagnetic analysis of planar optical waveguides, TE and TM modes of a symmetric and asymmetric planar waveguide, power associated with a mode.	12
2.`	Optical Fiber: Optical fiber waveguide, the numerical aperture, pulse dispersion in a step-index fiber, scalar wave equation and modes of a fiber, LP modes, single-mode fibers, material and waveguide dispersion for a communication link, attenuation, splice loss, methods of fabrication of optical fibers, optical fiber components, directional coupler, multiplexer, demultiplexer, fiber Bragg gratings, long-period fiber gratings, optical fibers in sensors, photonic crystal fibers.	12
3.	Nonlinear Optics: (a) Nonlinear optical media, nonlinear polarization and susceptibility 2 nd order nonlinear optics: optical second harmonic generation, sum frequency generation, difference frequency generation, optical parametric amplification and oscillation, three wave mixing.	
	(b) 3 rd order nonlinear optics: third harmonic generation, optical Kerr effect, self phase modulation, self focusing, spatial solitons, Raman gain, four wave mixing, optical phase conjugation, Raman and Brillouin scattering.	8
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1.	Ghatak A K and Thyagarajan K, "Optical Electronics", Cambridge University Press	2003
2.	Ghatak A K and Thyagarajan K, "Introduction to Fiber Optics", Cambridge University Press	1998
3.	Laud B B, "Lasers and Nonlinear Optics", Wiley Eastern	1992
4.	Saleh BEA and Teich MC, "Fundamantals of Photonics", Wiley Interscience	2007
5.	Snyder A and Love J, "Optical Waveguide Theory", Chapmann and Hall	1983
6.	Keiser G, "Optical Fiber Communications", McGraw Hill	2000

NAME OF DEPTT./	CENTRE:	DEPARTM	ENT OF	PHYSICS		
1. Subject Code: Pl	H-623	Course Title:	Genera	l Relativit	y	
2. Contact Hours:	L: 3	T: 0		P: 0		
3. Examination Durat	tion (Hrs.):	Theory	.3	Practical	0	i de la Applica El El Companyo
4. Relative Weightag	e: CWS 15	PRS 0	MTE 35	ETE 50	PRE	0
5. Credits: 3	6. Sem	nester: Autumi	า 7. S	Subject Area:	DEC	
8. Pre-requisite:	PH-505					
9. Objective:		the basics of eral relativity			etry and l	Einstein's

S.No.	Contents	Contact Hours
1.	Inertial mass and gravitational mass, gravitational redshift, action in relativity	3
2.	Principle of equivalence, metric tensor and the affine connection, geodesics.	5
3.	Covariant differentiation, analogy with electromagnetism, p-forms, generalized Stokes theorem.	5
4,	Curvature tensor, parallel transport, algebraic properties of the curvature tensor, Bianchi identities.	7
5	Lorentz transformation, representation of Lorentz group, conserved currents and energy momentum tensor	5
6	Einstein's field equations and some of their solutions: Robertson-Walker metric, Schwarzschild metric, black holes, deflection of light by Sun, precession of perihelia of planets. Expanding universe	8
7.	Expanding universe, Tetrad formalism, Killing vectors, maximally symmetric spaces.	5
8.	Kaluza-Klein theories an approach towards unification of, e.g., electromagnetism and gravity.	4
	Total	42

11.	Suggested Books:	. :
S.No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1.	Landau L D and Lifshitz E M, "The Classical Theory of Fields", 4th Ed. Elsevier.	. 2005
2.	Weinberg S, "Gravitation and Cosmology: Principles and Applications of the General Theory of Relativity", Wiley	1972
3.	Kaku M, "Quantum Field Theory: A Modern Introduction", Oxford University Press.	1993

NAME OF DEPTT./CENTRE:	DEPARTMEN	IT OF PHYSICS	
1. Subject Code: PH-625	Course Title: Pa	article Physics	
2. Contact Hours: L: 3	T: 0	P: 0	
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3. Examination Duration (Hrs.):	Theory 3	Practical	0
4. Relative Weightage: CWS 19	5 PRS 0 MT	TE 35 ETE 50	PRE 0
5. Credits: 3 6. Ser	nester: Autumn	7. Subject Area:	DEC

8. Pre-requisite:

PH-510

To introduce the basics of elementary particle physics.

9. Objective:10. Details of Course:

S.No.	Contents	Contact Hours
1.	Qualitative preview: A preview of particle physics, basic ideas of the four	2
	interactions – gravitational, electromagnetic, strong and weak.	
2.	Tools	8
	(i) Tensors: Definitions of contravariant, covariant and mixed tensors, need	
	to use tensors in relativistic quantum mechanics and particle physics; (ii)	
	Relativistic Kinematics: Lorentz transformations, 4-Vectors, energy and	
	momentum, collisions;	
	(iii) Scattering: Lifetimes and Cross Sections, Fermi's Golden Rule,	
	Feynman Rules, evaluation of scattering amplitudes and cross sections using	
	Feynman Rules.	
3.	Symmetries: Symmetries, Groups and Conservation Laws; Spin and Orbital	6
	Angular Momentum, Addition of Angular Momentum; Flavor symmetries;	
	Parity; Charge Conjugation; CP violation; Time reversal symmetry; CPT	
	Theorem; Noether's Theorem: Symmetry and conservation laws.	
4.	Electromagnetic Interaction:	8
	(i) Gauge Field Theory: Covariant formulation of Maxwell's equations,	
	(transverse) canonical quantization of the gauge field (in the Coulomb gauge);	
	(ii) QED (quantization of abelian gauge theories with fermions): Feynman	
	Rules, Compton effect, Møller Scattering, radiative corrections, Anomalous	
	Magnetic Moment, Lamb shift.	1

5.	Strong Interaction: (i) Pre-QCD: The structure of Hadrons, Probing a charge distribution with electrons: Inelastic electron -proton scattering, Partons and Bjorken scaling; (ii) QCD (quantization of non-abelian gauge theories with fermions): Yang-Mills theory, Parton model revisited, Feynman rules, Asymptotic freedom.	8
6.	Weak Interaction: (i) Phenomenology: Parity violation and the V-A form of the weak current, Muon decay, Pion decay, charged current, neutral currents, Cabibbo angle, weak mixing angle, CP Invariance, CP violation; (ii) Electroweak Unification (Glashow-Salam-Weinberg model): The basic electroweak interaction, effective current-current Interaction, Spontaneous symmetry breaking, Higgs mechanism and choice of the Higgs field, masses of gauge bosons and fermions, the complete Lagrangian.	10
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1	Halzen F and Martin A D, "Quarks and Leptons: Introductory Course in Modern Particle Physics", John Wiley and Sons, Inc	1990
2	Griffiths D, "Introduction to Elementary Particles", John Wiley and Sons Inc.	1987
3	Perkins D H, "Introduction to High Energy Physics", Cambridge University Press	2000
4	Georgi H, "Weak Interactions and Modern Particle Theory", Benjamin-Cummings Pub Co	1984
5	Kane G L and Kane G, "Modern Elementary Particle Physics", Westview Press	1993

NAME OF DEPTT./CENTRE:	DEPARTMENT	OF PHYSICS
1. Subject Code: PH-627	Course Title: Qua	ntum Theory of Solids
2. Contact Hours: L: 3	T: 0	P: 0
3. Examination Duration (Hrs.):	Theory 3	Practical 0
4. Relative Weightage: CWS 15	PRS 0 MTE	35 ETE 50 PRE 0
5. Credits: 3 6. Sem	ester: Autumn	7. Subject Area: DEC

solids using the many body technique.

PH-504 and PH-508

10. Details of Course:

8. Pre-requisite:

9. Objective:

S.No.	Contents	Contact Hours
1.	Many Body Techniques and the Electron Gas: Creation and annihilation operators, many particle wave function in occupation number representation, commutation relations, N-electron Hamiltonian in creation- annihilation operators form; One electron and two-electron, parts. Hartree-Fock ground state energy, free electron gas; Ground State energy in 1st order. Elementary idea of Greens functions.	12
2.	Plasma Oscillations in Free Electron Gas: Resume of plasma theory, quantum mechanical plasma theory, Energy of the ground state; Correlation Energy; Short range and long range correlation energy.	10
3.	Magnetism: Magnetism in Insulators; Heisenberg model; Spin waves; quantization of spin waves; Acoustic and optical magnons; Magnon specific heat; Antiferromagnitic Magnons; Magnetism in metals; Itinerant Ferromagnetism.	10
4.	Superconductivity: Electron-phonon interactions; Bound electron-pairs in a Fermi gas; Superconducting ground state; Hamiltonian solution of BCS equation for the energy-gas; Electrodynamics of superconductors, coherence length.	10
	Total	42

To provide deeper understanding of cooperative phenomenon in

S.No.	Name of Authors/ Books/Publishers	Year of Publication /Reprint
1.	Raimes S, "Many Electron Systems", North Holland Publishing Co.	2000
2.	Kittel C, "Quantum Theory of Solids", John Wiley and Sons	1987
3.	Ziamn J M, "Principles of Theory of Solids", Cambridge Univ. Press	2000
4.	Chaikin P M and Lubensky T C, "Principles of Condensed Matter". Cambridge Univ. Press	2000
5.	Kantorovich L, "Quantum Theory of the Solid State: An Introduction", Kluwer Academic Publishers	2004

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NAME OF DEPTT./	CENTRE:	DEPARTM	ENT OF	PHYSICS	3
1. Subject Code: PF	ł-629	Course Title:	Weather	Forecast	ing
2. Contact Hours:	L: 3	T: 0		P: 0	
3. Examination Durat	ion (Hrs.):	Theory	· · · · · · · · · · · · · · · · · · ·	Practical	0
4. Relative Weightage	e: CWS 15	PRS 0	MTE 35	ETE 50	0 PRE 0
5. Credits: 3	6. Sem	ester: Autumr	7. S	Subject Area	a: DEC
8. Pre-requisite:	None				
9. Objective:	To familiarize	with the dynam	ic meteorol	ogy of earth	's atmosphere
10. Details of Course:	•				

S.No.	Contents	Contact Hours
1.	Atmospheric Dynamics: Equation of motion, the geostrophic approximation, cyclostrophic motion; The thermal wind equation; The equation of continuity.	8
2.	The General Circulation: A symmetric circulation, Inertial instability, Barotropic instability; Baroclinic instability; Sloping convection; The general circulation of the middle atmosphere.	8 .
3,	Numerical Modelling of Weather: A barotropic model; Baroclinic models; Primitive equation models; Moist processes; Radiation transfer; Forecasting models.	.10
4.	Global Observations: Conventional observations; Remote sounding from satellites; Remote sounding of atmospheric temperature; Remote measurements of composition.	8
5.	Atmospheric Predictability and Climate change: Short term predictability; Variations of climate; Atmospheric feedback processes; Different kind of predictability	8
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication/ Reprint
1.	Houghton JT, "The physics of atmospheres", Cambridge University Press	1997
2.	Holton J R, "Introduction to dynamic meteorology", Academic Press,	1992
3.	Zdunkowski W and Boot A, "Dynamics of the Atmosphere", Cambridge University Press,	2003

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NAME OF DEPTT/CENTRE:	Department of	Physics	
1. Subject Code: PH-602	Course Title: Nucl	ear Astrophysics	
2. Contact Hours: L: 3	T: 1	program P: 0 Prokanto de la Prodrama	
3. Examination Duration (Hrs.):	Theory 3	Practical 0	áj i en
4. Relative Weightage: CWS 25	PRS 0 MTE	25 ET 50 PRE 0	
5. Credits: 4 6. Semes	ster: Spring	7. Subject Area: DEC	
8. Pre-requisite: PH-510			

9. Objective: To introduce the emerging field of nuclear astrophysics which attempts to understand how nuclear processes generate the energy of stars over their lifetimes and synthesize heavier elements.

S. No.	Particulars	Contact Hours
1.	Introduction: Astronomy-Observing the universe, Astrophysics- 'Explaining' the universe; General characteristics of Thermonuclear reactions; Sources of nuclear energy; Cross sections, stellar reaction rates, mean lifetime; Maxwell-Boltzmann velocity distribution, Astrophysical S – factor,	10
2.	Determination of reaction rates: Neutron and charged particle induced non-resonant reactions; Reactions through narrow and broad resonances	8
3.	Hydrogen and Helium burning : p-p chain, CNO cycles, other cycles like NeNa, MgAl; Creation and survival of ¹² C	9
4,	Explosive Burning and Nucleosynthesis beyond Iron: Silicon burning; Nucleosynthesis in massive stars, s – process, r - process	9
5.	Indirect methods in Nuclear Astrophysics: Coulomb dissociation, Trojan Horse and ANC methods; Neutron stars; Radioactive Ion Beams	6
	Total	42

SI. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Rolfs C E and Rodney W S, "Cauldrons in the Cosmos: Nuclear Astrophysics", The University of Chicago Press	2005
2.	Clayton D D, "Principles of Stellar Evolution and Nucleosynthesis", The University of Chicago Press	1984
3.	Glendenning N K, "Compact Stars", Springer	2000
4.	Boyd R, "An Introduction to Nuclear Astrophysics", The University of Chicago Press	2008

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NAME OF DEPTT.	/CENTRE	DEPARTN	MENT C	OF PHYSI	CS		
1. Subject Code: P	'H-604	Course Title:	Phys	ics of Na	nosys	tems	
2. Contact Hours:	L: 3	T: 1		P: (כ		
3. Examination Dura	ation (Hrs.):	Theory	3	Practic	al	0	
4. Relative Weighta	ge: CWS 2	PRS 0	MTE	25 ETE	50	PRE	0
5. Credits: 4	6. Se	mester: Spring	l <u>.</u> .	7. Subject A	Area: DE	EC	
8. Pre-requisite:	PH-303 or e	quivalent					
9 Objective:	To introduc	e the emerging	area of	nanotechno	logy.		

S.No.	Contents	Contact Hours
1.	Introduction - An overview of quantum mechanical concepts related to low-dimensional systems.	2
2.	Hetrostructures – Heterojunctions, Type I and Type II heterostructures, Classification of Quantum confined systems, Electrons and holes in Quantum wells, Electronic wavefunctions, energy subbands and density of electronic states in Quantum wells, Quantum wires, and Quantum dots, Effective mass mismatch in heterostructures, Coupling between Quantum wells, Superlattices	5
3.	Electron states - Wavefunctions and Density of States for superlattices, Excitons in bulk, in Quantum structures and in heterostructures, The unit cell for quantum well, for quantum wire and for quantum dot	6
4.	Nanoclusters and Nanoparticles – introduction, Metal nanoclusters- Magic numbers, Geometric structures, Electronic structure, Bulk to nanotransition, Magnetic clusters; Semiconducting nanoparticles; Rare-gas and Molecular clusters.	4
5.	Carbon Nanostructures – Introduction, Carbon molecules, Carbon clusters, Structure of C60 and its crystal, Small and Large Fullerenes and Other Buckyballs, Carbon nanotubes and their Electronic structure	3
6.	Properties of Nano Materials: Size dependence of properties, Phenomena and Properties at nanoscale, Mechanical/Frictional, Optical, Electrical Transport, Magnetic properties.	4
7.	Nanomaterial Characterization: Electron Microscopy, Scanning Probe Microscopies, near field microscopy, Micro- and near field Raman spectroscopy, Surface-enhanced Raman, Spectroscopy, X-ray photoelectron spectroscopy.	1

8.	Synthesis of nanomaterials: Fabrication techniques: Self-Assembly, Self-Replication, Sol-Gels. Langmuir-Blodgett thin films, Nanolithograph, Bioinspired syntheses, Microfluidic processes, Chemical Vapor Deposition, Pulse laser deposition.	8
9.	Applications of Nanomaterials: Nanoelectronics, Nanosensors, Environmental, Biological, Energy Storage and fuel cells.	3
	Total	42

S.No.	Name of Authors/ Books/Publishers	Year of Publication /Reprint
1.	Edelstein A. A. and Cammarata R. C., "Nanomaterials-Synthesis, Properties and Applications", Institute of Physics Publishing, London	1998
2.	Shik, A, "Quantum Wells: Physics and Electronics of two-dimensional systems", World Scientific	1999
3.	Benedek et al G., "Nanostructured Carbon for advanced Applications", Kluwer Academic Publishers	2001
4.	Harrison, P, "Quantum Wells, Wires, and Dots: Theoretical and Computational Physics", John Wiley	2000
5,	Mitin, VV, Kochelap, VA and Stroscio, MA "Quantum Heterostructures: Microelectronics and Optoelectronics", Cambridge University Press	1999
6.	Poole, Jr. CP and Owens, FJ, "Introduction to Nanotechnology", Wiley India.	2006

NAME OF DEPTT./CENTRE:	DEPARTMENT OF	PHYSICS
1. Subject Code: PH-606	Course Title: Supe	rfluidity and Superconductivity
2. Contact Hours: L: 3	T: 1	P: 0
3. Examination Duration (Hrs.):	Theory 3	Practical 0
4. Relative Weightage: CWS 25	PRS 0 MTE	25 ETE 50 PRE 0
5. Credits: 4 6. Sem	nester: Spring	7. Subject Area: DEC
8. Pre-requisite: PH-404		

9. Objective: It introduces advanced concepts of superfluidity and superconductivity and their interrelationship.
 10. Details of Course:

S. No.	Contents	Contact Hours
1.	Superfluidity: Basic properties of superfluid ⁴ He and ³ He; Bose-Einstein condensation in an Ideal Bose Gas; Bose-Einstein Condensation in Interacting Gases, Condensate Wave Function.	, 8
2.	Theory of Bose Fluids: Landau Criterion for Superfluidity. Excitations in a uniform Gas — Bogoliubov Transformation; Excitations in a Trapped Gas — Weak Coupling, Excitations in Non-uniform Gases.	9
3.	Vortex States: Quantization of Circulation, Quantized Vortices in He-II; Quantized Vortices in Superconductors; Comparison of He-II and Superconducting Vortices; Dynamics of Vortex States.	9
4.	Ginzburg-Landau Theory: Ginzburg Landau equations, second order critical fields; Abrikosov vortex lattice; Relation of GL theory with BCS theory; Ginzburg-Pitaevskii equations for He-II; Broken symmetry.	8
5.	High-Tc Superconductivity: Nature and various mechanisms of High Tc superconductivity; Equation for the critical temperature and strong electron-phonon coupling; SDW and CDW.	8
	Total	42

S. No.	Name of Authors/ Books/Publishers	Year of Publication/Reprint
1.	Chaikin P M and Lubensky T C, "Principles of Condensed Matter Physics", Cambridge University Press	1995
2.	Tilley D R and Tilley J, "Superfluidity and Superconductivity" (3 rd Ed), Overseas Press	2005
3.	Suneto T and Nakahara M, "Superconductivity and Superfluidity", Cambridge University Press	2005
4.	Pethick C J and Smith H, "Bose-Einstein Condensation in Dilute Gases", Cambridge University Press	2002
5.	Pitaevskii L and Stringari S, "Bose-Einstein Condensation", Clarendon Press	2003

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Department Core (DCC)

B. ARCH. I YEAR

NAME OF DEPTT./CENTRE: Architecture and Planning Department

1. Subject Code: AR-101 Course Title: Basic Design and Model Making

2. Contact Hours:

L: 1

T: 0

P: 6

3. Examination Duration (Hrs):

Theory 0

Practical 3

4. Relative Weightage: CWS 0 PRS 50 MTE 20 ETE 0 PRE 30

5. Credits 4

6. Semester: Autumn

7. Subject Area: DCC

8. Pre-requisite: Nil

9. Objective:

To enable students to develop an understanding of basic design elements and forms as basis for architectural design and to gain basic skills needed for model making and carpentry.

S. No.	Contents	Contact Hours
1.	Relationship between basic design and architectural design, comprehensive understanding of space, form, order and design.	2
2.	Study of space, its scale, proportions and form, conception and space breaking through compositions and models.	2
3.	Study of elementary two dimensional shapes and three dimensional form.	2 .
4.	Ordering combination principles and their application in building through exercises in design of murals, screens and voids in walls.	2
5.	Block and detailed building models.	2
6.	Cutting, fine joints and joinery in cardboard and wood/ plastic.	2
7.	Model making in paper, wood and plastics and demonstration in metal work.	2
	Total	14

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Parmar, V.S., "Design Fundamentals in Architecture", Somoiya Publications.	1973
2.	Morgan, C.L., "Jean Nouvel – The Elements of Architecture", Thames and Hudson.	1998
3.	Ching, F.D.K., "Architecture: Form, Space and Order", 3 rd Ed., John Wiley & Sons.	2007
4.	Kieran, S. and Timberlake, J., "Elements of a New Architecture", Princeton Architectural.	2008

NAME OF DEPTT./CENTRE: Architecture and Planning Department

Subject Code: AR-103 Course Title: Architectural Graphics -I

P: 6

2. Contact Hours: L: 1 T: 0

3. Examination Duration (Hrs): Theory 0 Practical 3

4. Relative Weightage: CWS 0 PRS 50 MTE 20 ETE 0 PRE 30

5. Credits 4 6. Semester: Autumn 7. Subject Area: DCC

8. Pre-requisite: Nil

9. Objective:

To introduce the fundamentals of architectural drawing through the study of architectural graphic symbols, orthographic projections, section of solids and sciography and to impart an understanding of architectural plans, elevations and sections.

S. No.	Contents	Contact Hours
1,	Architectural Graphic Fundamentals: Lines, lettering and dimensioning, reduction and enlargement of drawings on different scales, representation of materials and architectural elements through architectural graphic symbols, introduction to architectural plans, elevations and sections.	5
2.	Orthographic Projections: Principles and projection methods of orthographic projection (third angle projection), straight lines, planes, solids and development of surfaces, section of solids	7
3,	Sciography: Study of shades and shadows cast by simple architectural forms on plain surfaces.	2
	. Total	14

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Bhatt, N.D. and Panchal, V.M., "Engineering Drawing – Plane and Solid Geometry", 48th Ed., Charotar Publishing House.	1996
2.	Griffin, A.W. and Brunicardi, V.A., "Introduction to Architectural Presentation Graphics", Prentice Hall.	1998
3.	Ching, F.D.K., "Architectural Graphics", 4th Ed., John Wiley.	2003

NAME OF DEPTT./CENTRE: Architecture and Planning Department

i.	Subject Code: AR-102	Course Title: Architectural Design-
		· · · · · · · · · · · · · · · · · · ·

2. Contact Hours: L: 1 T: 0 P: 6

3. Examination Duration (Hrs): Theory 0 Practical 7

4. Relative Weightage: CWS 0 PRS 60 MTE 20 ETE PRE 20

5. Credits 4 6. Semester: **Spring** 7. Subject Area: **DCC**

8. Pre-requisite: Nil

9. Objective:

To introduce architectural design process to the students and to lead them to understand the fundamentals of space through the design of simple buildings.

S. No.	Contents	Contact Hours
1.	Study of anthropometrics and their relationship with the dimensions of objects of daily use.	3
2.	Determining spaces for activities such as living, dining, sleeping and conveniences.	3
3.	Measured drawing of a small building such as, a small house or office.	2
4.	Simple circulation/flow diagrams for small building projects	2
5.	Three dimensional organization of a variety of forms to create built forms, importance of shades and shadows in the entire composition, layout of repetitive units within a site to create interesting and functional compositions.	4
	Total	14

Design Exercises:

Very small building structures, such as a compound wall, gate, milk booth, vendor stall, guard room, cycle stand, bus stop etc.

Visits to architectural sites.

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Ching, F.D.K., "Design Drawing", Van Nostrand Reinhold.	1998
2.	Neufert, P., "Architects' Data", 3 rd Ed., Blackwell Science.	2000
3.	Watson, D. (Editor), "Time-saver Standards for Architectural Design: Technical Data for Professional Practice", 8 th Ed., McGraw-Hill.	2005

NAME OF DEPTT./CENTRE: Architecture and Planning Department

1. Subject Code: AR-104 Course Title: Architectural Graphics -!!

2. Contact Hours: L: 1 T: 0 P: 6

3. Examination Duration (Hrs): Theory 0 Practical 3

4. Relative Weightage: CWS 0 PRS 50 MTE 20 ETE 0 PRE 30

5. Credits 4 6. Semester: **Spring** 7. Subject Area: **DCC**

8. Pre-requisite: AR-103

12. Objective:

To impart knowledge of isometric views, axonometric views and perspectives.

13. Details of Course:

S. No.	Contents	Contact Hours
1.	Isometric and Axonometric Views: Solids, compositions and buildings	6
2.	Perspective Drawing: Definition of perspective technique and their role in drawing perspectives, one point, two point and three point perspectives of geometrical shapes leading to perspectives of built forms, exercises in parallel, angular and bird's eye views, shades and shadows cast by simple forms on plain surfaces	8
	Total	14

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Bhatt, N.D. and Panchal, V.M., "Engineering Drawing – Plane and Solid Geometry", 48th Ed., Charotar Publishing House.	1996
2.	Griffin, A.W. and Brunicardi, V.A., "Introduction to Architectural	1998

	Presentation Graphics", Prentice Hall.	,i
3.	Ciriello, M., "Architectural Design Graphics", McGraw-Hill.	2002
4.	Ching, F.D.K., "Architectural Graphics", 4th Ed., John Wiley.	2003
5.	Carpo, M., "Perspective, Projections and Design: Technologies of Architectural Representation", Routledge.	2008

NAME OF DEPTT./CENTRE: Architecture and Planning Department

1. Subject	Code: AR-106	Course Title: Introduction to Building Materials an	ıc
		Construction - I	

2. Contact Hours:

L: 1

T: 0

P: 4

3. Examination Duration (Hrs):

Theory

Practical

- 4. Relative Weightage: CWS 50 PRS 0 MTE 20 ETE 30 PRE 0
- 5. Credits 3
- 6. Semester: Spring
- 8. Subject Area: DCC

- 7. Pre-requisite: Nil
- 9. Objective:

To introduce the students to building materials, their properties and application in building construction, alongwith the construction of some basic components of a building.

S. No.	Contents	Contact Hours
1.	Mud: Stabilisation and use for walling and terracing.	1
2.	Bricks: Constituents and properties of brick earth, manufacturing of various types of bricks, decorative brick work and jail work Stone: Properties, varieties and their characteristics, stone masonry.	2
3.	Timber: Varieties of Indian timbers, their characteristics and suitability for different uses, defects and decay, seasoning and preservation; manufactured timber products and their applications as insulation materials and decorative materials etc.	2
4.	Surface Finishes: Various types of plaster and their application, guniting and its application; pointing, glazing and gluing.	2
5.	Lime and Cement: Sources, classification, properties and method of manufacturing, testing, mixing and uses.	1
6.	Concrete: Composition, properties and uses, light weight and no	1

	fines concrete and its uses.	
7.	Various Construction Components of Buildings	5
	Foundations: Definition, safe bearing capacity of soils and methods of improving foundations, causes of failure and remedies, types of footings.	
	Temporary work: Excavation and timbering of trenches, shoring, underpinning and scaffolding.	
	Basement: Damproofing, construction details of walls, floors, foundations etc.	<u></u>
`	Walls: Masonry walls, hollow and panel walls, practical considerations during construction, hollow concrete and glass block construction	
	Arches: Different forms and centering.	
	Total	14

Practicals: Construction of simple brick walls, stone masonry and brick arches.

Site Visits: Visits to construction sites,

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Kumar, S.K., "Building Construction", 19 th Ed., Standard Publishers Distributors.	2001
2.	Allen, E. and Iano, J., "Fundamentals of Building Construction: Materials and Methods", Wiley.	2004
3.	Mehta, M., Scarborough, W. and Armpriest, Diane, "Building Construction: Principles, Materials and Systems", Pearson Prentice Hall.	2008

NAME OF DEPTT./CENTRE: Architecture and Planning Department

1. Subject Code: AR-108 Course Title: Principles of Architecture

2. Contact Hours: L: 2 T: 1 P: 0

3. Examination Duration (Hrs) Theory 2 Practical 0

4. Relative Weightage: CWS 25 PRS 0 MTE 25 ETE 50 PRE 0

5. Credits 3 6. Semester: **Spring** 7. Subject Area: **DCC**

8. Pre-requisite: Nil

9. Objective:

To introduce the students to the theory of architectural design and the principles pertaining to visual and aesthetic aspects of architecture.

S. No.	Contents	Contact Hours
1.	Understanding of relevant terms — architecture, design art, fine art, visual art, architectural design and other types of design; comparisons of aesthetics in art and architecture.	3
2.	Principles of two dimensional design elements, such as, point, line, direction, shape, size, colour and texture; Three dimensional design profiles of geometric forms and their arrangements in different compositions.	3
3.	Harmony and contrast in 2-D and 3-D design; interplay of light and shade on building blocks and their effect.	5
4.	Scale and proportion in architecture; Le Modular and other concepts.	5
5.	Style, rhythm, balance, unity and order.	6
6.	Form and function in architecture; use of building materials, construction techniques and engineering services for different functions.	6
	Total	28

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Ching. F.D.K., "A Visual Dictionery of Architecture", Van Nostrand Reinhold.	1997
2.	Ching. F.D.K., "Architecture Theoretician', Wiley.	2007
3.	Fisher, T., "Architectural Design and Ethics: Tools for Survival", Architectural Press.	2008

NAME OF DEPTT./CENTRE: Architecture and Planning Department

1. Subject Code: AR-110 Course Title: Colour Fundamentals and Its Application

2. Contact Hours:

L: 1

T: 0

P: 2

3. Examination Duration (Hrs):

Theory 0

Practical 2

4. Relative Weightage: CWS 0 PRS 50 MTE 20 ETE 0 PRE 30

5. Credits 2

6. Semester: Spring

7. Subject Area: DCC

8. Pre-requisite: Nil

12. Objective:

To study colour theory, classification of colours and psychological factors which govern the choice of colour schemes in buildings.

13. Details of Course:

S. No.	Contents	Contact Hours
1.	Physiology of colour, colour mixtures, colour systems, colour organisation	7
2.	Psychological factors governing colour schemes, application of colour schemes	. 5
3.	National and International standards on colours	. 2
	Total	14

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Toy, Maggie (Ed.), "Colour in Architecture", Academy Editions.	1996
2.	Yanes, M.D. and Dominguez, E.R., "Freehand Drawing for Architects and Interior Designers", Norton.	2005

NAME OF DEPTT./CENTRE.	DEPARTMENT OF MATHEMATICS
1. Subject Code: MA-905	Course Title: Evolutionary Algorithms
2. Contact Hours: L: 3	T: 1 P: 0
3. Examination Duration (Hr	rs.): Theory 3. Practical 0
4. Relative Weightage: CV	NS 25 PRS 0 MTE 25 ETE 50 PRE 0
5. Credits: 3	6. Semester: Both 7. Subject Area: DEC

8. Pre-requisite: M.Sc./M.Tech/MCA with sufficient background in Optimization Techniques and Computer Programming

9. Objective:

To acquaint the students with recent advances in Evolutionary Algorithms and their use in solving discrete and continuous optimization problems in Engineering, Science and Industry.

S.No.	Contents	Contact Hours
1	Genetic Algorithms: GA concepts – encoding, fitness function, population size, selection, crossover and mutation operators, along with the	10
	methodologies of applying these operators. Binary GA and their operators, Real Coded GA and their operators.	
2	Particle Swarm Optimization: PSO Model, global best, Local best, velocity update equations, position update equations, velocity clamping, inertia weight, constriction coefficients, synchronous and asynchronous updates, Binary PSO.	10
3	Memetic Algorithms: Concepts of memes, Incorporating local search as memes, single and multi memes, hybridization with GA and PSO, Generation Gaps, Performance metrics.	5
4	Discrete Optimization: Use of Evolutionary Computations to solve Traveling Salesman Problem. Time Table Problem, Vehicle Routing Problem.	5
5	Constrained Optimization: Methods based on rejection strategies, repair strategies, specialized operators and penalty functions.	5
6	Multi-Objective Optimization: Linear and nonlinear multi-objective problems, convex and non – convex problems, dominance – concepts and properties, Pareto – optimality. Use of Evolutionary Computations to solve multi objective optimization, bi level optimization. Theoretical Foundations.	7
1.7	TOTAL	42

S. No.	Name of Authors / Books / Publishers	Year of Publication/ Reprint
1	Hart W.E., Krasnogor N., Smith J.E., "Recent Advances in Memetic	2005
	Algorithms", Springer Berlin Heidelberg	
2	Deb K., "Multi-Objective Optimization using Evolutionary Algorithms",	2002
	John Wiley and Sons.	
3	Coello C.A., Van Veldhuizen D.A. and Lamont G.B., "Evolutionary	2002
	Algorithms for solving Multi Objective Problems", Kluwer.	·
4	Deb K., "Optimization for Engineering Design Algorithms and Examples",	1998
	Prentice Hall of India	
5	Gen M. and Cheng R., "Genetic Algorithms and Engineering Design",	1997
<u>. </u>	Wiley.	
6	Michalewicz Z., "Genetic Algorithms + Data Structures = Evolution	1992
	Programs", Springer-Verlag, 3rd Ed.	

NAME OF DEPARTMENT:	Metallurgical & Mate	erials Engineering
1. Subject Code: IMT-01	Course Title:	Non Destructive Testing
2. Contact Hours:	L: 3 ; 1 5 T; 1;	P: 0
3. Examination Duration (Hrs):	Theory: 0 3	Practical: 0 0
4. Relative Weightage: cws. 2	5 PRS: 0 0 MT	E. 2 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 6 Sem	nester: Autumn	Spring Both
7. Pre-requisite: Nil	8. Subject Area:	ESEC

To make students aware of the importance of quality control & quality assurance and role of non destructive testing..

10. Details of the Course:

9. Objective

S. No	Particulars	Contact Hours
1.	Non destructive testing and its comparison with destructive	3
	testing. Role of NDT in quality control	
2.	Liquid penetrant inspection its, principles, advantages,	6
ļ	limitations & applications.	. •
3.	Magnetic particle inspection	6
4.	Ultrasonic inspection	6
5.	Eddy current inspection	6
6.	Radiography,	6
7.	Organizing for quality, fitness for use concept.	5
	Quality control: Statistical quality control, control charts, control	
	chart attribute & variables and acceptance sampling	
8.	Quality assurance & ISO 9000:2000	4
	Total	42

SI No	Name of Authors/BootsPublisher	Year of Publication
1.	Metals Hand Book Ninth edition volume 17, Non Destructive	·
	Evaluation & Quality Control ASM Hand book committee	1989
	ASM International Metals Park, Ohio,	
2.	K.C. Srivastava, Handbook of magnetic particle testing American Book Centre, Delhi.	1998
3	K.C. Srivastava, Handbook of liquid penetrant testing, American Book Centre, Delhi.	1997
4.	K.C. Srivastava, Handbook of ultrasonic testing, International Inspection Services, Delhi	2001
5.	Engene. L. Grant and Richard. S. Larenwork: Statistical quality control Tata McGraw-hill Publishing Co. Ltd., New Delhi, India	2000
6.	Non destructive evaluation science and technology for quality	
	improvement: 658.562 W83NDDC, Indian National Academy of Engineering:	1999

DEPARTMENT OF METALLURGICAL AND MATERIALS ENGINEERING INDIAN INSTITUTE OF TECHNOLOGY

TEACHING & EXAMINATION SCHEME FOR METALLURGICAL AND MATERIALS ENGINEERING PROGRAMME First Year

Structure-A

Teaching Scheme Contact Hours/Week Exam Duration Relative Weightage (%) Subject Code Course Title Practical Subject Area No. Credits MTE Semester - I (Autumn) MA-101 Mathematics-I BSC 0 3 25 25 50 2. PH-101 Physics-I BSC 3 2 3 15 15 15 40 15 ESC EE-101 Electrical Science 3 2/2 3 3. 15 15 30 40 : 4... EC-101A Computer System & programming 3 4. 2 3 15 15 30 40 ESC EC-101B Fundamentals of Object oriented Programming 3 0 2 3 15 15 30 40 ESC 2 MI-102 Manufacturing Techniques 3 0 2 2 15 15 30 40 GSC BT-101 Fundamentals of Biotechnology 2 0 0 2 15 35 50 HS-101 Technical Communication HSSMC 2 2 25 25 50 24 17 3 9 Total Semester - II (Spring) BSC 25 50 0 3 25 MA-102 Mathematics-II BSC 15 3 2 3 15 15 40 15 CY-101 Chemistry 2 ESC 2/2 15 EC-102 Fundamentals of Electronics 3 15 30 40 3. ESC 3 25 2 0 4 25 50 4. CE-101 **Engineering Graphics** Ee ESC 2 25 2 0 25 50 Thermodynamics 5. MI-103 GSC 2 15 2 2 0 0 35 50 CE-102 Environmental Studies No HSSMC 2 2 0 0 2 15 35 50 HS-102 Behavioural Science 24 Total 17

Appendix 'E Senate/29.2.6

Appendix - 1

TEACHING & EXAMINATION SCHEME FOR METALLURGICAL AND MATERIALS ENGINEERING PROGRAMME Second Year

Appendix - 1

		Teaching Scheme			Contact Hours/Week			Exam Duration		Relative Weightage (%)				
S. No.	Subject Code	Course Title	Subject Area	Credits	L	T	P	Theory	Practical	CWS	PRS	MTE	ETE	PRE
		_1,	Semester	III (Au	tumn)		<u> </u>				1		··· J	
1.	CE-201	Computer Aided Graphics	ESC	2	1	0	2	2		-	25	25	50	-
2.	HS-201	Economics	HSSMC	3	2	1	0	2	-	25	-	25	50	-
3.	MI-201A	Solid Mechanics	ESC	4	3	I	0	3	-	25	•	25	50	-
4.	MT-203	Metallurgical Thermodynamics and Kinetics	DCC	3	3	0	0	3	-	15	-	35	50	-
5.	MT-205	Physical Metallurgy	DCC	4	3	0	2	3	-	15	15	30	40	-
6.	MT-207	Transport Phenomena in Metallurgy	DCC	3	2	1	0	2	-	25	-	25	50	-
7-	MT-209	Particulate Processing	DCC	3	2	0	2	2	-	15	15	30	40	-
)		Total		22	16	3	6							
	<u> </u>		Semester-	- IV (Sp	ring)		-		•					
1.	PH-201	Physics-II	BSC	3	3	0	0	3	-	15	-	35	50	-
2.	BM-201	Managements concepts and practices	HSSMC	3	2	1	0	2.		25	-	25	50	-
3.	CH-201	Energy Resources and Conversion	GSC	2	2	0	0	2	-	25	-	25	50	-
1.	MT-202	Electrical and Electronic materials	DCC	3	2	I	0	2	-	25	-	25	50	
5.	MT-204	Metal Extraction Principles	DCC	3	2	0	2	2	-	15	15	30	40	-
j	MT-206	Fuels, Furnaces and Refractories	DCC	3	2	0	2	2	-	15	15	30	40	-
7.	MT-208	Metallurgical and Instrumental Analysis lab.	DCC	2	0	0	4	-	3	-	25	25	-	50
<u> </u>	IE	Institute Elective-I (BSC)	BGSEC	3	2	1	0	2	-	25	-	25	50	-
	<u> </u>	Total		22	15	3	8							

14.

TEACHING & EXAMINATION SCHEME FOR METALLURGICAL AND MATERIALS ENGINEERING PROGRAMME Third Year

Teaching Scheme			Cont	act Hou	rs/Week	Exam Duration		Relative Weightage (%)						
S. No.	Subject Code	Course Title	Subject	Credits	L	T	P	Theory	Practi- cal	CWS	PRS	MTE	ETE	PRE
			Semester	– V (Aut	umn)		<u> </u>	-1			l			
1.	MT-301	Ceramics and Metal Powder Processing	DCC	3	3] 0	2/2	3	-	15	15	30	40	 _
2.	MT-303	Mechanical Behaviour of Materials	DCC	4	3	1	2/2	3	-	15	15	30	40	-
3.	MT-305	Modelling and Simulation of Metallurgical Processes	DCC	3	12	0	2	2		15	15	30	40	
4.	MT-307	Metal Processing	DCC	4	3	0.	2	3	-	15	15	30	40	-
5.	MT	Departmental Elective-I	DEC	3	3	0	0	3	-	15		35.	50	-
6.	IE	Institute Elective-II (HSSMC)	IE .	3	3	10	0	3	-	15	-	35	50	
		Total		20	17	10	07			- 	-			+
	_4		Semeste	r VI (Spr	ing)		_ 	<u> </u>	<u></u>					
1.	MT-302	Corrosion Science and Technology	DCC	3	2	0	2	2	-	15	15	30	40	Ţ-
2. 2.	MT-306	Phase Transformation and Heat Treatment	DCC	4	3	0	2	3	3	15	15	15	40	15
3.	MT-304	Iron Making	DCC	4	3	i	0	3		25		25	50	-
4.	MT-308	Communication Skills	. DCC	2	0	0	2	-	-	100	 -	<u> -</u> -	-	
5.	MT-	Departmental Elective-II	DEC	3	3	0	0	3	-	15	-	35	50	-
6.	IE .	Institute Elective-III (BSC)	IE	3	3	0	0	3	-	15	-	35	50	-
7.	IE	Institute Elective-IV (HSSMC)	IE	3	3	0	0	3	-	15	-	35	50	-
		Total		22	17	1	6						1.	
	 			<u> </u>			<u> </u>							

	Teaching Scheme			Con	tact Hou	rs/Week	Exam	Duration	Relati	e Weightan	e (%)		
Subject Code	Course Title	Subject	Credits	L	T	P	Theory	Practi-	CWS	PRS		ETE	PRE
		Semester -	- VII (A	atumn)	I							_:	
MT-401	Composites and Polymers	DCC	4	3	1	0	3		25		25	50	- -
MT-403	Steel Making	DCC	4	3	1	0	3		25	-	25	50	
MT-405	Non-ferrous Technology	DCC	4	3	1	0	3	-	25	-	25	50	
MT-491	Minor Project	DCC	2	0	0	4	-		-	100	-		-
MT-492	Major Project	DCC	-	0	0	3	-		-		-		-
MT-	Departmental Elective-III	DEC	3	3	0	0	3		15	: - : -	35	50	~
MT-	Departmental Elective-IV	DEC	3	3	0	0	3		15		35	50	-
IÉ	Institute Elective-V (ESC)	IE	4	3	I	0	3		25	-	25	50	-
	Total		24	18	4	07	† 		-		+-		
									_				<u> </u>
		Semester -	- VIII (S	pring)				:		<u></u>			
MT-402	Mechanical working of metals	DCC .	4	3	0	2	3	Ţ-	15	15	30	40	. -
MT-490	Practical/Field Training	DCC	2	0	0	2	-	-	-	100	-	-	-
MT-492	Major Project	DCC	8	0	0	12	-	 	-	-	-	-	100
MT-	Departmental Elective-V	DEC	3	3	0	0	3	-	15	-	35	50	- -
IE	Institute Elective VI (ESC)	IE .	4	3	1	0	3		25	li -	25	50	-
-	Total		21	09	1	16.	1			1			_
					+	1			 		1		
	MT-403 MT-405 MT-491 MT-492 MT- IE MT-402 MT-490 MT-492 MT-	MT-401 Composites and Polymers MT-403 Steel Making MT-405 Non-ferrous Technology MT-491 Minor Project MT-492 Major Project MT- Departmental Elective-III MT- Departmental Elective-IV IE Institute Elective-V (ESC) Total MT-402 Mechanical working of metals MT-490 Practical/Field Training MT-492 Major Project MT- Departmental Elective-V IE Institute Elective VI (ESC)	Subject Code Course Title Semester MT-401 Composites and Polymers DCC MT-403 Steel Making DCC MT-405 Non-ferrous Technology DCC MT-491 Minor Project DCC MT-492 Major Project DCC MT- Departmental Elective-III DEC MT- Departmental Elective-IV DEC IE Institute Elective-V (ESC) IE Total Semester MT-402 Mechanical working of metals DCC MT-490 Practical/Field Training DCC MT-492 Major Project DCC DCC DCC DCC DCC DEC DEC DCC DEC DCC DCC DEC DCC DEC DCC DCC DCC DCC DEC DCC DCC	Subject Code Course Title	No. Subject Code Course Title	No. Subject Code Course Title	MT-401 Composites and Polymers DCC 4 3 1 0	MT-401 Composites and Polymers DCC 4 3 1 0 3	MT-401 Composites and Polymers DCC 4 3 1 0 3 -	Subject Code Course Title	MT-401 Composites and Polymers DCC 4 3 1 0 3 - 25 -	Subject Code Course Title	Subject Code Course Title Fig. 2 Fig. Course Title Fig. 2 Fig. Course Title Fig. 2 Fig. Fig.

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NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-301	Course Title: Ceramics and Metal Powder Processing
2. Contact Hours:	L: 3 ; T: 0; P: 2/2
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS: 1	5 PRS: 1 5 MTE: 3 0 ETE: 4 0 PRE: 0 0
5. Credits: 0 3	6 Semester: Autumn
7. Pre-requisite: MT-209	8. Subject Area: DCC

9. **Objective:** This course aims at imparting knowledge on principles of ceramic and metal powder processing and technologies for making ceramic and metal powder based

engineering products.

Sl. No.	Contents	Contact Hours
. 1	Introductory Overview: General characteristics of ceramics, ceramic microstructures, introduction to ceramic crystal structures.	2
2	Ceramic Powder Production and Consolidation: Synthesis by mechanical methods and chemical methods; Powder characterizations: physical characterization relating particles types, shapes, size, size distribution, their measurements, surface area, porosity; chemical	8
	compositions, phase composition and surface characterization; Dry and semidry pressing methods: die compaction, isostatic pressing; Casting methods: slip casting, pressure casting, tape casting; Additives in forming process, plastic forming methods: extrusion and injection moulding.	
3	Colloidal Processing : Colloid types, electrostatic stabilization, polymeric stabilization, rheology of colloidal suspension.	3
4	Sintering of Ceramics: Defects and defect chemistry; solid state sintering, atomic mechanisms, coarsening, densification, sintering kinetics: sintering stages – initial stage sintering, intermediate sintering, coarsening and grain growth growth kinetics; Liquid phase sintering: introduction, the different stages, controlling kinetics and thermodynamic factors.	7
5	Ceramic Phase Diagrams: Binary systems: complete solid solubility, eutectic diagrams with partials solid solubility and no intermediate compounds, partial solid solubility with formation of intermediate compounds; Ternary systems.	3

6	Glasses: Processing of glasses and glass-ceramics	3
	Metal Powders-Characteristics and Testing: Introduction to powder metallurgical applications, pressing properties, properties of sintered compacts.	3
7	Powder Production, Characterization, Compaction and Sintering: Mechanical processes, physico chemicals and chemical processes; Pressure-less shaping, cold pressure shaping and pressure shaping technique; tooling system, Sintering furnaces and their classifications; sintering atmosphere; vacuum sintering.	10
8	Powder Products and Selected Applications: Classifications; Friction materials; tool materials.	3
	Total	42

List of Practicals:

- 1. Determination of ceramic oxide powder densities using pycnometric method.
- 2. Powder processing and preparation of green compacts using uniaxial pressing.
- 3. Study of solid state sintering behaviour of the samples at different temperature and soaking time periods and study their densification characteristics.
- 4. Estimation of the particle size of the powder from X-ray diffraction line broadening technique.
- 5. Determination of average grain size by linear interception method.
- 6. Study of voltage vs. current (V-I) characteristics of ceramic varistors and determination of their nonlinear exponent.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Reed, J. S., "Introduction to the Principles of Ceramic Processing", A Wiley-Interscience Publication.	1988
2	Rahaman, M. N., "Ceramic Processing and Sintering", Marcel Dekker Inc.	1995
3	Richerson, David W., "Modern Ceramic Engineering – Properties, Processing and use in Design", Marcel Dekker, Inc.	1992
4	German, R.M., "AZ of Powder Metallurgy", MPIF	2005
5	Hirschhorn, J.S., "Introduction to Powder Metallurgy", APMI, USA	1975
6	Masuda, H., "Powder Technology Handbook", Taylor and Francis	2006

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-302	Course Title: Corrosion Science and Technology
2. Contact Hours:	L: 2 ; T: 0; P: 2
3. Examination Duration (Hrs):	Theory: 0 2 Practical: 0 0
4. Relative Weightage: CWS:	1 5 PRS: 1 5 MTE: 3 0 ETE: 4 0 PRE: 0 0
5. Credits: 0 3	6. Semester: Spring
7. Pre-requisite: CY-101	8. Subject Area: DCC

9. Objective: To familiarise the students with the basic knowledge of corrosion behaviour of metals and their protection.

Sl.No.	Contents	Contact Hours
1	Thermodynamics of Corrosion: Free energy change, EMF, Galvanic, Pourbaix diagram.	4
2	Electrochemical Theory: Corrosion rates; Activation polarization, concentration polarization, anodic, cathodic, mixed control. Passivation, Tafel equation.	6
3	Types of Corrosion: Different forms of corrosions-uniform, Galvanic, crevice, pitting, intergranular, erosion-corrosion, scc, hydrogen cracking, corrosion fatigue, fretting corrosion. Effect of metallurgical variables and environments on different forms of corrosion.	6
4	Corrosion Protection: Corrosion prevention method-anodic protection, cathodic protection, inhibitors.	4
5	Corrosion Testing : Electrochemical techniques-potentiostat, galvanostat, impedence. Thermogravimetric technique.	4
6	Hot Corrosion: High temperature oxidation of metals and alloys, laws governing oxidation, molten salt corrosion, liquid metal corrosion.	4
	Total	28

List of Practicals:

- 1. Corrosion rate measurements by potentiostat/Galvanostat.
- 2. Effect of alloying elements on passivity.
- 3. To study the effect of cathodic protection on given couple of metallic samples.
- 4. To study the effect of intergranular corrosion on microstructure of given metallic samples.
- 5. To study the Effect of various inhibitors on corrosion protection.
- 6. To study the effect of various atmospheric conditions on degradation of coatings.
- 7. To study the macro/micro structure of crevice corroded metallic samples.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Fontana, M.G., and Greene, N.D., "Corrosion Engineering", McGraw Hill.	2005
2	Uhling, H.H. and Revie, R.W., "Corrosion and Corrosion Control", 3 rd Ed., John Wiley & Sons,	1985
3	"Corrosion", Metals Handbook, Vol.13 A&B, 9th Ed., ASM	1989

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-303	Course Title: Mechanical Behaviour of Materials
2. Contact Hours:	L; 3 - ; T: 1; P: 2/2
3. Examination Duration (Hrs): Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS:	1 5 PRS: 1 5 MTE: 3 0 ETE: 4 0 PRE: 0 0
5. Credits: 0 4	6 Semester: Autumn

7. Pre-requisite: MT-205

8. Subject Area:

DCC

9. Objective

The course is aimed at developing an understanding of the response of engineering materials (metals, ceramics, polymers and composites) to mechanical loading.

S.No.	Contents	Contact Hours
1	Introduction, Stress and strain relations, mechanical testing, elastic behavior, and viscoelasticity.	5
2	Elements of plasticity, the flow curve, Strain hardening, Strain rate and temperature dependence of flow stress.	4
3	Plastic deformation, slip in crystals, dislocations, and dislocation motion.	5
4	Twins, strengthening mechanisms, grain boundaries, solid solution strengthening, strengthening from fine particles and strain hardening.	5
5	Fracture, types of fracture, brittle fracture, Griffith theory of brittle fracture of material, ductile fracture, notch effects, and fracture mechanics.	6
6	Fatigue, the S-N curve, low cycle fatigue, structural features, surface effects, and metallurgical variables.	5
7	Creep, the creep curve, stress rupture test, structural changes, creep mechanisms, DMM (deformation mechanism maps), and superplasticity	6
8	Embrittlement, residual stresses, mechanical behavior of Ceramics, glasses, polymeric materials, and composite materials.	6
	Total	42

List of Practicals:

- 1. To determine the hardness of given metallic sample by Brinnel Hardness Testing Machine.
- 2. To determine the hardness of given metallic sample by Vickers Hardness Testing Machine.
- 3. To determine the hardness of given metallic sample by Rockwell Hardness Testing Machine.
- 4. To determine the tensile properties of given steel and cast iron samples.
- 5. To determine the impact strength of low, medium, high carbon steel and cast iron by Izod and Charpy method.
- 6. To study the deep draw-ability of given metallic sheet samples by Ericson method.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Dieter, G. E., "Mechanical Metallurgy", 3 rd Ed., McGraw Hill.	1988
2	Courtney, T.H., "Mechanical Behavior of Materials", 2 nd Ed., McGraw Hill.	1990
3	Meyers, M.A. and Chawla, K.K., "Mechanical Behavior of Materials", Prentice Hall.	1999
4	Hull, D. and Bacon, D.J., "Introduction to Dislocations", 4 th Ed., Butterworth-Heinemann.	2001
5	Honeycombe, R.W.K., "The Plastic Deformation of Metals", Edward Arnold.	1977

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-304	Course Title: Iron Making
2. Contact Hours:	L: 3;
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
1. Relative Weightage: CWS: 2	5 PRS: 0 0 MTE: 2 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 4	6 Semester: Spring

7. Pre-requisite: MT-204

8. Subject Area: DCC

9. Objective: To provide knowledge about iron making from iron ore through blast furnace route and an elementary idea about alternative methods of iron production.

Sl.	Contents	Contact
No.		Hours
1	Resources, characteristics and functions of Indian iron Ores, metallurgical coal and limestone; factors governing values of iron ore, coal, and lime stone for use in blast-furnace.	5
2	Significance of agglomeration, processes of agglomeration; Principles and mechanism of production of sinters and pellets from iron ore fines; Self-fluxing and super-fluxing agglomerates.	7
3	Basic principles and design aspect of blast furnace (BF), Cowper hot stove; design and working of BF gas cleaning system; Conditions prevailing in different zones of B.F. and refractories required.	7
4	Physico-chemical principles of production of iron in BF: C-O, Fe-C-O, and Fe-H-O equilibria; Direct and indirect reduction reactions; comparison of CO and H2 as reducing agents on BF performance; Change in the composition of BF gas at different levels of BF; Stages of slag formation in blast-furnace and effects of Al ₂ O ₃ , MgO and CaO on slag behaviour; Physical chemistry of desulphurization; External desulphurization and its importance; Combustion race-way and effects of its shape and size on BF performance.	12
5	Irregularities in B.F. and their remedies: Hanging, scaffolding, breakout, channeling, and pillaring.	5
6	Auxiliary fuel injection, lime-injection, high-top pressure operation, humidification of blast; their effects on BF performance.	3
7	Elementary idea of alternative methods of iron making; Brief outline of direct reduction methods, their scope and limitations with reference to Indian conditions.	3
	Total	42

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Basforth, G. R., "The Manufacture of Iron and Steel", Vol I,II and III, Chapman and Hall Ltd.	1965
2	Tupkare, R.H., "Introduction to Modern Iron Making", Khanna Publishers.	2004
3	Biswas, A.K., "Principles of Blast-furnace Iron Making", S.B.A. Publications.	2005
4	Ward, R.G., "Physical Chemistry of Iron and Steel Making", ELBS, Edward Arnold Publishers	1999
5	Bodsworth, C., "Physical Chemistry of Iron and Steel Making", ELBS, Edward Arnold publishers.	1988

NAME OF DEPART	MENT	: Metallurgical and Materials Engineering
1. Subject Code:	MT-305	Course Title: Modelling and Simulation of Metallurgical Processes
2. Contact Hours:		L: 2 ; T: 0; P: 2
3. Examination Dura	•	Theory: 0 2 Practical: 0 0
4. Relative Weightag	e: CWS:	1 5 PRS: 1 5 MTE: 3 0 ETE: 4 0 PRE: 0 0
5. Credits: 0 3		6 Semester: Autumn
7. Pre-requisite: MA	-101 and MA	-102 8. Subject Area: DCC

9. Objective: To familiarize the students about the fundamentals of process modeling and simulation of metallurgical interest.

S.No.	Contents	Contact Hours
1	Introduction: Classification, functions, physical modeling: Similarity criteria, materials for modeling, measurement techniques,	8
2	Mathematical Modeling: Transport equations, applications for the metallurgical processes like- Continuous casting, hot rolling, powder rolling, powder pressing, and melt spinning.	8
3	Mathematical Modeling of Plants: Nominal model, calibration model, building models and model complexity, state space models, linear and nonlinear models, linearization, linear continuous time models, Laplace transformation, transfer functions, poles, zeroes, and time responses essentials of SISO control loop, stability and polynomical analysis. PD, PI and PID control.	6
4	Neural Network: Neural architecture, knowledge representation, learning process-error correction, and back propagation learning algorithm.	3
5	Fuzzy Logic Modeling: Fuzzy concepts and fuzzy variables, fuzzy sets, membership functions, controllers and relationship with PI, PD and PID control, Developing fuzzy models, and template based fuzzy system modeling.	3
	Total	28

List of Practicals:

- 1. Software development to determine temperature variation in a metal droplet moving in a gas stream.
- 2. Software development to determine temperature variation in a slab/strip during hot rolling.
- 3. Software development to determine porosity in powder metallurgy compact using neural network.
- 4. Software development for continuous casting of steel using neural network/fuzzy logic/genetic algorithm/conventional techniques.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Goodwin, G.C. and Graebe, S.F. and Salgado, M.E., "Control System Design", Prentice Hall of India.	2002
2	Yager, R. R. and Filer, D.P., "Essentials of Fuzzy Modeling and Control", John Wiley and Sons.	2002
3	Haykin, S., "Neural Networks- A Comprehensive Foundation", Pearson Education.	1999

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-306	Course Title: Phase Transformation and Heat Treatment
2. Contact Hours:	L: 3; T: 0; P: 2
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 3
4. Relative Weightage: CWS: 1	5 PRS: 1 5 MTE: 1 5 ETE: 4 0 PRE: 1 5
5. Credits: 0 4	6. Semester: Autumn
7. Pre-requisite: MT-205	8. Subject Area: DCC

9. Objective: To develop understanding of phase transformations in metals and alloys by using concepts of thermodynamics and diffusion kinetics and to give insight into the microstructural changes occurred due to different heat treatments.

Sl.No.	Contents	Contact Hours
1	Thermodynamics and Phase Diagram: Equilibrium, binary solutions, equilibrium in heterogeneous systems, kinetics of phase transformations	4
2 .	Diffusion : Atomic mechanism, interstitial and substitutional diffusion, atomic mobility, tracer diffusion in binary alloys, and diffusion in multiphase binary systems.	4
3	Crystal Interfaces and Microstructure: Interfacial energy, solid/vapour interfaces, boundaries in single phase solids, interphase interfaces in solids, and interface migration.	6
4	Diffusional Transformation : Homogeneous nucleation in solids, heterogeneous nucleation, precipitate growth, overall transformation kinetics — TTT diagram, precipitation in age-hardening alloys, the precipitation of ferrite from austenite, eutectoid transformation, pearlite and Bainite transformation in steel, continuous cooling diagrams, massive transformations, and order-disorder transformation.	10
5.	Diffusionless Transformations : Characteristics of diffusionless transformation, martensite crystallography, theory of martensite nucleation, martensite growth, and pre-martensite phenomena.	4
6	Hardenability of Steels: Concept of critical diameter, joining-end- quench test, effect of parameters viz: alloying elements, carbon content,	4

7 .	austenic grain size, section size and quenching media. Heat Treatment Processes: Various methods of heat treatments, surface hardening treatments, heat treatment of non-ferrous alloys, heat treatment	10
	schedules/case studies of some important steels and special types of treatments viz. martempering, austempering and thermo-mechanical treatments and intercritical treatments.	
	Total	42

List of Practicals:

- 1. Metallographic examination and hardness determination of as received steel samples (low, medium and high carbon steels)
- 2. To study the effect of various heat treatments viz annealing, normalizing, quenching and tempering on low, medium and high carbon steel samples.
- 3. To study the effect of case carburizing on microstructure and hardness of steel samples.
- 4. To determine the hardenability of given steel by Jominey-End-Quench test method.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Porter, D.A. and Easterling, K.E., "Phase Transformations in Metals and Alloys", Chapman and Hall.	1997
2	Reed-Hill, R.E. and Abbaschian, R., "Physical Metallurgy Principles", The PWS-KENT Series in Engg.	1992
3	Lakhtin, Y., "Physical Metallurgy", Mir Publisher.	1975
4	"Heat Treatment", Metals Handbook, Vol.4, 9th Ed., ASM	1989

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-307	Course Title: Metal Processing
2. Contact Hours:	L: 3 ; T: 0; P: 2
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weight age: CWS:	5 PRS: 1 5 MTE: 3 0 ETE: 4 0 PRE: 0 0
5. Credits: 0 4	6 Semester: Autumn
7. Pre-requisite: MI-102	8. Subject Area: DCC

9. Objective: To familiarise the students with fundamentals of metal casting principles and metal joining processes and their applications.

Sl.No.	Contents	Contact Hours
1	Introduction: Application and advantages of metal casting processes and classification of foundries. Types of patterns, patterns materials patterns allowances, and color codification.	5
2	Mould and Core Making: Ingredients: sand, clay, binders, and moisture, characterization of moulding sand, general requirement, and sand conditioning. Classification of moulding methods on the basis of technique used viz., green sand, dry sand, CO ₂ process, shell moulding etc. and machine moulding, floor and pit moulding.	6
3	Gating and Risering: Various component of gating system, types of gates, principle and practice of gating and risering for various metals and alloys, methods for directional solidification, plane front vs denderitic solidification, and concept of constitutional supercooling.	5
4	Special Casting Processes: Die casting, centrifugal casting, squeeze casting and reocasting etc.	· 2
5	Casting Defects Causes and Remedies: Defects arise due pattern, moulding sand and due to improper designing of gating system viz lap, scab, rattail, pipe, and porosity.	3
6	Melting furnaces and melting practice: Various types of furnaces used in foundry technique, melting practice of cast iron, steel, Al and its alloys and Cu and its alloys.	4
7	Welding: Scope, welding Processes: Gas, arc, resistance and plasma welding, welding consumables: flux, gas, electrode, fillet weld, butt weld, weld thermal cycle, weld testing, microstructure, and weld properties. Scope of micro-joining, resistance-spot, laser welding, ultrasonic welding, micro-plasma welding, and properties. Mechanisms of soldering and brazing, brazing materials, and properties.	8

8	The Use of Adhesives for Making Structural Joints: Scope and applications, bonding between adhesive and substrate, polymers, the properties of adhesive polymers, bonding procedures, joints design and applications.	3
9	Similar and Dissimilar Joining of Ceramics: Scope of applications, properties of ceramics, ceramic-ceramic joining, ceramic-metal joining, brazing materials, mechanical and materials aspects,	2
10	Mechanical Joining: Riveting, bolting, fastening, designing of joints, and properties	2
11	Application Oriented Comparison of Joining Processes: Properties of different joints, efficiency of joints, merits and demerits, maintenance, life and reliability	2
	Total	42

List of Practicals:

- 1. Preparation of green sand mould.
- 2. To estimate AFS grain fineness number for dry silica sand.
- 3. To estimate the clay content in the sand.
- 4. To estimate the moisture content in the green sand.
- 5. To estimate the permeability of the green sand.
- 6. To estimate hardness, compressive, shear and tensile strength for core sand.
- 7. To estimate refractoriness of the sand.
- 8. To study the effect of gas and arc welding processes on microstructure and hardness of given steel samples
- 9. To study the effect of various parameters of soldering and brazing processes on strength of joint.
- 10. To study the effect of TIG and MIG welding processes on microstructure and hardness of given metallic samples.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Heine, R.W., Loper C.R. and Rosenthal, P.C. "Principles of Metal Casting", Tata McGraw-Hill	2002
2	Mukhrjee, P.C., "Fundamentals of Metal Casting Technology", Oxford and IBH Publishing Company	1996
3	"Welding and Brazing", Metals Handbook, Vol. 6, ASM	1996
4	Parmar, R.S., "Welding Engineering and Technology", Khanna Publishers.	2002

NAME OF DEPARTMENT:

Metallurgical and Materials Engineering

1. Subject Code:

MT-401

Course Title: Composites and Polymers

2. Contact Hours:

L: 3

T: 1;

P: 0

3. Examination Duration (Hrs):

Theory:

Practical:

4. Relative Weightage: CWS:

PRS: 0 0

MTE: 2 5

PRE: 0 0 ETE: 5 0

5. Credits:

6. Semester:

Autumn

7. Pre-requisite: MT-205 and MT-303

8. Subject Area: DCC

9. Objective: To impart basic knowledge about the preparation, characterisation, application and reliability of the composites and engineering polymers.

Sl. No.	Contents	Contact Hours
1	Introduction: Definition of composite material, constituents of	, 3
	composites, interfaces and interphases; Distribution of	
	constituents and their synergy; Classification of composite -	·
	continuously and discontinuously reinforced composites, metal,	
	ceramic and polymer based composites.	
2	Fabrication of Composites: (a) Metal matrix composites – solid	6
·	state and solidification processing routes; Powder metallurgy	·
	route. (b) Ceramic matrix composites - vapour deposition; liquid	
	phase method; hot pressing or ceramic method. (c) Polymer	
	matrix composites and (d) Nano-composites.	
3	Characterisation of Composites: Control of particle distribution	5
	and defects, particle segregation; microstructure; particle-	
	matrix interfacial reactions; Mechanical properties; Composite	·
•	models; Fibre strengthening; Fracture behaviour of fibre	
	reinforced composite.	·
4	Joining of Composites: Induction heating, fusion bonding,	4
	ultrasonic welding, gas tungsten arc welding, gas metal arc	
	welding, resistance spot and seam welding, resistance brazing,	
	resistance spot joining, resistant spot brazing, resistance welding	1
	of thermoplastic- composite, weld bonding, brazing of MMC	

5	Application of Composite Materials: Civil constructions of	2
	structures/pannels, aerospace industries, automobile and other	
	surface transport industries, packaging industries, house hold	
	and sports components.	!
6	Fracture mechanics and safety of composite: Griffith theory of	5
	brittle fracture and modification for structural materials, basic	
	fracture mechanics of composite (fracture toughness, COD and	
	J-integral approaches, fatigue crack growth rate).	
7	Introduction and classification and synthesis of polymers: Basic definitions and nomenclature, molar mass and degree of polymerization, synthesis, glass transition temperature and crystallinity in polymers, structure and its relation to thermal, chemical, electrical and optical properties.	6
8	Mechanical and thermomechanical characteristics: General characteristics, viscoelasicity, deformation behaviour of elastomers, deformation mechanisms, fractures, and toughened polymers.	5.
9	Polymer processing, characterisation and applications: Introduction, plastics, elastomerics and fibres, compounding and processing techniques, practical aspects of polymer blending, standards and engineering applications of polymers.	6
	Total:	42
i		I

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Chawala, K.K., "Ceramic Matrix Composites", Chapman & Hall, London	1993
2	"Composites", Metals Handbook, Vol. 21, 9th Ed., ASM.	1989
3	Rudin, A., "The Elements of Polymer Science and Engineering", Academic Press.	1999
4	Young, R.L. and Lovell, P.A., "Introduction to Polymers", Stanley Thornes Publishers Ltd.	2000

NAME OF DEPARTMENT:

Metallurgical and Materials Engineering

1. Subject Code:

MT-402

Course Title: Mechanical Working of Metals

2. Contact Hours:

L: 3

T: 0;

P: 2

3. Examination Duration (Hrs):

Theory:

3

Practical:

4. Relative Weightage: CWS:

PRS:

MTE: 3 0 1 5

ETE: 4 0

PRE: 0 0

5. Credits:

6 Semester:

Spring

7. Pre-requisite: MT-303

8. Subject Area: DCC

9. Objective: To introduce various metal forming processes, mechanics involved and effect of

these on structure and properties of metals.

Details of the Course: 10.

Sl. No.	Contents	Contact Hours
1	Fundamental of Working: Yield criteria, Levi-Mises equation, classification of forming processes, mechanics of metal working, temperature in metal working hot working vs. cold working, strain rate effects, sliding and sticking friction, recovery and recrystallization, preferred orientation.	6
2	Forging: General aspects, closed-die and open-die forging, spread law, different types of drop and press forging equipment, forging loads, forging defects, case studies.	5
3	Rolling: Terminology of rolled products, different kinds of rolling mills, forces and geometrical relationship in rolling, draft Ekelund's expression for no-slip angle, forward slip, rolling variables, problem and defects in rolled products, roll pass sequences used in rolling of blooms to billets and various structural shapes, billets to bars or rods, roll pass	8
4	Extrusion: Direct and indirect extrusion, impact extrusion and Hooker process, hydrostatic extrusion, equipment, extrusion variables, derivation of extrusion pressure, deformation, lubrication and defects in extrusion, production of seamless tubes by extrusion process.	6
5	Rod, Wire and Tube Drawing: Processes and equipments, hydrodynamic lubrication, maximum possible reduction in a pass under ideal condition, draw stress with friction and back tension, effect of diameter, angle and mode of flow on drawing stress, shaving defects.	

6	Other Conventional Processes: Piercing, punching and blanking, stretch forming spinning, embossing and coining, powder performs forging.	5
7	High Energy Rate Forming Processes: Brief description of explosive forming, electrical discharge forming and electro magnetic forming.	6
	Total	42

List of Practicals:

- 1. To study the various equipments used for mechanical working metals.
- 2. To study the effect of type of loading viz hammer forging and press forging on microstructure and hardness of given metallic sample.
- 3. To study the effect of amount of deformation by rolling on microstructure and hardness of given metallic sample.
- 4. To study the effect of wire drawing on microstructure and hardness of given metallic samples.

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
_1	Dieter, G.E., "Mechanical Metallurgy", Tata-McGraw Hill	1988
2	Rowe, R., "Principles of Metal Working," Edward Arnold Publications Ltd.	1965
3	Wilson, F.W., "High Velocity Forming of Metals", Prentice-Hall.	1980
4	"Metal Working", Metals Handbook, Vol.14 A&B, 9th Ed., ASM.	1989

NAME OF DEPARTMENT: Metallurgical and Materials Engineering

1. Subject Code: MT-403 Course Title: Steel Making

2. Contact Hours: L: 3 ; T: 1; P: 0

3. Examination Duration (Hrs): Theory: 0 3 Practical: 0 0

4. Relative Weightage: CWS: 2 5 PRS: 0 0 MTE: 2 5 ETE: 5 0 PRE: 0 0

5. Credits: 6. Semester: Autumn

7. Pre-requisite: MT- 204 and MT-304 8. Subject Area: DCC

9. Objective: To familiarize the students to the basic fundamentals and physico-chemical principles and practice of steel production.

Sl. No.	Contents	Contact Hours
1	Principle of Steel Making: Thermodynamics and physico-chemical	10
	aspect of C, S, P, Si and Mn removal reactions of hot metal, mechanism	
	and kinetics of C-O reactions, deoxidation of steel.	
2	Conventional and Modified Methods: Bessemer Process; Open-hearth process: design, construction and working of acid and basic furnaces;	10
	Electric steel making: types of electric steel making furnaces, their	
	construction, refractory lining and working.	
. 3	Modified Methods and Recent Processes: Duplex, triplex, cupola-open	12
	hearth process; Role of oxygen in steel making, L.D., KALDO, Rotor,	
	Ajax, LDAC, OLP; Vacuum induction melting; Suitability of the above	
	processes under Indian conditions w.r.t. integrated steel plant and heavy	
	foundries.	
4	Secondary Steel Making: Concept of duplexing, ASEA-SKF process,	6
	VAD process, Ladle furnace process, electroslag refining, stainless steel	
	making practice. AOD, VOD and CLU processes.	
5 .	Casting Pit Practice: Ingot moulds, methods of teeming steel, killed,	4
	semi killed, rimming steels ingots and continuous casting of steels.	
	Total	42

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
ī	Bashforth, G.R., "The Manufacture of Iron and Steel", Vol II, Chapman & Hall Ltd.	1965
2	Bodswarth, C., "Physical Chemistry of Iron and Steel Manufacture", ELBS, Edward Arnold Publications Ltd.	1988
3	Ward, R.G., "Physical Chemistry of Iron and Steel Making", ELBS, Edward Arnold Publications Ltd.	1999
4	Tupkari, R.H., "Introduction to Modern Steel-making", Khanna Publishers.	2004

NAME OF DEPARTMENT: Metallurgical and Materials Engineering 1. Subject Code:

2. Contact Hours: L: 3 T: 1; P: 0

3. Examination Duration (Hrs): Practical: Theory: 0 0 3

MTE: 2 5 4. Relative Weightage: CWS: 2 5 PRS: PRE: 0 ETE: 5 0

Course Title: Non-Ferrous Technology

5. Credits: 6 Semester: Autumn

MT-405

7. Pre-requisite: MT-204 8. Subject Area: DCC

9. Objective: To familiarize the students to extraction and refining of the common non-ferrous metals including their pyro-, hydro-, and electro-metallurgical aspects.

Sl.No.	Contents	Contact Hours
1	Copper: Roasting for Pyro and Hydro Metallurgical routes, Matte smelting, converting, fire-refining, electro refining, leaching and electrowinning.	9
2	Lead: Principles and process of roasting of concentrate; Blast furnace: design, principles and process of smelting, effects of presence of Cu and Zn on roasting and smelting practice; Refining of lead.	6
3	Zinc: Principles and processes of roasting for pyro-and hydro-metallurgical routes; Principles and processes of smelting; Imperial smelting; Leaching of concentrate, Jarosite precipitation, purification of leach-liquor; Electro-winning.	7
4	Magnesium: Sources of Mg, recovery of MgCl2 from chloride sour ces, electrowinning of Mg from Mgcl2; principles and processes of production of Mg from oxide sources, Hansgrig process, Pidgeon process and Magnetherm processes; Conversion of MgO to MgCl ₂ .	7
5	Aluminium: Sources of Al; Effect of mineralogical forms and chemical composition of bauxite on production of alumina; Principles and process of production of alumina by Bayer technique; principles and process of production of Al by Hall-Heroult technique, anode effect, its causes and remedial measures; factors effecting cell performance; Refining of Al.	8
6	Titanium: Sources of Ti, ore break-down processes, production of Ti from TiO ₂ ; Conversion of TiO2 into TiCl ₄ ; Principles and process of production of Ti from TiCl ₄ , Kroll process; Refining of Ti by Van Arkel deBoer process.	
	Total	42

S. No.	Name of Authors /Books/ Publisher	Year of Publication/ Reprint
1	Dennis, W.H., "Non-ferrous Metallurgy", Sir Issac Pitman & Sons Ltd.	1980
2	Bray, J.L., "Non-ferrous Production Metallurgy", John Wiley & Sons.	1985
3	Biswas, A.K. and Davenport, W.G., "Extractive Metallurgy of Copper", Pergamon Press.	2002
4	Embey, E.F., "Principles of Magnesium Technology", Pergamon Press.	1981
5	Mantell, C.L., "Electro-chemical Engineering", Tata-McGraw Hill	1999

NAME OF DEPARTMENT

Metallurgical and Materials Engineering

1. Subject Code: MT-310

Course Title: Advances in Foundry Technology

2. Contact Hours:

L: 3

T: 0;

P: 0

3. Examination Duration (Hrs):

Theory: 3 Practical:

4. Relative Weightage: CWS:

PRS:

0 0 MTE: 3 5 ETE: 5 0 PRE:

5. Credits:

6 Semester:

Spring

7. Pre-requisite: MT-307

8. Subject Area: DEC

9. Objective: To provide advances in the field of casting technology.

Sl. No.	Contents	Contact Hours
1	Introduction: Comparison of various forming process with foundry technique, classification of foundries, scope of foundry industry in India	2
2	Patterns: Functions of patterns, type and applications of patterns, considerations in design of patterns for a given geometry, pattern allowances and materials.	4
3	Moulding Materials: Metallic materials such as cast iron, alloy steels and copper base alloys, their constitution properties and applications; Non-metallic materials such as moulding sands: General characteristics, ingredients and their effect on properties. Sources of foundty sands in India, testing of sands and sand conditioning.	6
4	Cores: Types of cores, core sands, special additives, preparation of cores, core baking and finishing.	2
5	Moulding Process: Metallic moulding, non-metallic moulding such as plaster of paris molding, ceramic moulding and sand moulding processes; depending upon the equipment used; recent moulding techniques and materials.	6
6	Gating and Feeding of Casting: Gating system, requirements of an ideal gating and feeding systems, types of gates; functions of risers, chvorinor's equation, types of risers; computer application for designing of gating system and risers for ferrous and non-ferrous castings.	

7	Melting Practice: Melting furnace used in foundry; foundry	6
	considerations in melting non ferrous metals such as Al, Cu and Zn alloys;	
	melting process and problems encountered during melting of steels and	
	cast irons; familty of cast irons and their founding;	
8	Special Casting Processes: Pressure die casting, centrifugal casting,	4
	investment casting and continuous casting processes.	
9	Casting Defects and Fettling: Common casting defects their causes and	4
	remedies, fettling and inspection, heat treatment of castings.	
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Heine, R.W., Loper, C.R. and Rosenthal, P.C., "Principles of Metal Casting", Tata McGraw-Hill.	2002
2	Mukhrjee, P.C., "Fundamentals of Metal Casting Technology" Oxford & IBH Publishing Co.	1996
3	"Casting", Metals Handbook, Vol.15, 9th Ed., ASM.	1989

NAME OF DEPARTMENT : Metallurgical and Materials Engineering

1. Subject Code: MT-311 Course Title: Engineering Materials

2. Contact Hours: L: 3 ; T: 0; P: 0

3. Examination Duration (Hrs): Theory: 0 3 | Practical: 0 0

4. Relative Weightage: CWS: 1 5 PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0

5. Credits: 0 3 6 Semester: Autumn

7. Pre-requisite: MT-205 8. Subject Area: DEC

9. Objective: To impart knowledge of materials used for different engineering applications such

as structural, electrical, magnetic and high temperature.

Sl. No.	Contents	Contact Hours
1	Introduction: Materials selection and related factors such as design,	4
· · · · · · · · · · · · · · · · · · ·	processing and economics; case histories related to material selection.	
2	High Temperature Materials: Refractory metals and alloys, creep	8
	resistant alloys, typical creep curve: effect of load, temperature and	
	composition; super alloys: Fe-base, Ni-base and Co-base super alloys and	
	their composition, heat treatment, properties and applications.	<u> </u>
3	Low Temperature Materials (Cryogenic Materials): Concept of ductile	8
	to brittle transition, determination of transition temperature; Ni-base low	
į	temperature materials, 9% Ni-steel, austenitic stainless steels, their	
	composition, heat-treatment, properties and applications.	
4	Structural Steels: Plain carbon, high strength low alloys (HSLA), micro	8
	alloyed steels, and dual-phase steels their composition, heat treatment,	,
	properties and applications.	
5	Tool and Die Steels: Plain carbon and alloy tool and die steels: high speed	6
	steel, high carbon-high chromium die steel, their composition heat	
	treatment, properties and applications.	
6	Cu base and Al base Non Ferrous Metals and Alloys: Brasses, bronzes,	5
1	duralumin, and Al-Si alloy their composition, properties and applications.	
7	Engineering Polymers and Ceramics: Thermoplastic, thermosetting	3
	polymers and elastomers; High strength engineering ceramics.	
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Callister, W.D. Jr., "Material Science and Engineering -An	2000
	Introduction", 5th Ed. John Wiley and Sons	
2	Askland, R.A., "The Science and Engineering of Materials", 2nd Ed.,	1989
	PWS-KENT Publishing Company.	
3	Lakhtin Y., "Engineering Physical Metallurgy", Mir Publishers	1992
4	Avner S.H., "Introduction to Physical Petallurgy", McGraw Hill Book	2005
	Co.	
5	Raghavan, V., "Materials Science and Engineering: A First Course",	2004
	5th Ed., Prentice-Hall of India.	

NAME OF DEPARTMENT

Metallurgical and Materials Engineering

1. Subject Code:

MT-312

Course Title: Advances in Metal Joining

2. Contact Hours:

L: 3

T: 0;

P: 0

3. Examination Duration (Hrs):

Theory:

0 3

Practical: 0 0

4. Relative Weightage: CWS: 1 5 PRS:

0 0 MTE: 3 5

5. Credits:

6 Semester:

Spring

7. Pre-requisite: MT-307

8. Subject Area:

DEC

9. Objective: To impart the knowledge of advances in various techniques of metal joining.

Sl. No.	Contents	Contact Hours
1	Review of Principles of various joining methods for metals and non- metals namely adhesive joining, welding, brazing, soldering and mechanical joining processes.	4
2	Heat Flow in Welding: Temperature distribution in welding, heat flow in fillet and circular welds, heat flow in resistance welding and electroslag welding.	3
3	Solidification of Weld Pool: Solidification structure such as dendritic, columnar and equiaxed and their effect on mechanical properties, heat affected zone (HAZ) its structure and effect on mechanical properties; Post welding heat treatment.	8
4	Joining Methods: Fusion welding processes, solid phase welding, gas welding, arc welding, TIG welding, MIG welding, Submerged arc welding, electro slag, welding, electron beam welding, laser welding, cold pressure welding, ultrasonic welding, friction welding and explosive welding; brazing and soldering; principles involved, process details and equipment used for above welding processes.	10
5	Defects in Weldments: Classification of weld defects, causes and remedies in arc welding, resistance welding, friction welding and other welding processes.	4
6	Weld Inspection and Quality Control: Visual inspection, destructive tests, non-destructive tests (NDT), pressure and leak testing, and NDT of adhesive joints.	

7	Weldability: Weldability assessment, weldability tests, weldability of	8
	carbon steels, HSLA steels, stainless steels, high alloy steels, cast irons,	
}	nickel alloys, copper alloys, aluminium alloys, magnesium alloys and	
	titanium alloys.	
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	"Metallography and Microstructures", Metals Handbook, Vol. 9, ASM.	2000
2.	"Welding Handbook", Vol. 3, Part 2, 9 th Ed. The American Welding Society.	2000
3	Kalpakjian, S., Steven R.S., "Manufacturing Engineering and Technology", Prentice Hall.	2001
4	Parmar, R.S., "Welding Engineering and Technology", Khanna Publishers.	2002

NAME OF DEPAR	FMENT	: Metallurgical and Materials Engineering			
1. Subject Code:	MT-313	Course Title: Tribology of Engineering Materials			
2. Contact Hours:	•	L: 3 ; T: 0; P: 0			
3. Examination Dur	ation (Hrs):	Theory: 0 3 Practical: 0 0			
4. Relative Weightage: CWS: 1 5 PRS: 0 0 MTE 3 5 ETE 5 0 PRE: 0 0					
5. Credits: 0 3		6 Semester: Autumn			

7. Pre-requisite: MT-205

8. Subject Area: DEC

9. Objective: To familiarize the students about the cause of friction and wear; and the methods to minimize wear in engineering components.

Sl. No.	Contents	Contact Hours
1.	Surface Properties and Surfaces in Contact: Nature of metallic surface, surface geometry, measurement of surface topography (roughness), quantifying surface roughness, contact between surfaces; friction, the laws of friction, measurement of friction, origin of friction, theories of friction adhesion: theory, extension of the adhesion theory.	8
2.	Wear: Types of wear, adhesive wear, Archard's law, abrasive wear, erosion wear, factors affecting wear, wear map, various wear testing methods: pin on disc, pin on drum, slurry wear, air jet and water jet erosion as per ASTM standards	9
3.	Tribological Properties of Solid Materials: Hardness, strength, ductility and work hardening rate, effect of crystal structure, effect of microstructure, mutual solubility of rubbing pairs and effect of temperature.	9
4.	Surface Treatments to Reduce Wear: Surface treatments without change of composition and with change of composition application of surface coating by welding, by flame, by spraying, by plasma spraying, by electroplating and by electroless coating, by chemical vapour absorption (CVD) and by physical vapour deposition (PVD).	9

5.	Development of Wear Resistant Materials: Alloy design and development of dual-phase steels, 13/4 martensitic stainless steels, Nitronic steel, their characterization and evaluation of wear resistance properties.	7
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Hutchings, I.M. "Tribology – Friction and Wear of Engineering Materials", Edward Arnold Publications Ltd.	1992
2	Arnold, R.D., Davies, P.B., Halling, J. and Whomes, T.L., "Tribology-Principles and design applications", Springer – Verlag.	1991
3	Bhusan, B., "Introduction to Tribology", John Wiley and Sons.	2002

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT- 314	Course Title: Powder Metallurgy
2. Contact Hours:	L: 3 ; T: 0; P: 0
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS 1 5	PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 3	6. Semester: Spring
7. Pre-requisite: MT-209	8. Subject Area: DEC

9. Objective: To introduce the advanced concepts of powder metallurgy with special reference

10. Details of the Course:

to recent developments.

Sl.No.	Contents	Contact Hours
1	Review of basic features of powder metallurgy (P/M), scope and limitations of P/M technique.	3
2	General characteristic of metal powders, bulk properties of metal powders: particle shape, flow, apparent density, specific surface area, particle size distribution and their determination, different methods of production of metal powders.	5
3	Compressibility and compactibility of powders, different compacting processes and their impact on the properties of green compact, continuous compacting of metal powders, powders perform forging, hot isostatic pressing.	5
4	Theory of sintering, sintering practice, furnaces and atmosphere control, activated sintering techniques, after sintering treatments; industrial sintering practice for various ferrous and non ferrous products	6
5	Classification of powder products, selected applications relating to dense structural components.	6
6	Case histories relating to applications of the products, development of technology under Indian condition.	5

	Application of powder metallurgy in the manufacture of self-lubricating bearing, magnetic and electrical materials, tungsten carbide tool bits, bearing materials, dispersion strengthen materials for high temperature	
	applications, and manufacture of diamond based cutting tools.	
8	Development of friction material through P/M route for clutch plate, and break pads for aero planes.	5
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Masuda, H., "Powder Technology Handbook", Taylor & Francis.	2006
2	German, R.M., "A to Z of Powder Metallurgy", Elsevier Advanced Technology.	2005
3	Sands, R.L. and Shakespeare, C.R., "Powder Metallurgy Practice and Applications", Newness Publication.	1970
4	"Powder Metal Technologies and Applications", Metals Handbook, Vol.7, 9th Ed., ASM.	1989
5	Hirschhorn, J.S., "Introduction to Powder Metallurgy", APMI, USA	1975
6	Upadhyaya, G.S., "Powder Metallurgy Technology", Cambridge Press.	1996

NAME OF DEPARTMENT

Metallurgical and Materials Engineering

1. Subject Code:

MT-315

Course Title: Introduction to Biomaterials

2. Contact Hours:

L: 3

T: 0;

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical:

4. Relative Weightage: CWS:

5 PRS:

MTE: 3

ETE: 5 0

5. Credits:

6 Semester:

Autumn

7. Pre-requisite: MT-205

8. Subject Area: DEC

9. Objective: Materials science and Engineering concepts are integrated with biology in order to

develop a good understanding of metallic implants.

Sl.No.	Contents	Contact Hours
1.	Introductory aspects and historical background.	1
2.	Metallic materials, definitions, steel, cobalt-chromium, titanium, new titanium alloys, shape memory alloys, niobium alloys, and tantalum alloys.	8
3.	Polymeric bio materials, definitions, types, structure property relations and their characterization.	3
4.	Ceramic biomaterials, definitions, types, structure and properties.	2
5.	Corrosion of metallic biomaterials, mechanism, pitting corrosion, fretting corrosion, crevice corrosion, intergranular corrosion, stress corrosion cracking, galvanic corrosion, fatigue and wear Corrosion, microbiological corrosion and some case studies	8

	standards for biocompatibility. Total	42
	methods of sterilization, assessment of sterility; device failure, analysis of retrieved implants, development of voluntary consensus	
	cardiovascular applications, grafts, catheters, valves, embolic agents, sterility and infection, interactions of bacteria with biomaterials,	6
8.	Applications of biomaterials and product development,	
	assessment of material performance.	-
	evaluating protein adsorption, inflammatory responses, acute inflammation, chronic inflammation, foreign body response, and	6
	structure, interaction of proteins with synthetic materials, methods for	
	materials, characterization of cell-material interactions, protein	
7.	Biocompatibility of biomaterials, cells, interactions with proteins and	
	plasma spraying, PVD, CVD, atomized liquid spray, dip, spin-on, sol-gel, and electrochemical techniques.	8
6.	Surface modification of implants, ion implantation, ion plating,	

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1.	Bhat, S.V., "Introduction to Biomaterial", Narsova Publications	2000
2.	Park, J.B., "Biomaterials", Addison-Wiley Publishing Company.	2001
3.	Ratner, B., "Biomaterials Science", Academic Press	2000
4.	Ratner, B.D., Hoffiman, A.S., Schoen, F.J. and Lemons, J.E. "An Introduction to Materials in Medicine", Academic Press.	2004

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT- 407	Course Title: Surface Engineering and Coating Technology
2. Contact Hours:	L: 3 ; T: 0; P: 0
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS: 1	5 PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 3	6 Semester: Autumn
7. Pre-requisite: MT 205	8. Subject Area: DEC

9. Objective: The course aims to provide an understanding of improvement in the surface properties of engineering materials for improved performance.

Sl.No.	Contents	Contact Hours
1.	Review of Surface Engineering Techniques: Carburising, nitriding, cyaniding, hot dipping, galvanizing, hard facing; Electrodeposition, anodizing, scope and application of conventional surface engineering techniques in engineering materials; advantages and limitations of conventional processes.	6
2	Surface Dependent Engineering Properties: Wear, friction, fatigue, and corrosion; Surface initiated engineering failures, importance and necessity of surface engineering	8
3	Classification and Scope of Surface Engineering in Metals: ceramics, polymers and composites, tailoring of surfaces of advanced materials. Surface modification: physical techniques, surface modification: chemical techniques, classification, methods, and technology.	8

4.	Modern Techniques in Surface Engineering: PVD, CVD, plasma spray coating; HVOF and detonation gun processes, ion implantation; Surface modification by directed energy beams like ion, electron and laser beams; energy transfer, compositional and microstructural aspects, and applications.	12
5.	Characterization: Microstructural and compositional method; Testing and evaluation of surface-properties, structure-property correlation; Evaluation of coatings for wear, erosion, and erosion- corrosion	8
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Bunshah R.F. (Ed.), "Handbook of Deposition Techniques for Films and Hard Coatings", 2 nd Ed., Noyes Publications.	1994
2	Burakowski, T., and Wierzchon, T., "Surface Engineering of Metals: Principles, Equipment, Technologies", CRC Publishers.	1998
3	"Surface Engineering", Metals Handbook, Vol. 5, 10th Ed., ASM.	1994
4	Sudarshan, S., (Ed.) "Surface Modification Technologies-An Engineers Guide", Marcel Dekker Inc.	1989
5	Grainger S. and Blunt J., "Engineering Coatings: Design and Application", 2 nd Ed., Woodhead Publishing.	1998
6	Pawlowaski L., "The Science and Engineering of Thermal Spray Coatings", John Wiley & Sons.	1995

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering		
1. Subject Code: MT- 409	Course Title: Characterization Techniques		
2. Contact Hours:	L: 2 ; T: 0; P: 2		
3. Examination Duration (Hrs):	Theory: 0 2 Practical: 0 0		
4. Relative Weightage: CWS: 1 5 PRS: 1 5 MTE: 3 0 ETE: 4 0 PRE: 0 0			
5. Credits: 0 3	6 Semester: Autumn		
7. Pre-requisite: MT-306	8. Subject Area: DEC		

9. Objective: To introduce structural characterization of materials at different scales and using different techniques to get the required information.

Sl.No.	Contents	Contact Hours
1.	Quantitative Light Microscopy: Introduction to light microscopy, concept of resolution, Airy rings, numerical aperture, magnification, depth of field, depth of focus, lens defects and their corrections, principles of phase contrast – bright-field and dark-field contrast, polarized light microscopy, Quantitative microscopy, estimation of grain size, grain boundary area, relevance of light microscopy ideas to electron microscopy.	8
2.	X-ray Diffraction and Diffractometry: Introduction, production of X-rays, crystal geometry, lattice directions and planes, zone axis, interplaner spacing and angle, Stereographic projection, Bragg's condition of diffraction, X-ray scattering, application of X-ray diffraction - estimation of grain size, particle size, macro texture, residual stress.	8
3.	Transmission Electron Microscopy (TEM): Principle, construction and operation of TEM, Interaction of electrons with specimen, camera constant, reciprocal space and lattice, Ewald sphere, diffraction from finite crystal, preparation of specimens, bright and dark field imaging, selected area diffraction, indexing of diffraction patterns, contrast from precipitates, dislocations and stacking faults.	8
4.	Scanning Electron Microscopy (SEM): Principle, construction and operation of SEM, microtexture (Electron Backscattered Diffraction), study of fractured surfaces, EDX, electron probe microanalysis.	
	Total	28

List of Practicals:

- 1. Preparation of sample for metallography. Steel and Brass
- 2. Grain size measurement and volume fraction of alpha-beta phases in brass
- 3. Demo of X-ray diffraction equipment
- 4. Indexing of the powder pattern
- 5. Demo of Transmission Electron Microscope
- 6. Indexing of the Selected Area Diffraction (SAD) patterns
- 7. Demo of Scanning Electron Microscope and EDS

S.No.	Name of Author/Book/Publisher	Year of Publication/ Reprint
1.	Goodhew, P.J., Humphreys J. and Beanland, R., "Electron Microscopy and Analysis", Taylor and Francis.	2001
2.	Gifkins, R.C., "Optical Microscopy of Metals", Sir Isaac Pitman and Sons.	1970
3.	Cullity, B.D., "Elements of X-Ray Diffraction", Addison-Wesley Publishing Company.	1980

Metallurgical and Materials Engineering NAME OF DEPARTMENT Course Title: Principles of Material Selection 1. Subject Code: MT-410 P: 0 2. Contact Hours: L: 3 T: 0; Practical: 3. Examination Duration (Hrs): Theory: PRS: | 0 | 0 MTE: 3 5 1 5 4. Relative Weightage: CWS:

5. Credits: 0 3 6. Semesters: Spring

7. Pre-requisite: MT-205 8. Subject Area: DEC

9. Objective: The course aims to provide various criteria for selection of materials for different applications along with case studies.

Sl.No.	Contents	Contact Hours
1	Selection criteria, service requirement, design fabricability, functionability, structure-property relationship reappraisal of the role of microstructure; crystal structure and defect structure vis-à-vis properties; materials and their applications, compositions, codes and properties	6
2	Applications of important ferrous materials like stainless steels, maraging steels, tool and die steels, high speed steels, and alloyed cast irons: their composition, heat treatment and properties.	8
3	Applications of important non ferrous metals like Cu base, Al base, Ti base and Mg base alloys: their compositions, heat treatment, and properties.	5
4	Some important composites like metal-matrix and composite, ceramic matrix composites: their composition, preparation, properties and their applications, some important structural ceramics.	6
5	Thermoplastic, thermo setting polymers and elastomers: structures, properties and specific applications.	6
6	Important wear resistant alloys for hydro and thermal power stations; low and high temperature materials.	7
7	Case studies highlighting selection of materials for specific applications.	4
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Raghavan, V., "Physical Metallurgy: Principles and Practice", 2 nd Ed., Prentice-Hall of India.	2007
2	Callister, W.D. Jr., "Material Science and Engineering-An Introduction", 5 th Ed., John Wiley and Sons	2000
3	Askland, R.A., "The Science and Engineering of Materials", 2 nd Ed., PWS-KENT Publishing Company.	1989
4	Raghavan, V., "Materials Science and Engineering: A First Course", 5 th Ed., Prentice-Hall of India.	2004

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-411 2. Contact Hours:	Course Title: Nanomaterials and their Applications L: 3; T: 0; P: 0
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS: 1	5 PRS: 0 0 MTE: 3 5 ETÉ: 5 0 PRE: 0 0
5. Credits: 0 3	6. Semester: Autumn

7. Pre-requisite: MT-201 and MT-205 8. Subject Area: DEC

9. Objective: This course aims to provide various concepts of nanosized materials, their preparation, properties and applications.

Sl.No.	Contents	Contact Hours
1	Historical Perspective: An overview of natural and classical nanosystems.	2
2	Classification and Nomenclature of Nanomaterials: Nanosized metals and alloys, semiconductors, ceramics - a comparison with respective bulk materials; organic semiconductors, carbon nanotubes, nanorods, nanocomposites consisting of organic, inorganic and biomaterials; zero-, one-, two-, and three dimensional nanostructures - quantum dots, quantum wells, quantum rods, quantum wires, quantum rings.	5
3	Novel Properties of Nanomaterials: Size and shape dependent optical, emission, electronic, transport, photonic, refractive index, dielectric, mechanical, magnetic, non-linear optical properties; catalytic and photocatalytic properties.	
4	Synthesis of Nanoparticulates: Nucleation and growth of nanosystems; Physical methods: mechanical milling, laser ablation, sputtering, microwave plasma etc.; Metallo organic chemical vapor deposition; designing of advanced integrated nanocomposites, functional nanomaterials and nanostructured thin films.	

	composites, polymers, ceramics, biomaterials, pharmaceuticals, optical, fluorescent, electronic, magnetic and photonic devices, future perspectives of nanotechnology in miniaturization of	
7	Applications and Perspectives of Nanomaterials: Development of nanoscale catalysts, photocatalysts, sensitizers, sensors,	5
	forging, accumulated roll bonding, equichannel angular pressing (ECAP), Hall-Petch effect, reverse Hall-Petch effect; Deformation mechanism: grain boundary sliding, rotation and nano-twinning.	
6	plasmon absorption, quantum confinement in semiconductors – particle in a box like model for quantum dots, origin of charge on colloidal sols and its implications in making building blocks. Bulk Nanomaterial: Severe plastic deformation, multiaxial	4
5	Theories of Nanosized Materials: Transition metal sols, origin of plasmon band, Mie theory, influence of various factors on the	9

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Poole, C.P., and Owens, F.J. (Eds.), "Introduction to Tanotechnology", John Wiley & Sons Inc.	2003
2	Rao, C.N.R., Müller, A., and Cheentham, A.K. (Eds.), "Chemistry of Nanomaterials", Wiley – VCH.	2004
3	Liz-Marzán, L.M., and Kamat, P.V. (Eds.), "Nanoscale Materials", Kluwer Academic Publishers.	2003
4	Rotello, V. (Ed.), "Nanoparticles - Building Block for Nanotechnology", Kluwer Academic/Plenum Publishers.	2004

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-412	Course Title: Special Steels and Superalloys
2. Contact Hours:	L: 3 ; T: 0; P: 0
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS:	5 PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 3	6.,Semester: Spring

7. Pre-requisite: MT-306 8. Subject Area: DEC

9. Objective: The course aims to provide substantial knowledge of various alloy steels and super alloys, their processing and applications.

Sl.No.	Contents	Contact Hours
1.	Introduction of Fe-C phase diagram, classification of alloying elements and their effect on Fe-C system.	4
2.	Phase transformation in steels effect of alloying elements, TTT diagram of plain C and alloy steels.	4
3.	Classification of stainless steels, martensite, ferrite, austenite and P-H stainless steels, their composition, heat treatment, mechanical properties, their weldability, formability and applications	6
4.	Higher strength low alloy (HSLA) steels, history of development and applications; ferrite steels, bainite steels and dual phase steels, mechanics of strengthening viz. grain refinement, solid solution strengthening, precipitation hardening, factors affecting strength, ductility and impact transition temperature.	8
5.	Ultra high strength steels, major requirements, maraging steels, TRIP steels, their types, heat treatment cycles, properties and applications.	6
6.	Tool steels, their types, alloying elements and their effects, classification, compositions, heat treatments, properties and applications.	6
7.	Super alloys, historical developments, physical metallurgy of Ni base alloys, Ni-Fe alloys, cobalt base alloys, chromium base alloys, microstructures, properties and weldability of super alloys.	8
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Pickering, F. B., "Physical Metallurgy and the Design of Steels", Allied Science Publishers Ltd.,	1978
2	Leslie, W.C., "The Physical Metallurgy of Steels", Tata McGraw-Hill.	1981
3	"Properties and Selection: Irons, Steels, and High Performance Alloys", Metals Handbook, Vol.1, , 9 th Ed., ASM.	1989

NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-413	Course Title: Principles of Solidification
2. Contact Hours:	L: 3 ; T: 0: P: 0
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS: 1	5 PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 3	6. Semester: Autumn
7. Pre-requisite: MT-307	8. Subject Area: DEC

9. Objective: This course aims to provide advanced concepts of solidification in materials and their applications in real life situations.

Sl.No.	Contents	Contact Hours
1	The liquid phase, crystal structure of liquid; Thermodynamics of	8
,	solidification processes, structure and characteristics of solid liquid interface, energies, kinetics and topography of interfaces.	
2	Defect formation, micro-, macro-, normal, inverse and gravity segregation; constitutional supercooling, crystal growth morphology such as plane front, cellular and dendritic.	7
3	Transport processes, mathematical modeling of heat flow and fluid flow in solidification of metal ingot, continuously cast billets, splat cooling, and atomization.	7
4	Solidification of single phase alloys like Cu-Ni, Pt-Au systems, solidification of multiphase alloys like Fe-C, Cu-Zn and Cu-Sn systems; Effect of rate of cooling on the microstructure and properties of single phase and multiphase alloys; Structure of cast ingot: chilled zone, columnar zone and equiaxed zone.	
5	Structure and properties of metals and alloys in special casting techniques such as rheocasting, compocasting and spray forming; Effect of vibration during solidification on structure and properties on metals and alloys.	
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Heine, R.W., Loper C.R. and Rosenthal, P.C. "Principles of Metal Casting", Tata McGraw-Hill.	2002
2	Mukhrjee, P.C., "Fundamentals of Metal Casting Technology", Oxford & IBH Publishing Co.	1996
. 3	Kalpakjian, S., "Manufacturing Processes for Engineering Materials", 5 th edition, Prentice Hall.	2007

NAME OF DEPARTMENT

Metallurgical and Materials Engineering

1. Subject Code:

MT-414

Course Title: Metallurgy of Joining

2. Contact Hours:

L: 3

T: 0;

P: 0

3. Examination Duration (Hrs):

Theory:

3 Practical: 0 0

4. Relative Weightage: CWS:

1 5 PRS:

0 0

MTE: 3 5

5 0

PRE: 0 O

5. Credits:

6. Semester:

Spring

7. Pre-requisite: MT-307

8. Subject Area:

DEC

9. Objective: To introduce the advanced concepts of metal joining with special reference to recent welding techniques.

Sl.No.	Contents	Contact Hours
1	Review of principles of various joining methods for metals and non-metals namely adhesive joining, welding, brazing, soldering and mechanical joining processes.	4
2	Solidification of weld pool, its structure such as dendritic, columnar and equiaxed and their effect on mechanical properties; Heat affected zone (HAZ) its structure and effect on mechanical properties; Pre and post welding heat treatments.	6
3	Metallurgical effects of the weld thermal cycle; Gas-metal equilibria, gas-metal reactions in arc welding, the mechanism of gas absorption in welding, porosity, diffusion, dilution, and uniformity of the weld deposit, weld cracking.	6
4	Effect of alloying elements on microstructure; Heat treatment related processes in fusion welding: annealing, subcritical annealing, spherodisation, normalizing, quenching hardening, tempering, secondary hardening, precipitation hardening, temper Embrittlement and graphitization.	8 .
5	Welding of carbon and ferritic alloy steels: Weldability, metallurgy of liquid weld metal; Transformation and microstructure steel, mechanical properties of welded joints, stress, intensification, Embrittlement and cracking of fusion welds.	

6	Welding of austenitic and high-alloy steels: Weldability, weld pool, alloy	6
	constitution, metallurgical transformations, Embrittlement and cracking;	ļ
	Austenitic Cr-Ni alloys for repair welding, cladding; Metallurgical	
	problems like weld decay in austenitic stainless steels.	
7	Weldability of non-ferrous metals: Weldability of Al and its alloys, Mg and	4
	its alloys, reactive and refractory metals and low melting metals.	
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	"Metallography and Microstructures", Metals Handbook, Vol.9, ASM.	2000
2	"Welding Handbook", Vol. 3, Part 2, 9th Ed. The American Welding Society.	2000
3	Kalpakjian, S., Steven R.S., "Manufacturing Engineering and Technology", Prentice Hall.	2001
4	Parmar, R.S., "Welding Engineering and Technology", Khanna Publishers.	2002

NAME OF DEPARTMENT

Metallurgical and Materials Engineering

1. Subject Code: MT-415

Course Title: Failure Analysis

2. Contact Hours:

L: 3 ; T: 0;

P: 0

3. Examination Duration (Hrs):

Theory: 0 3

Practical: 0

4. Relative Weightage: CWS: 1 5 PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0

5. Credits:

0 3

6. Semester:

Autumn

7. Pre-requisite: MT-205

8. Subject Area: DEC

9. Objective: To familiarize the students how to investigate a failure component by applying basic metallurgical principles.

Sl.No.	Contents	Contact Hours
1	Deformation and general approach to analysis of failure; Fracture aspects: Type of fracture, ductile, brittle and mixed mode fractures, models of nucleation and growth of cracks, fractography.	6
2	Determination of chemical composition by various analytical techniques; determination of mechanical properties like tensile, hardness, bend tests of failed components, comparison with Bureau of Indian Standards, quality assurance.	7
3	Fracture mechanics: Plain strain plain stress fracture toughness, K_{1C} , elastic-plastic fracture mechanics, concept of stretched zone width, its identification and determination, ductile to brittle transition of metals and failure at low temperature.	7
4	Microstructural aspects: Critical appraisal of the role of microstructure in failure, application of quantitative metallography, role of grain size and second phase particles, grain boundary and segregation, temper and hydrogen embrittlement, macro and micro examination.	8
5	Environment assisted failures: Basis principles of aqueous corrosion and high temperature corrosion and oxidation, causes and their remedies.	7
6	Case studies: Failed components from hydropower, thermal power, oil exploration rigs and rope ways, inspection of failed components on the site.	
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Raghavan, V., "Physical Metallurgy: Principles and Practice", 2 nd Ed. Prentice-Hall of India.	2007
2	Callister, W.D. Jr., "Material Science and Engineering – An Introduction", 5 th Ed. John Wiley and Sons	2000
3	Askland, R.A., "The Science and Engineering of Materials", 2 nd Ed., PWS-KENT Publishing Company.	1989
4	Raghavan, V., "Materials Science and Engineering: A First Course", 5 th Ed, Prentice-Hall of India.	2004

NAME OF DEPARTMENT Metallurgical and Materials Engineering 1. Subject Code: MT-416 Course Title: Corrosion Protection Methods

L: 3 T: 0; 3. Examination Duration (Hrs): Theory: 3 Practical:

MTE: 3 5 4. Relative Weightage: CWS: 1 | 5 PRS: 0

P: 0

5. Credits: Spring 6 Semester:

7. Pre-requisite: MT-302 8. Subject Area: DEC

9. Objective: To provide an understanding of the principles involved in corrosion and its

protection of metallic materials.

10. Details of the Course:

2. Contact Hours:

Sl.No.	Contents	Contact Hours
1	Introduction: Thermodynamics of Corrosion: Free energy change, EMF, Galvanic, Pourbaix Diagram, Electrochemical Theory; Corrosion Rates, activation polarization, concentration polarization, anodic, cathodic, mixed control; Passivation, Tafel equation.	10
2	Different Forms of Corrosions: Uniform, Galvanic, crevice, pitting, intergranular, erosion-corrosion, SCC, hydrogen cracking, corrosion fatigue, fretting corrosion; Effect of Metallurgical variables and Environments on different forms of corrosion.	8
3	Classification of Protection Techniques: By changing physical, chemical, and mechanical properties of the materials. By modification of environmental parameters, Cathodic and Anodic protection.	8
4	Coatings Technology: Various coating techniques, thermal spray, plasma spray, sputtering, metallic and non-metallic coatings for corrosion protections; Smart coatings.	8 .
5	Corrosion Resistant Alloys: Stainless steels, brasses, bronzes, and Al and Ti base alloys.	4
6	Corrosion Control Standards: Standards for conducting electrochemical tests, high temperature corrosion tests, and SCC.	4
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1	Fontana, M.G., Greene, N.D. "Corrosion Engineering", Tata McGraw-Hill.	2005
2	Uhling, H.H., Revie, R.W., "Corrosion and Corrosion Control", John Wiley& Sons.	1984
3	Aryan, R., "An Introduction to Metallic Corrosion and its Prevention". Oxford & IBH Publisher.	1988
4	Uhlig, H.H., "The Corrosion Handbook", John Wiley & Sons.	1969

	•
NAME OF DEPARTMENT	: Metallurgical and Materials Engineering
1. Subject Code: MT-417	Course Title: Inspection and Quality Control
2. Contact Hours:	L: 3 ; T: 0; P: 0
3. Examination Duration (Hrs):	Theory: 0 3 Practical: 0 0
4. Relative Weightage: CWS:	1 5 PRS: 0 0 MTE: 3 5 ETE: 5 0 PRE: 0 0
5. Credits: 0 3	6 Semester: Autumn
7 Pro requisites MT 205	9 Subject Areas DEC

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9. Objective: To impart the importance of non-destructive testing in assuring quality control in engineering components.

Sl.No.	Contents	Contact Hours
1.	Introduction: Non destructive testing and its comparison with destructive testing, role of NDT in quality control.	4
2.	Liquid Penetrant Inspection: its principles, equipment, advantages, limitations and applications.	5
3.	Magnetic Particle Inspection: its principles, equipment, advantages, limitations and applications.	6
4.	Ultrasonic Inspection: its principles, equipment, advantages, limitations and applications.	6
5.	Eddy Current Inspection: its principles, equipment, advantages, limitations and applications.	6
6.	X-ray Radiography: its principles, equipment, advantages, limitations and applications.	5
7.	Quality Control: Statistical quality control, control charts, control chart attribute and variables and acceptance sampling	6
8.	Quality Assurance: Principles and ISO 9000:2000	4
	Total	42

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1.	"Non Destructive Evaluation and Quality Control". Metals Handbook, Vol. 17, 9 th Ed., ASM.	1989
2.	Srivastava, K.C., "Handbook of Magnetic Particle Testing", Oscar Publications.	1998
3.	Srivastava, K.C., "Handbook of Liquid Penetrant Testing", Oscar Publications.	1997
4.	Grant, E.L. and Larenwork, R.S., "Statistical Quality Control", Tata McGraw-Hill.	2000

NAME OF DEPARTMENT

Metallurgical and Materials Engineering

1. Subject Code:

IMT-01

Course Title: Non Destructive Testing

2. Contact Hours:

L: 3

T: 1;

P: 0

3. Examination Duration (Hrs):

Theory:

0 3

Practical:

0 0

4. Relative Weightage: CWS:

2 5 PRS:

0 0 MTE: 2 5

2 5 ETE:

5 0

PRE: 0 0

5. Credits:

0 4

6 Semester: Both

7. Pre-requisite:

Nil

8. Subject Area:

ESEC

9. Objective: To impart the importance of non-destructive testing in assuring quality control in engineering components.

Sl.No.				
1.				
2.	Liquid Penetrant Inspection: Its principles, equipment, advantages, limitations and applications.	6		
3,	Magnetic Particle Inspection: Its principles, equipment, advantages, limitations and applications.	6		
4.	Ultrasonic Inspection: Its principles, equipment, advantages, limitations and applications.	6		
5.	Eddy Current Inspection: Its principles, equipment, advantages, limitations and applications.	- 6		
6.	X-ray Radiography: Its principles, equipment, advantages, limitations and applications.	5		
7.	Quality Control: Statistical quality control, control charts, control chart attribute and variables and acceptance sampling; Quality assurance and ISO 9000:2000	8		
	Total	42		

S.No.	Name of Author/Book/ Publisher	Year of Publication/ Reprint
1.	"Non Destructive Evaluation and Quality Control". Metals Handbook, Vol. 17, 9th Ed., ASM.	1989
2.	Srivastava, K.C., "Handbook of Magnetic Particle Testing". Oscar Publications.	1998
3.	Srivastava, K.C., "Handbook of Liquid Penetrant Testing". Oscar Publications.	1997
4.	Grant, E.L. and Larenwork, R.S., "Statistical Quality Control", Tata McGraw-Hill.	2000

NAME OF DEPTT./CENTRE:	Alternate l	Hydro Energy	Centre	
1. Subject Code: IAH-01	Course Title:	Small Hydro Po	wer Developmer	ıt
2. Contact Hours: L: 3	T: 1	P:	0	
3. Examination Duration (Hrs.):	Theory	3 Praction	cal 0	·
4. Relative Weightage: CWS 25	PRS 0	MTE 25 ETE	50 PRE 0	
5. Credits: 4 6. Sen	nester: Both	7. Subject	Area: ESEC	
8. Pre-requisite: NIL			;	
9. Objective: To provide basic knowledg designs.	e about Small	Hydro Power Tec	hnology and SHP	project
10. Details of Course:			•	

S. No.	Contents	Contact Hours
1.	Necessity and Importance of harnessing small hydro power; National policies, laws & clearances; Small hydro power scenario and type of schemes.	6
2.	Site selection and investigations; Environmental aspects; Flow duration, water power studies; Cost estimation. Economic and financial aspects.	9
3.	Diversion structures & power channels; Desilting arrangements, forebay tank and balancing reservoir. Penstock and power house building	9
4.	Types of turbines and their selection; Gates and valves; Governing system (mechanical & electrical).	8
5.	Load forecasting.	2
6.	Types of generators—synchronous and induction; Protection & controls, Power evacuation system.	8
	Total	42

S. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Adam Harvey, "Micro Hydro Design Manual", Intermediate Technology	1993
2.	Fritz, "Small Hydro Mini Power Systems", McGraw Hills	1994
3.	Emil Mosonyi, "Water power development", Vol.1&2, Nem Chand & Bros	2009
4.	Small Hydro Stations. (Publication No. 175), Central Board Of Irrigation and Power	1985
5.	Civil Engineering Guidelines for Hydroelectric Projects. (Vol.4-Small Hydro), ASCE	1989
6.	P.S. Nigam, "Hand book of Hydroelectric Engineering", Nem Chand & Bros	2001
7.	Brown G., "Hand Book Hydro-electric Engineering Practice", Vol.1, 2 &3, CBS Publication,	1984

NAME OF DEPTT./CENTRE:	Alternate l	lydro Energy Centre
1 Subject Code: IAH-02	Course Title:	Renewable Energy Sources Development Technology
2. Contact Hours: L: 3	T: 1	P: 0
3. Examination Duration (Hrs.):	Theory	3 Practical 0
4. Relative Weightage: CWS	25 PRS 0	MTE 25 ETE 50 PRE 0
5. Credits: 4 6. S	emester: Both	7. Subject Area: ESEC
8. Pre-requisite: NIL	·	
9. Objective: To provide the basic know technologies.	viedge to students	about various renewable energy resources and

S. No.	Contents	Contact Hours
1.	Energy sources & demand in different sectors, conventional & non	5
	conventional energy sources; Importance of new and renewable	
	energy sources in the present energy scenario and types of resources.	· .
2.	Small Hydro Power potential and classification of SHP projects; Basic components of civil works; Selection of electro-mechanical	8
	equipment.	
3.	Estimation of Biomass resources, Biomass Technologies for thermal	6
	and biological conversion; Biomass based electricity generation and	,
	application of bio fuels.	
4.	Solar energy estimation and different routes of solar energy	8
	applications; Technologies for solar thermal power generation;	
	Photovoltaic power generation system.	
5.	Estimation of wind energy potential and site selection; Types of wind	8
	mills, their basic characteristics and applications, Recent technologies	
	of wind energy conversion system (WECS), wind farms.	
6.	Ocean energy-potential, methods of harnessing; Geothermal energy;	7
	New technologies for renewable energy; Integrated renewable energy	
	systems.	
	Total	42

S. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Charles Y. Wereko, "Energy: The Biomass Option", John Willey.	1991
2.	Lysen, "Introduction to Wind Energy", Georgia Institute.	1998
3.	ICIMOD, "Small Hydro Design Manuals, Vol. I to IV", AHEC Publication.	1998/2005
4.	Godfrey Boyle, "Renewable Energy Power for a Sustainable Future", Oxford University Press.	1996
5.	Sukhatme S.P., "Solar Energy Principles of Thermal Collection and Storage-II Edition", Tata McGraw Hill.	1996
6.	Clare R., "Tidal Power: Trends and Development", Thomas Telford.	1992
7.	"Key World Energy Statics", International Energy Agency.	2008



ACADEMIC CALENDAR FOR THE AUTUMN SEMESTER SESSION 2009-2010

1.	Registration for Summer Term	May 20-22, 2009	Wed - Fri
2.	Summer Vacation	June 01-July 20, 2009	
3.	Joint Entrance Examination Counselling for B.Tech. / B.Arch./ IDD / Integrated M.Sc. & Integrated M.Tech. Programmes	June 09-16, 2009	Tue - Tue.
4.	Submission of Progress reports by the Ph.D. students to Departments/Centres	June 12, 2009	Friday
5.	Joint Entrance Examination Architecture/Design Aptitude Test	June 12 and June 15, 2009	Friday and Monday
6.	Declaration of JEE-2009 Counselling results	June 28, 2009	Sunday
7.	Joint Entrance Examination Counselling for Preparatory Courses 2009-10	June 30, 2009,	Tuesday
8.	Last date for submission of M.Tech. / M.Arch./ MURP / M.Sc. & M.Tech.(ES)/ IDD dissertation.	June 30, 2009	Tuesday
9.	Classes for Summer Term	May 25-, 2009 to July 06, 2009	100 M
10.	Examinations for Summer Term Courses	July 09-11, 2009	Thursday Saturday
11.	Declaration of Results for Summer Term Examination	July 15, 2009	Wednesday
12.	Institute Reopens for the Autumn Semester 2009-2010	July 21, 2009	Tuesday
13	Registration of all New UG Students including IDD/ 5 Year Integrated Courses and student of IIT Mandi	July 22, 2009	Wednesday
14.	Date of Registration of all students (Other than New Entrants)	July 23, 2009	Thursday
15.	Classes begin for all students in respective Department/Centres (0ther than New Entrants)	July 24, 2009	Friday
16.	Registration of all New PG Students including New Research Scholars	July 24, 2009	Friday .
.17.	Last date for Final Evaluation of MTech. Dissertation	July 24, 2009	Friday
18.	Orientation Programme for all New Students.	July 25, 2009	Saturday
19.	Classes begin for all New Entrants	July 27, 2009	Monday
20.	Registration/Counselling for left over seats of all UG/PG programmes	July 27, 2009	Monday
21.	Closing of admissions	July 31, 2009	Friday
22.	Selection for NSO at SCB ground	August 01, 2009 (Forenoon)	Saturday

23.	Selection for N.C.C. / N.S.S. at SCB ground	August 04 2000	Catani
23.	Selection for N.C.C. / N.S.S. at SCB ground	August 01, 2009 (Afternoon) &	Saturda y &
		August 02, 2009	Sunday
24.	On-line Subject Registration of all students (New Entrants)	August 02-05, 2009	Sunday to Wednesday
25.	Last date of addition/deletion of subject for other then new entrants.	August 07, 2009	Friday
26.	Academic/Research Section to send to Deptts/Centres final lists of Registered Students	August 10, 2009	Monday.
27.	Last date for On-line subject registration (New entrants) with late fine.	August 11, 2009	Tuesday
28.	Display/Dispatch of final list of on-line Subject Registration by Academic Section to the Department/Centres	August 14, 2009	Friday
29.	Assignment of UG Projects	August 21, 2009	Friday
30.	Departments to send the lists of Institute Elective to Academic section to be run in Spring Semester – 2009-10	August 31, 2009	Monday
31.	Last date of submission of Seminar Report (PG students)	August 31, 2009	Monday
32.	Mid Term Exam I for all UG/PG/Ph.D./Preparatory Course students	September 03-05, 2009	Thur - Sat
33.	Last date for withdrawal from a course (s)	September 11, 2009	Friday
34.	Intimation to the UG Students about the Institute Electives to be run during Spring Semester 2009-10 by the Academic Section	September 11, 2009	Friday
35.	Last date of display of attendance record of students falling short of minimum attendance requirements during the middle of semester (by departments / centres)	September 16, 2009	Wednesday
36.	On-line subject registration for Institute Electives by UG Students for Spring Semester 2009-10.	September 24-27, 2009	Thur -Sun.
37.	Intimation to parents/guardians of students having "short attendance" by Academic Section.	September 25, 2009	Friday
38.	Last date of submission of remaining document(s) by all new entrants (UG/PG/PhD)	September 30, 2009	Wednesday
39.	Semester break (for students only)	Sept.28 – Oct 01, 2009	Mon - Thur
40.	Last date for submission of Project (PG students)	Oct. 05, 2009	Monday
41.	Issue of blank progress forms for Ph.D. students by Research Section to respective Deptts./Centres	October 16, 2009	Friday
42.	Last date for finalization of time tables of Spring Semester -2009- 10 session by all departments / centres	October 16, 2009	Friday
43.	Mid Term ExamII for all UG/PG/Ph.D./Preparatory Course students	Oct. 22 – 24, 2009	Thur - Sat
44.	Jubilee Alumini Meet	Oct.24-25, 2009	Sat-Sun.
45.	Notification of seating plan Autumn Semester Exam,	October 26, 2009	Monday
46.	Notification of dates of Exam. for common subjects.	October 28, 2009	Wednesda
47.	On-line subject registration for Spring Semester for the session 2009-10 for UG /PG/ Ph.D. students	October 28-Nov 05, 2009	Wednesda -Thursday

48.	Thomso ~ 2009	Oct 30 Nov.01, 2009	Fri – Sun
49.	Filling of response forms by UG/PG students in the respective Departments /Centres	Nov. 09-11, 2009	Mon - VVed
50.	Annual Convocation	To be decided	
51.	Information regarding short attendance cases to be sent to Academic Sections by Departments/Centres	November 11, 2009	Wednesday
52.	Display of marks of course work evaluation	November 13, 2009	Friday
53.	Display of shortage of attendance cases on the Notice Boards by Departments/Centres	November 13, 2009	Friday
54.	Last day of evaluation of M.Tech. Dissertation for 3 rd Semester & sending of satisfactory/unsatisfactory report to Academic section by departments/centres	November 13, 2009	Friday
55.	Last date of Teaching for all UG/PG /Ph.D. classes	November 18, 2009	Wednesday
. 56.	Action by Academic Section to ascertain that the detained students do not appear in End Term Exam.	November 18, 2009	Tuesday
57.	Practical Examination, if any	Nov.19 to 21, 2009	Thu-Sat.
58.	End Term Exam. for all UG/PG/Ph.D. classes including Preparatory Course	Nov.23 to Dec. 03, 2009	Mon Thurs
59.	NCC Camp	To be decided by DOSW	
60.	J.M.E.T2010 Examination	December , 2009 (To be decided)	Sunday
61.	Last date to show the answer scripts of End Term Examination to the students	December 09, 2009	Wednesday
62.	Finalization of grades by the Grade Moderation Committees	December 10, 2009	Thursday
63.	Display of grades for all classes on Department Notice Board	December 10, 2009	Thursday
. 64.	Last date for sending of grades to Academic Section after scrutiny	December 11, 2009	Friday
65.	Submission of progress reports of the Ph.D. students by the Departments/ Centres	December 11, 2009	Friday
66.	Last date for declaration of Autumn Semester results	December 14, 2009	Monday
67.	Winter vacation for Teaching Staff	Dec. 15 - 29, 2009	Tue Tue.
68.	Spring Semester 2009-2010 begins	December 30, 2009	Wednesda
69.	Registration for all courses in respective Departments/Centres	December 31, 2009	Thursday
70.	Classes begin for all courses	January 01, 2010	Friday
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Note: - Heads of Department are requested to please plan the functions / Seminars on Saturdays and Sundays so that the Institute is able to maintain the minimum teaching day required in a semester.

Teaching days for Autumn Semester 2009-10 (w.e.f. 24.7.2009 to 17.11.2009) Excluding Mid-term semester break w.e.f. 28.09.2009 to 01.10.2009

Day						•
	July	August	September	October	November	Total days
Monday	27	3,10,17,24,31	7,14	5,12,26	9,16.	13
Tuesday	28	4,11,18,25	1,8,15,22	6,13,20,27	3,10,17	16
Wednesday	29	12,19,26	2,9,16,23	7,14,21,28	4,11,18	15
Thursday	30	6,13,20,27	10,17,24	8,15,29	5,12	13
Friday	24, 31	7,21,28	11,18,25	9,16,30	6,13	13
Total	6	19	16	17	12	70

Details of Saturday and Sunday used in MTE & ETE

	SATURDAY	SUNDAY
MTE (1&II)	5.9.2009; 24.10.2009	
ETE	28.11.2009	29.10.2009

APPENDIX 'H' Item No. Senate/29.2.11

PANEL OF SENATE NOMINEES

- 1. Alternate Hydro Energy Centre
- 2. Architecture & Planning
- 3. Biotechnology
- 4. Chemical Engineering
- 5. Chemistry
- 6. Civil Engineering
- 7. Paper Technology
- 8. Earth Sciences
- 9. Earthquake Engineering
- 10. Electrical Engineering
- 11. Electronics & Computer Engineering
- 12. Humanities & Social Sciences
- 13. Hydrology
- 14. Management Studies
- 15. Mathematics
- 16. Mechanical & Industrial Engineering
- 17. Metallurgical & Materials Engineering
- 18. Physics
- 19. Water Resources Development & Management

SENATE NOMINEES (S1 – S13)

Department: Alternate Hydro Energy Centre

SL. NO.	NAME	vil, Electrical, Mechanical)/Renewable Energy/Limnology-Aqu CONTACT ADDRESS			RESIDENCE ADDRESS
		OFFICE	SPECIALIZATION	MOB. NO.	
S1	Dr. S.N. Singh	Dr. S.N. Singh Deptt of Applied Mechanics, Indian Institute of	Fluid Mechanics and Solid Liquid Flow	9818300945S Fax:011 26855227 26862037 Email:sns@am.iitd.ernet.in sidhnathsingh@hotmail.com	136, Defence Enclave Vikas Marg, Delhi-110 092
i I		Technology, Delhi, New Delhi -110 016 Tel(O) 011 26591180		James James John Man. Com	
		Tel(R)011 68545519		0044000000	
S2	Prof. D. Thukaram	Prof. D. Thukaram Deptt. Of Electrical Engg., Indian Institute of Science Banglore-560012 Tel(O)080 22932362 22933168 Tel(R) 080 23600336	Renwable energy integration	09448383392 Fax:080 -23600444 Email:dtram@ee.iisc.ernet.in dtram_2001@yahoo.com	Qtr. No. NE-102, New Housing Cology HSc Campus Bangalore-560 012
S3	Dr. R. Balsubramanian	Dr. R. Balasubramanian, Centre for energy	Renewable energy	09868111541 Fax: 011 26855227, 26862037 Email:rbmanian@ces.iitd.ernet.in	House No. 6, C-Street, IIT Delhi Campus
		Studies, Indian Institue of Technology, Delhi- 110 016 Tel(O) 011 26591246 Tel® 011 26562182			

S4	Shri D.K. Mody Executive Director	Shri D.K. Mody Bharat Heavy Electrical Limited (BHEL) Hardwar- 249 403 Tel. (O) 01334 226459 Tel.(R) 01334 226461	Electrical Engineering Machines, Hydropower	9837089407 Fax:01334- 225096 Email:dkmody@bhellhwr.co.in	BHEL Campus
S5	Dr. Bhim Singh	Dr. Bhim Singh, Professor Room No. II/118, Department of Electrical Engineering, Indian Institute of Technology, Delhi Tel(O) 01126591890;011 26516223	Electrical	9811502125 Fax: 011-26581606 Email:bsingh@ee.iitd.ac.in	60, New Campus, IIT Delhi
S6	Dr. K.S. Sayann	Dr. K.S. Sayann Former Professor & Principal MANIT Bhopal Res.(O) 011 45599900 Tel(R) 011 25109242	Hydropower, Machines	09313439399 Fax:011-25136819 Email:sayannks@gmail. com	C-186, II Floor, Mansarovar Gardden, New Delhi-110 015

and the second s

S7	Dr. S. Mohan	Dr. S. Mohan Professor	Hydrological, Hydropower	09444022160 Fax:044- 22350509	1420, Sundarapuram 1 st Street Teachers
		Deptt. Of Civil Engg.		Email:smohan@iitm.ernet.in	Colony Road
		IIT Madras, Chennai-			Thanjavur-613006
		600 036			Tamil Nadu,
		Tel(O) 044 24458296			Ph.04362-255617
		2274261			
				·	
	-	Tel(R) 044			
	1	22300739,24459296			
S8	Prof. (Dr.) V.K.	Prof. (Dr.) V.K.	Remote sensing, planning	Fax: 0135 2741987	
36	Dadhwal	Dadhwal	Remote sensing, planning	Emai:dean@iirs.gov.in,	
	Daunwai	Indian Institute of		tso@iits.gov.in	
		Remote Sensing		isowans.gov.m	
	,	(National Remote			
<mark>.</mark>		Sensing Agency),			
178		deptt. Of Space,			1
φo	<u> </u>	Goevt. Of India, P.B.			
	1	No. 135, 4 Kalidas	·		-
	,	Road, Dehradun 248			·
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		Tel(O) 0135		·	
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		Tel.(R) 0135			
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S9	Prof. S.A. Khaparde	Prof. S.A. Khaparde	Deregulation in Power	9869991285	Staff Hostel Annexe
	ļ	Deptt. Of Electrical	Industry	Fax 91(0) 22-25723707	Flat No 10-IIT
		Engg. Indian Institue		Email:sak@eeiitb.ac.in	Bombay
,		of Technology,			
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		Tel(O) 91 22 2576-			
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		Tel(R) 91 22-			
<u></u>		252576-8434			

S10	Dr. J.K. Nayak	Dr. J.K. Nayak	Renewable energy ,solar	09930404624	IIT Bombay Campus
		Prof. & Head,	energy	Fax: 022-2572 6875	III Bollibay Campus
		Deptt. Of Energy		25754890	
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	· ·	Engineering, Indian			
}		Institute of			
		Technology			
	·	Bombay, Mimbai-			
		208 016		·	
		Tel(O) 022 2576			
		7881,7890			
		Tel(R) 022	· .	1.	•
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)
S11	Dr. P. Subramanian	Dr. P. Subramanian	Biomass gasifier, thermal	9442516013	93B, Park View Road
		Tel(R) 0431 2794884	conversion	Fax:0431-2500133	, Anna Nagar
				Email:psubbu44@hotmail.com	Tenur, Trichy 620 107
 		•			10110119 020 107
S12	Prof. H.R. Singh	Prof. A.S. Singh	Aquatic Ecology, Limnology	9410035821	3/33, HIG,
φ .		College of Fisheries	,	Email:profsinghhr@yahoo.co.in	Shradhapuri Phase I
		Pant Agricultural			Meerut-250001
		University Pant			·
}		Nagar, Utarakhand	·		
	1	Tel(O) 05944			
	•	233376 233377			
L		Tel(R) 9415340695			
S13.	Prof. R.S. Ambasht	Prof. R.S. Ambasht	Aquatic Ecology, Limnology	9451493555(M)	97, Jawaharnagar Ex.
		Botany Department		Email:rambasht@yahoo.com	Varanasi
		Banaras Hindu		Tel (O) 2307147/46	
		University Varanasi-		Tel (R) 0542-2277485	
		221 010			1
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L					

SENATE NOMINEE (S1-S15)

Department: Architecture & Planning

SL.	NAME	CONTACT ADDRES	RESIDENCE		
NO.		OFFICE	SPECIALIZATION	MOB. NO.	ADDRESS
S1	Prof. Kirti Trivedi	Industrial Design Centre, Indian Institute of Technology Bombay, Powai, Mumbai- 400076 Tel.(O) 022- 25767813 Tel.(R) 022-	Visual Communication & Indian Design Traditions	09892918096 Fax: 022-25783480,25767803 E-mail: kirti@iitb.ac.in	
<u>\$</u>	Prof. (Dr.) P.S. Shinde	25724244 Sri Venkateshwar College of Architecture, Hyderabad-81 Tel.(O) Tel.(R) 040- 23310059 23307134	Architecture & Planning	09866866631 E-mail: sysca_ses@yahoo.com	A-201 Satya Apt., Masaltanjk, Hyderabad-500028
S3	Prof. Ravi Poovaiah	Industrial Design Centre, Indian Institute of Technology Bombay, Powai, Mumbai- 400076 Tel.(O)022-2576820 Tel.(R)022- 25767801	Visual Art & Com. Design	Fax: 022-25783480,25767803 E-mail: ravi@iitb.ac.in	

S4	Prof. U.A. Athavankar	Industrial Design Centre, Indian Institute of Technology Bombay, Powai, Mumbai- 400076 Tel.(O)022- 25767817 Tel.(R)	Product Design, Product Semantics, Cognition & Imagery	Fax: 022-25783480/25767803 E-mail: <u>uaa@iitb.ac.in</u>
B. PI	LANNING			
S5	Prof. Utpal Sharma	Faculty of Planning & Public Policy,	Urban Planning & Housing Design	09825082808
		CEPT University, Kasturbhai Lalbhai		E-mail: utpalsharma@cept.ac.in
		Campus, University Road,		
181		Navarangpura, Ahmedabad 380009		
1		(Gujarat) Tel.(O) Tel.(R)		
S6	Prof. A.N.	Dean, MEASI	Urban Development and	09840324140
	Sachithanandan	Academy of	Housing Planning	Fun 044 08251050
		Architecture, 8T Peters Road,		Fax: 044-28351050
		Royapettah,		E-mail: sivusachi@yahoo.com
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		Tel.(O) 044-		
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S8	Prof. A.K. Gosain	Prof. A.K. Gosain,	Integrated Watershed	9810944776	
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25 10 10		Engg. Deptt. Of	Impact Assessment.	}	
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S9	Shri R.S. Varshney	Shri R.S. Varshney	Hydrology, Hydraulic	09818841616	Varshney, Siddhidatri,
1		Engineering in	Structures, Hydropower and		8/127,Sector -3 Rajendra Nagar,
		Chief (Retd.)	Dams	ę L	Ghaziabad-201005
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S10	Shri Rakesh Nath	Shri Rakesh Nath,	Thermal and Hydro Power	9810330909	The state of the s
}		Chairman Central	Stations and Transmission	Fax: 011 26109212	
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S11	Shri Gurdas Singh	Shri Gurdas Singh,	Electrical Engineering	Email: gdsingh51@yahoo.com	7.00
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}		Central Electricity			Puram, New Delhi-22
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		24670578			
S12	Shri M.K. Sharma	Shri M.K. Sharma,	Hydro Mechanical Equipment	Fax: 011 26442849	45/GF/Block-III, Charmwood
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S13	Shri S.M. Dhiman	Shri S.M. Dhiman		Fax 011-26108834	J-4/2, Sector-13 M.S. Flats. R.K.
ļ		Member (GO&D)		Email: smdhiman@gmail.com	Puram, New Delhi-22
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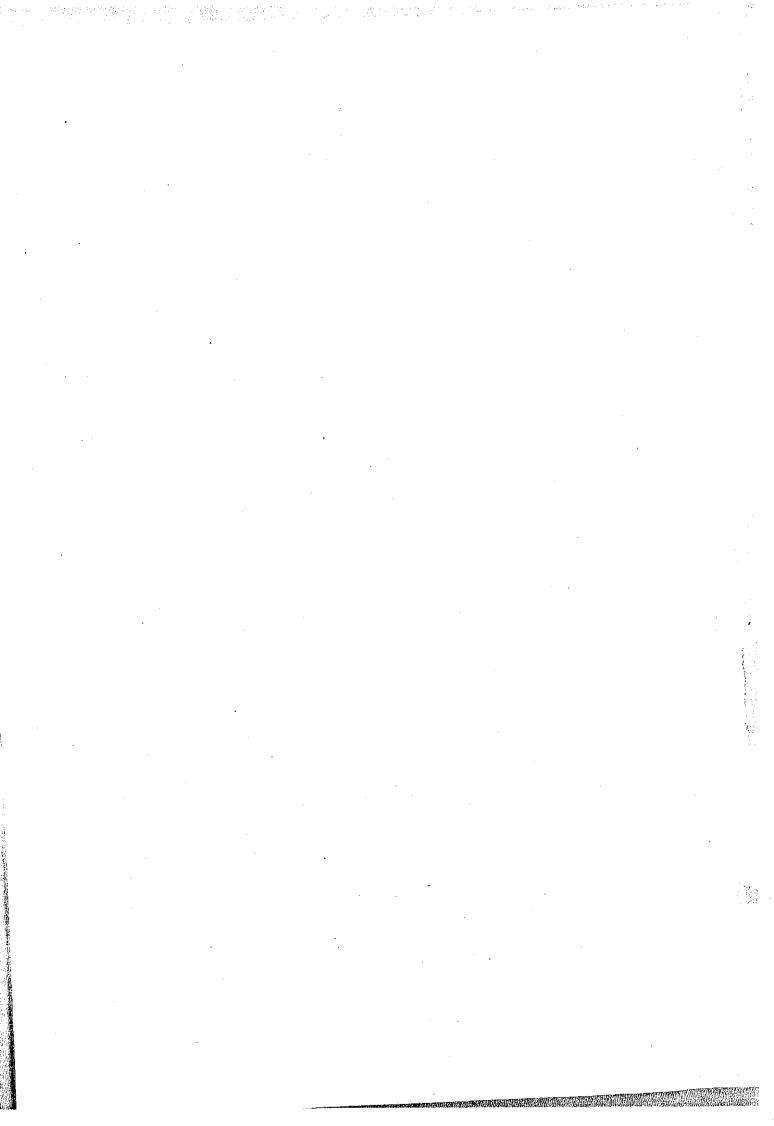
S14	Dr. R.C. Chauhan	Dr. R.C. Chauhan Ex Director,	Electrical Engineering, hydro power	9417222440 9721811655	
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S15	Dr. Bhim Singh	Dr. Bhim Singh	Power Electronics, Electrical Machines And Drives, HVDC,	Email bsingh@eeiitd.ac.in	60, New Campus, IIT Dellhi
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Î		Director (Technical	Storago		
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		Tel(O) 0135 2431468		}	
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Hydro) Mechanical/Cor		& Machinery (WRD Mecl	hanical)	ecoron i in ol.
S17	Shri M.K. Sharma	Shri M.K. Sharma	Hydro mechanical equipments	Email:sharmamk@jpcl.	45/GF/Block-III, Charmwood Vilalge, Faridabad-121 009
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QS 20	Shri R.K. sharma	Shri R.K. Sharma	Construction planning and	09810137215	B1/204 PWO Housing Coomplex,
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S22	Shri S. K Das	Shri S.K. Das Ex Chairman, CWC Tel. (R)033 24187814	River Basin	09433003508 Fax: 03323599260	16/9, W2A ®, Phase IV B Golf Gren. Kolkata 700095
\$23 \$0 \$0	Shri N.K. Tyagi	Agriculture Scientists Recruitment Board (ASRB), Room No. 204, Krishi Anusandhan Bhawan-1, Pusa, New Delhi 1100:2	Irrigation Engineering and Land Reclamation	Fax: 011-25841168 Email: nktyagi@icar.org.in Cssir@x400niegw.mic.in pamember@rediffmail.com.	
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S26	Dr. S.K. Sharma	Dr. S.K. Sharma	Natural Resource	09412551914	J-22, Shastri Nagar, Meerut -
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627	Prof. B.C. Mal	Prof. B.C. Mal	Ground Water Engineering,	09434714077	_
S27	FIOL B.C. Iviai	Professor & Ex Head	Irrigation & Drainage	Email: bmal@agfe.iitkgp.ernet.in	
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Recommendations

- 1. The sum of the credits of 4th &5th yr of integrated program should be, by and large, equivalent to the total credits for 2 yr MSc program.
- 2. The credits for dissertation in 4th semester of 2 yr program should be the same as that for dissertation in 10th semester of 5 yr program.
- 3. There should be greater emphasis on practical courses and Lab training to earn any Master's degree in science.
- 4. The practice of Lab courses being run as separate units in certain science departments, covering various aspects of different topics covered in theory papers, should be continued.
- 5. Having separate / independent courses on laboratory is a well established and standard practice particularly in Chemistry and Physics at all levels, e.g., BSc, MSc, in all the universities in India and equivalent programs in UK, Germany etc.
- 6. The department should assign more credits to dissertation and certainly less for seminar, project and comprehensive viva at MSc level.
- 7. In the summer vacation, after 2nd semester, there should be an industrial training or project in an outside institute/industry/national laboratory etc.
- 8. The practice of tutorial at MSc final year level (5th yr of integrated programs and 2nd yr of 2-yr programs) should be abolished. The courses at MSc level are in fact, by and large, specialized courses.
- 9. Technical Communication is not required for post BSc programs.
- 10. A range for credits and distribution among various components is proposed in Table: Table: Proposed Credit range and distribution among different components

SI.	Course Component	Credits
No		
1	Course on Modelling, Simulation and Computer Applications (Optional or modified as per specific needs of the department)	03-04
2	Department Core Courses	32-40
3	Lab / Practical courses	08-12
4	Departmental Electives (DEC)	12-16
5	IE or Minor Electives (electives from other departments / sections within the department / center	06-08
. 6	Dissertation	12-16
	Seminar, Project, field work and comprehensive Viva-voce	04-10
	Total	85-95

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